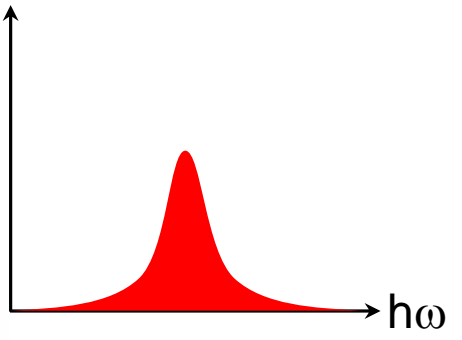
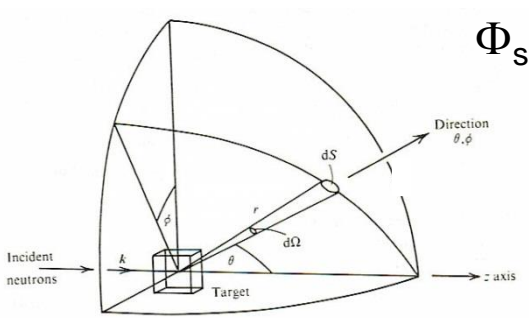
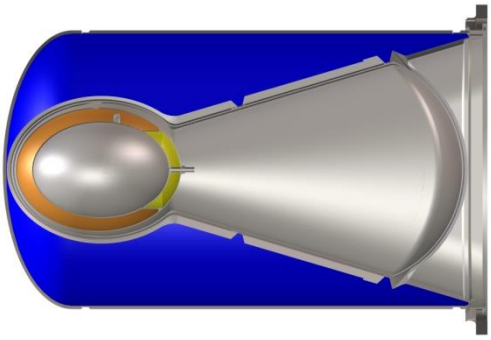
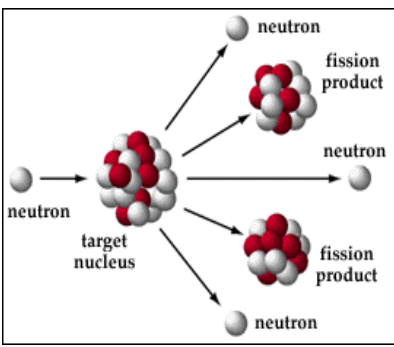


2017 NCNR Summer School on Methods and Applications of Neutron Spectroscopy

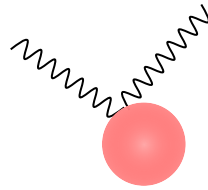


Basic Elements of Neutron Inelastic Scattering

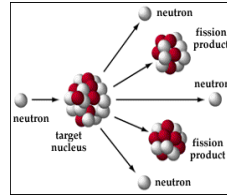
Peter M. Gehring
 National Institute of Standards and Technology
 NIST Center for Neutron Research
 Gaithersburg, MD USA



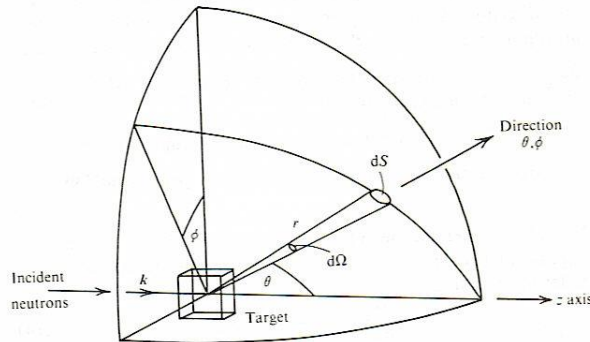
Outline



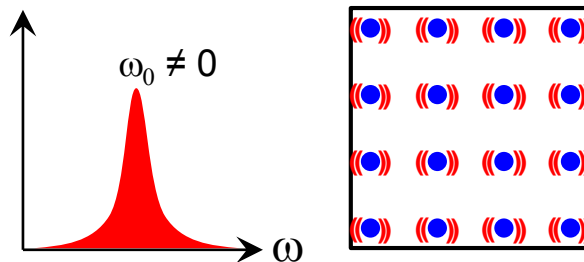
1. Introduction
 - Motivation
 - Scattering Probes



2. The Neutron
 - Production and Moderation
 - Wave/Particle Duality



3. Basic Elements of Neutron Scattering
 - The Scattering Length b
 - Scattering Cross Sections
 - Pair Correlation Functions
 - Coherent and Incoherent Scattering
 - Neutron Scattering Methods

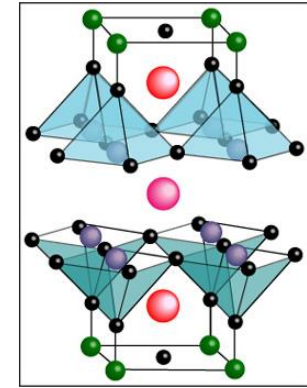


4. Summary of Scattering Cross Sections
 - Elastic (Bragg versus Diffuse)
 - Quasielastic (Diffusion)
 - Inelastic (Phonons)

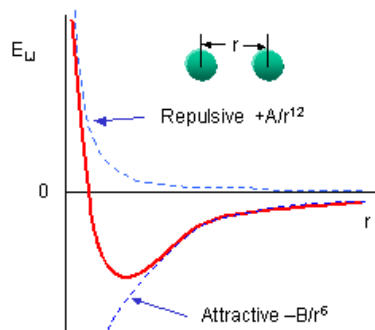
Motivation

Structure and Dynamics

The most important property of any material is its underlying atomic / molecular structure (structure dictates function).



$\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$

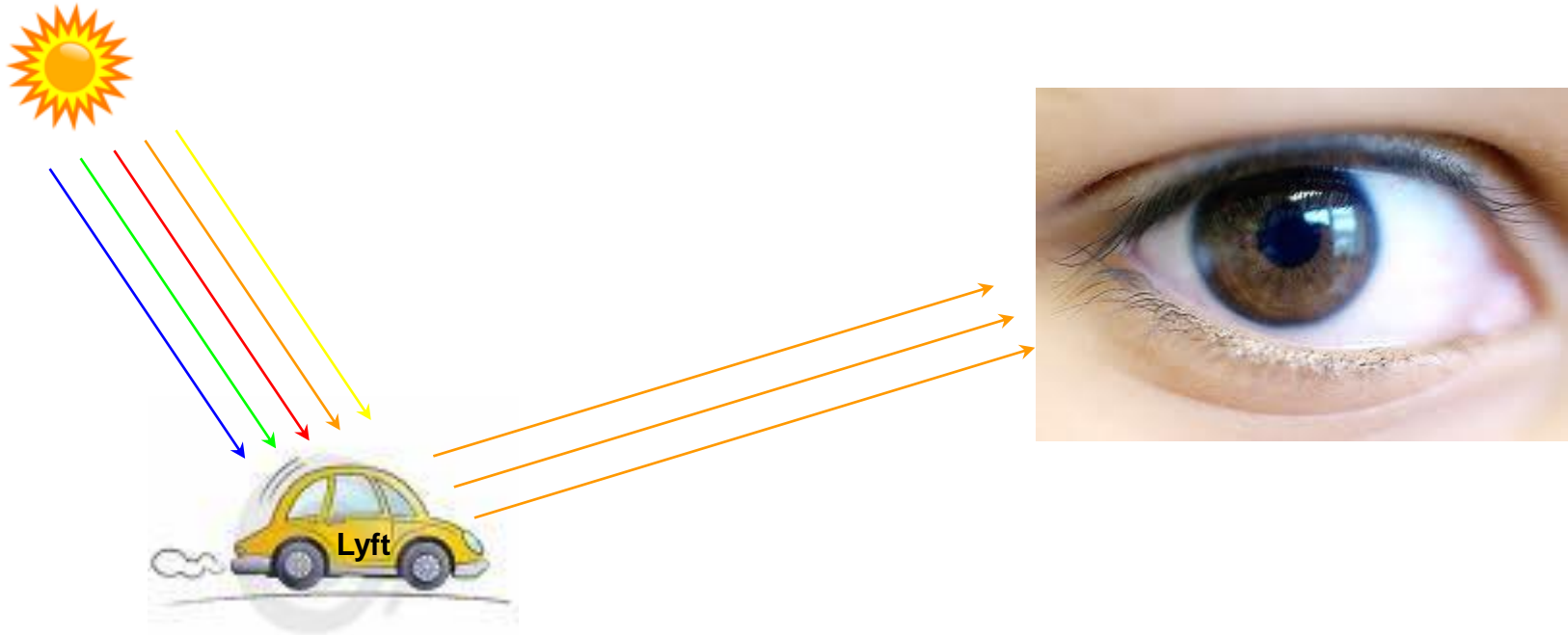


The motions of the atoms (dynamics) are extremely important because they provide information about the interatomic potentials.

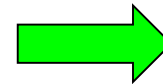
An ideal method of characterization would provide detailed information about both structure and dynamics.

Motivation

How do we “see”?



● We see something when light scatters from it.



Thus scattering conveys information!

● Light is composed of electromagnetic waves.



$\lambda \sim 4000 \text{ \AA} - 7000 \text{ \AA}$

● However, the details of what we can see are ultimately limited by the wavelength.

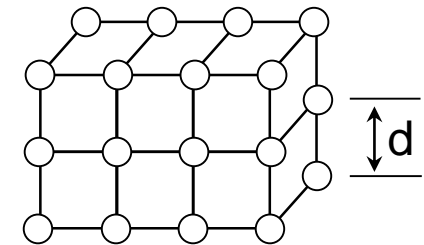
Motivation



The tracks of a compact disk act as a diffraction grating, producing a separation of the colors of white light when it scatters from the surface.

From this one can determine the nominal distance between tracks on a CD, which is 1.6×10^{-6} meters = 16,000 Angstroms.

To characterize materials we must determine the underlying structure. We do this by using the material as a diffraction grating.



Problem: Distances between atoms in materials are of order Angstroms \rightarrow **light is inadequate**. Moreover, most materials are opaque to light.

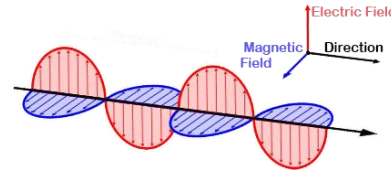
$$\lambda_{\text{Light}} \gg d \sim 4 \text{ \AA}$$

Scattering Probes

To measure atomic structure requires a probe with a $\lambda \sim$ length scale of interest.

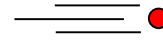
Some candidates ...

X rays



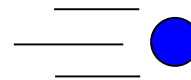
EM - wave

Electrons



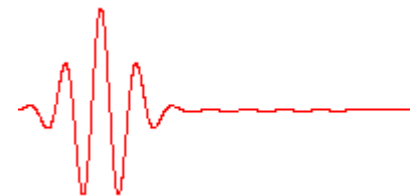
Charged particle

Neutrons



Neutral particle

Remember de Broglie: $\lambda = h/p = h/mv$
Particles have wave properties too.



Scattering Probes

Pros and Cons ...

Which one should we choose?

If we wish only to determine relative atomic positions, then we should choose **x rays** almost every time.

1. Relatively cheap
2. Sources are ubiquitous → easy access
3. High flux → can study small samples

However ...

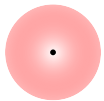
Scattering Probes

X rays are electromagnetic radiation.
Thus they scatter from the charge density.

Consequences:

Low-Z elements are hard to see.

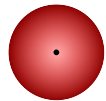
Hydrogen



(Z = 1)

Elements with similar atomic numbers have very little contrast.

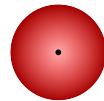
Cobalt



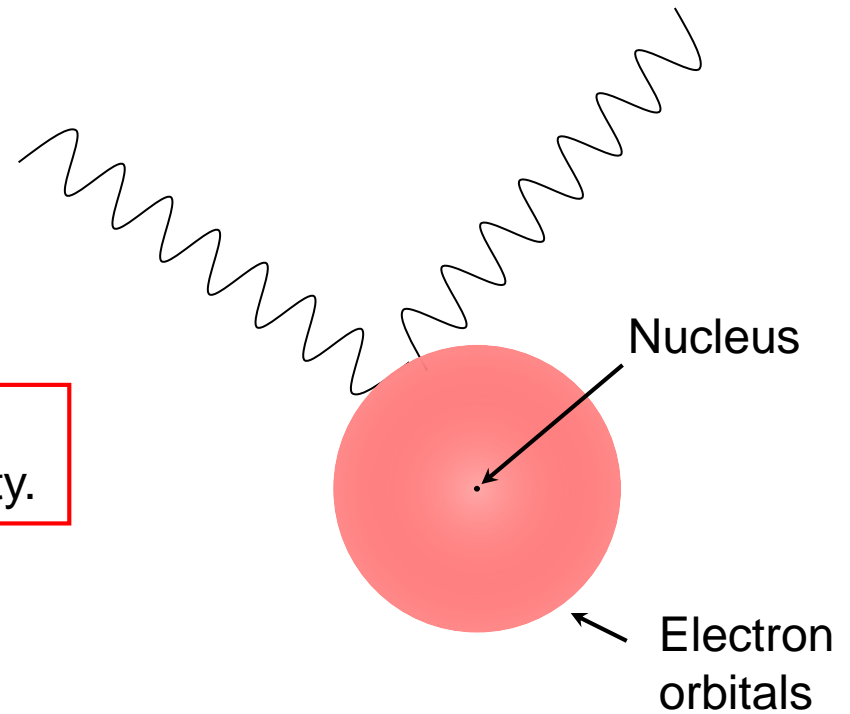
(Z = 27)

??

Nickel



(Z = 28)

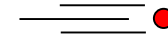


X rays are strongly attenuated as they pass through the walls of furnaces, cryostats, etc.



Scattering Probes

What about electrons?



Electrons are charged particles → they see both the atomic electrons and nuclear protons at the same time.

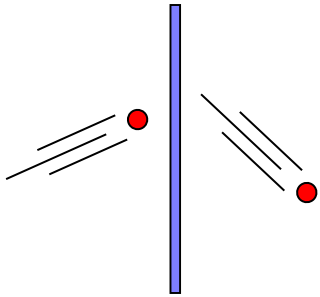
1. Relatively cheap
2. Sources are not uncommon → easy access
3. Fluxes are extremely high → can study tiny crystals
4. Very small wavelengths → more information

However ...

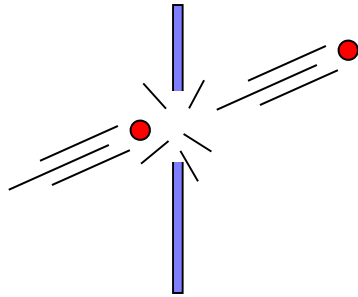
Scattering Probes

Electrons have some deficiencies too ...

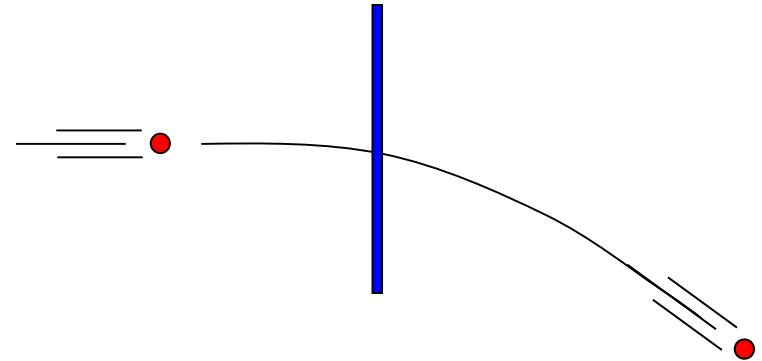
Requires very thin samples.



Radiation damage is a concern.

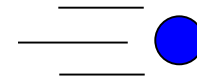


Magnetic structures are hard to determine because electrons are deflected by the internal magnetic fields.



Scattering Probes

What about neutrons?



Advantages

Wavelengths easily varied to match atomic spacings

Zero charge → not strongly attenuated by furnaces, etc.

Magnetic dipole moment → can study magnetic structures

Nuclear interaction → can see low-Z elements easily like H → good for the study of biomolecules and polymers.

Nuclear interaction is simple → scattering is easy to model
Low energies → Non-destructive probe

Disadvantages

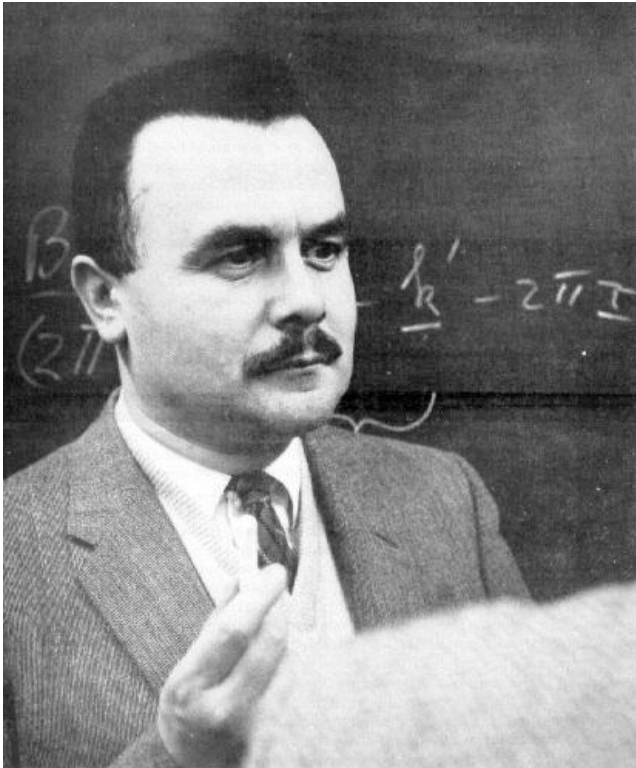
Neutrons expensive to produce → access not as easy

Interact weakly with matter → often require large samples

Available fluxes are low compared to those for x rays

Let's consider neutrons ...

The Neutron



“If the neutron did not exist, it would need to be invented.”

Bertram Brockhouse
1994 Nobel Laureate in Physics

The Neutron



“... for the discovery
of the neutron.”

Sir James Chadwick
1935 Nobel Laureate in Physics

The Neutron

$$m_n = 1.675 \times 10^{-27} \text{ kg}$$

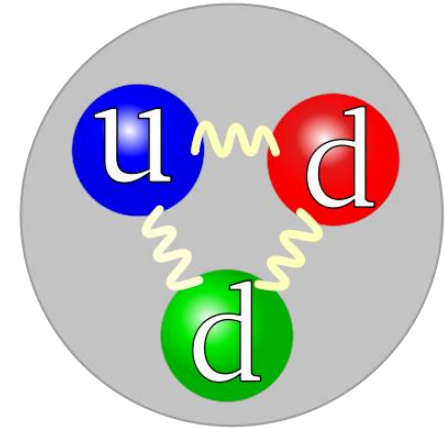
$$Q = 0$$

$$S = \frac{1}{2} \hbar$$

$$\mu_n = -1.913 \mu_N$$

1924: de Broglie Relation

$$\lambda = h/p = h/m_n v$$



$$\lambda = 1 \text{ \AA}$$

$$v = 4000 \text{ m/s}$$

$$E = 82 \text{ meV}$$

$$\lambda = 9 \text{ \AA}$$

$$v = 440 \text{ m/s}$$

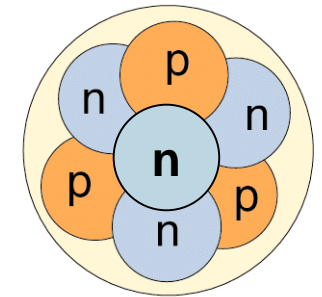
$$E = 1 \text{ meV}$$

Production

Free neutrons decay via the weak force.
Lifetime ~ 888 seconds (15 minutes).



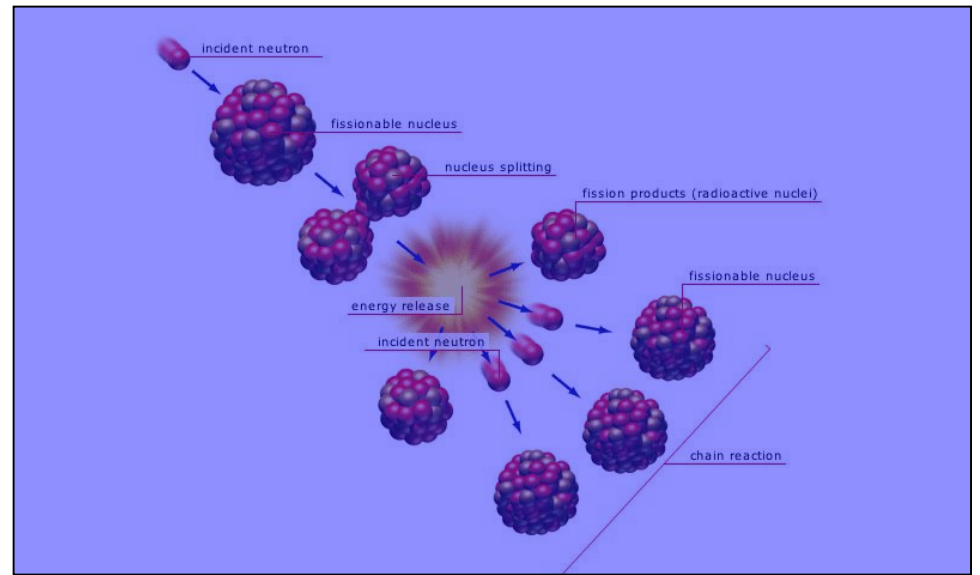
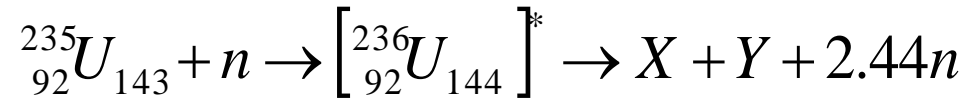
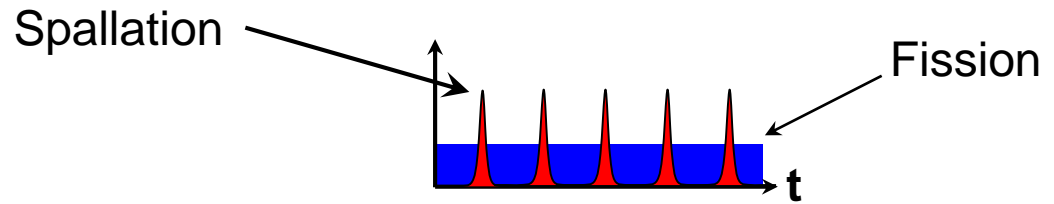
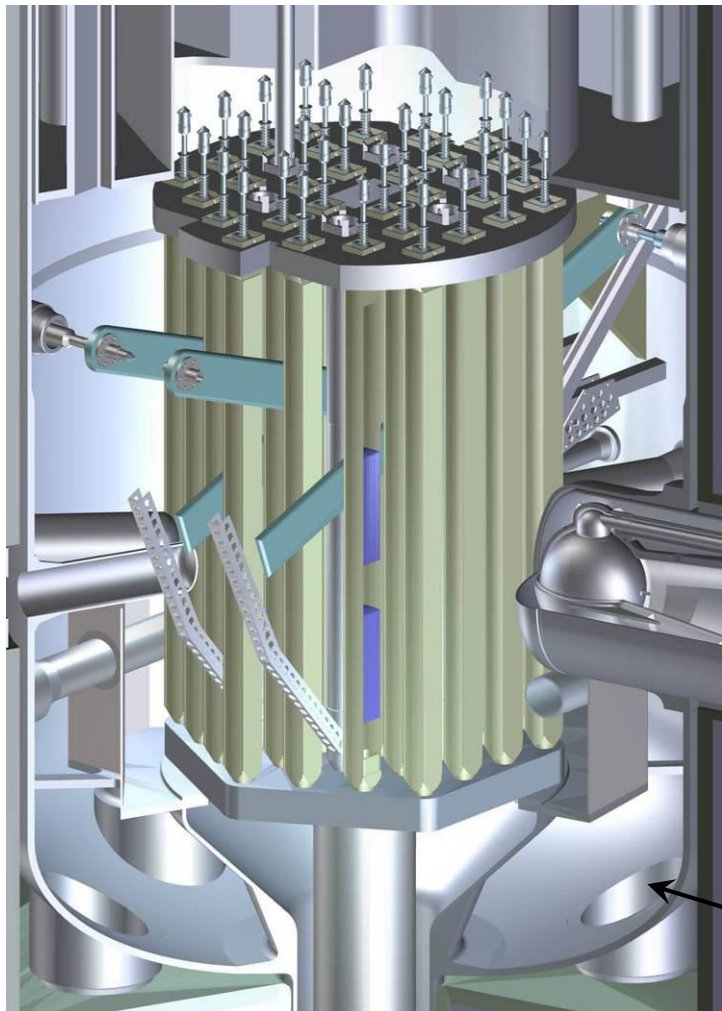
A useful source of neutrons requires a nuclear process by which bound neutrons can be freed from the nuclei of atoms and that is easily sustainable.



There are two such processes,
spallation and fission ...

Fission

Nuclear fission is used in power and research reactors.



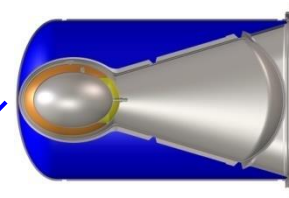
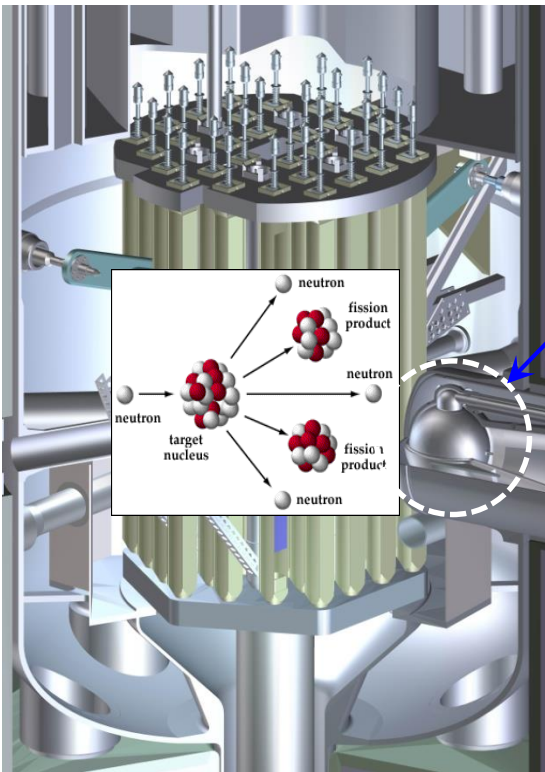
A liquid medium (D_2O , or heavy water) is used to moderate the fast fission neutrons to room temperature ($2 \text{ MeV} \rightarrow 50 \text{ meV}$).

The fission process and moderator are confined by a large containment vessel.

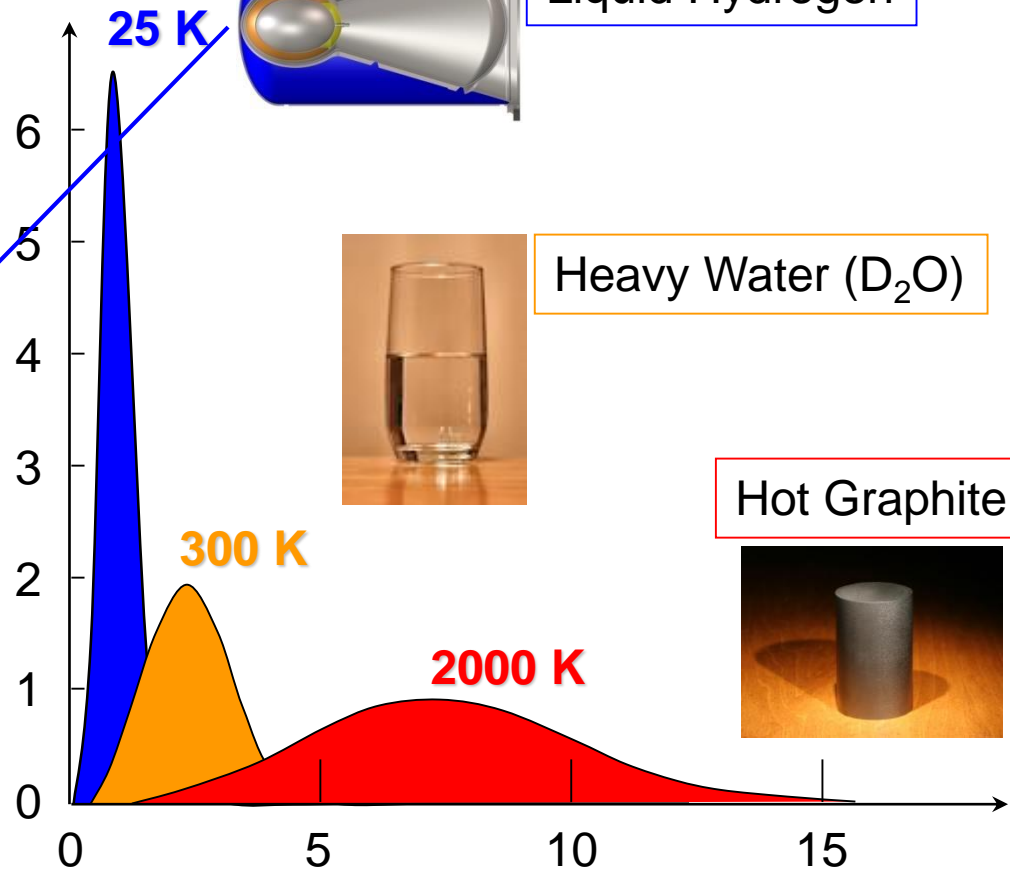
Moderation

Maxwellian Distribution

$$\Phi \sim v^3 e^{(-mv^2/2k_B T)}$$



Liquid Hydrogen



Heavy Water (D₂O)

Hot Graphite

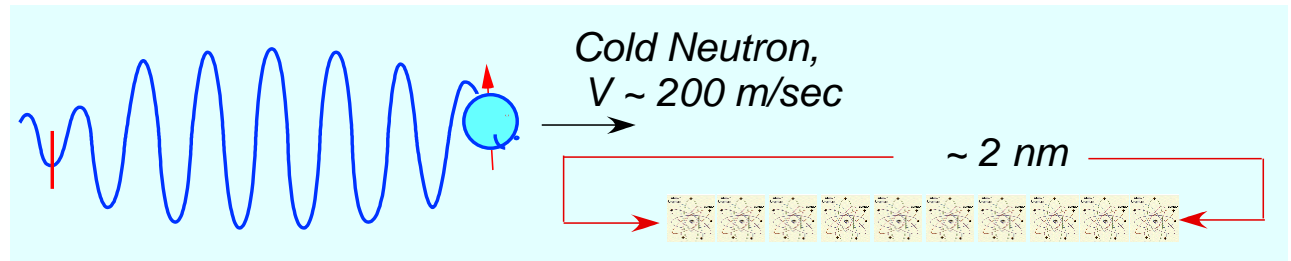
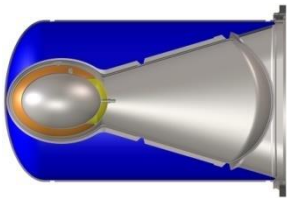
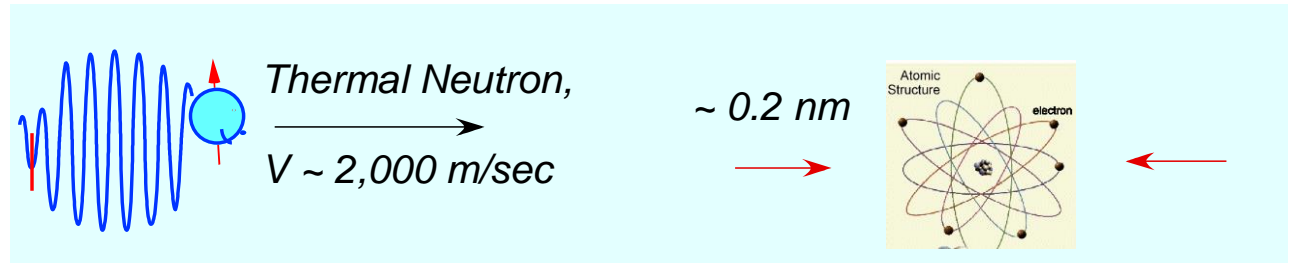
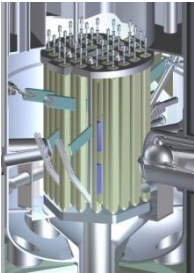
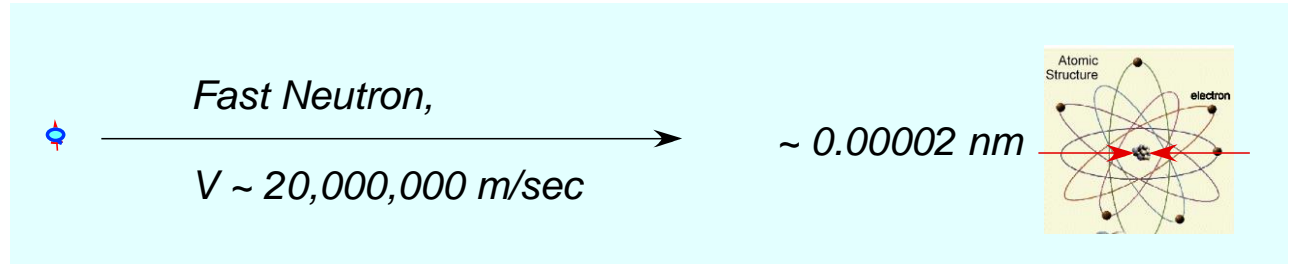
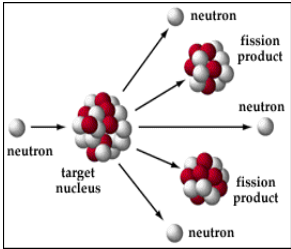


“Fast” neutrons: $v = 20,000$ km/sec

Neutron velocity v (km/sec)

Moderation

$$\lambda = h/m_n v$$



Scattering Basics

Neutron scattering experiments measure the flux Φ_s of neutrons scattered by a sample into a detector as a function of the change in neutron wave vector (\vec{Q}) and energy ($\hbar\omega$).

Momentum

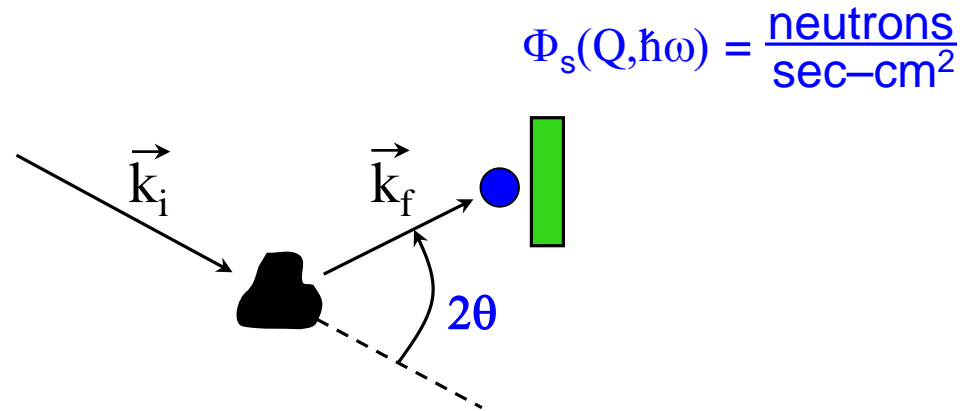
$$\hbar k = \hbar(2\pi/\lambda)$$

$$\hbar\vec{Q} = \hbar\vec{k}_i - \hbar\vec{k}_f$$

Energy

$$\hbar\omega_n = \hbar^2 k_n^2 / 2m$$

$$\hbar\omega = \hbar\omega_i - \hbar\omega_f$$



The expressions for the scattered neutron flux Φ_s involve the positions and motions of atomic nuclei or unpaired electron spins.

$$\Phi_s = \mathbb{F}\{\vec{r}_i(t), \vec{r}_j(t), \vec{S}_i(t), \vec{S}_j(t)\}$$

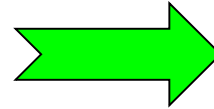


Φ_s provides information about all of these quantities!

Scattering Basics

These “cross sections” are what we measure experimentally.

Consider an incident neutron beam with flux Φ_i (neutrons/sec/cm²) and wave vector \mathbf{k}_i on a non-absorbing sample.

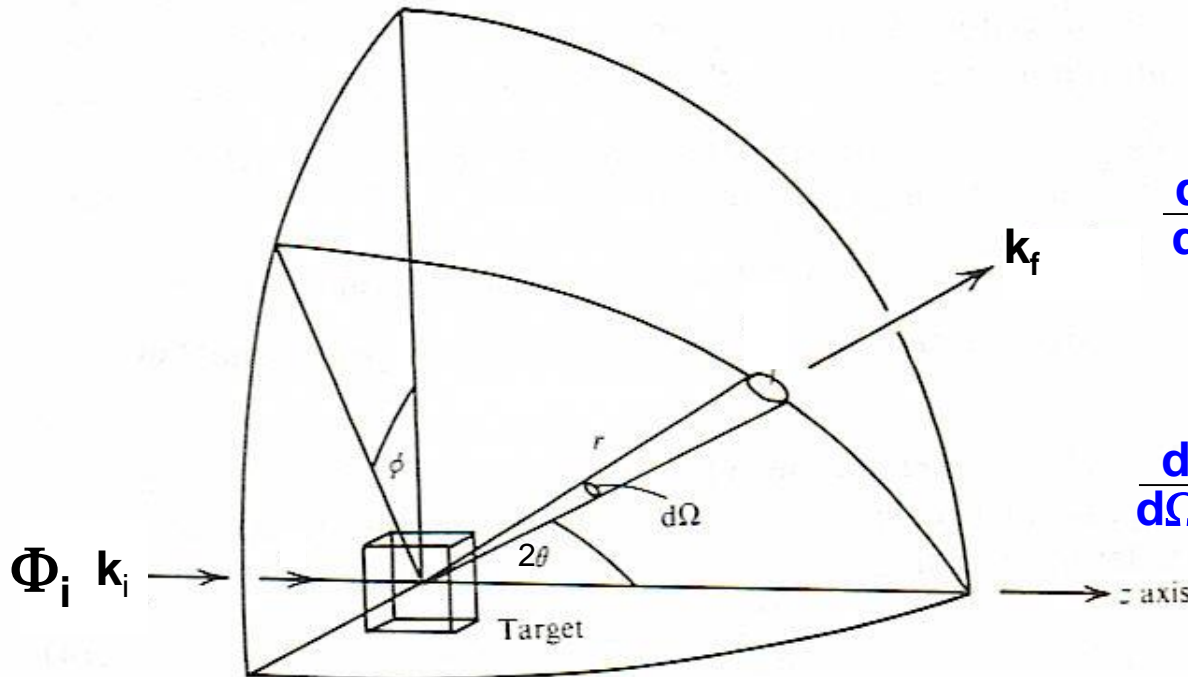


We define three cross sections:

σ = Total cross section

$\frac{d\sigma}{d\Omega}$ = Differential cross section

$\frac{d^2\sigma}{d\Omega dE_f}$ = Partial differential cross section



Cross Sections

What are the physical meanings of these three cross sections?

σ Total # of neutrons scattered per second / Φ_i .

$$\frac{d\sigma}{d\Omega}$$

Total # of neutrons scattered per second into $d\Omega$ / $d\Omega \Phi_i$.
(**Diffraction** → structure.)

$$\frac{d^2\sigma}{d\Omega dE_f}$$

Total # of neutrons scattered per second into $d\Omega$ with a final energy between E_f and dE_f / $d\Omega dE_f \Phi_i$.
(**Inelastic scattering** → dynamics.)

Cross Sections

What are the relative sizes of the cross sections?

Clearly: $\sigma = \int \frac{d\sigma}{d\Omega} d\Omega = \int \frac{d^2\sigma}{d\Omega dE_f} d\Omega dE_f$


Thus: $\sigma \gg \frac{d\sigma}{d\Omega} \gg \frac{d^2\sigma}{d\Omega dE_f}$



σ



$\frac{d\sigma}{d\Omega}$



$\frac{d^2\sigma}{d\Omega dE_f}$

Typically, $\frac{d\sigma}{d\Omega} \sim \underline{\underline{10^6}} \times \frac{d^2\sigma}{d\Omega dE_f}$

Elastic vs Inelastic

Note that both of these cases are described by ...

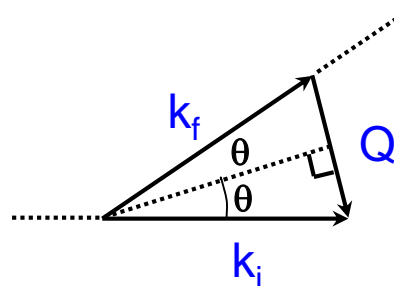
$$\frac{d^2\sigma}{d\Omega dE_f}$$

Elastic Scattering:

- Change in neutron energy = 0
- Probes changes in momentum only

Elastic ($k_i = k_f$)

$$\neq \frac{d\sigma}{d\Omega}$$



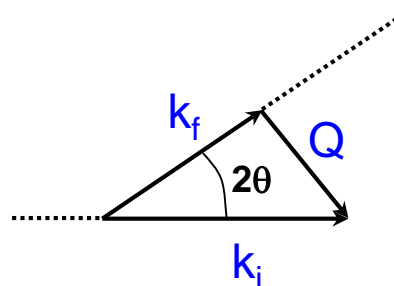
$$\sin\theta = (Q/2)/k$$

$$Q = 2k\sin\theta = 4\pi\sin\theta/\lambda$$

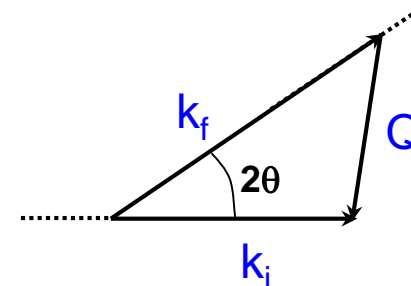
Inelastic Scattering:

- Change in neutron energy $\neq 0$
- Probes changes in both momentum and energy

Inelastic ($k_i \neq k_f$)



Energy loss ($\hbar\omega > 0$)



Energy gain ($\hbar\omega < 0$)

Nuclear Scattering

Consider the simplest case:
A fixed, isolated nucleus.

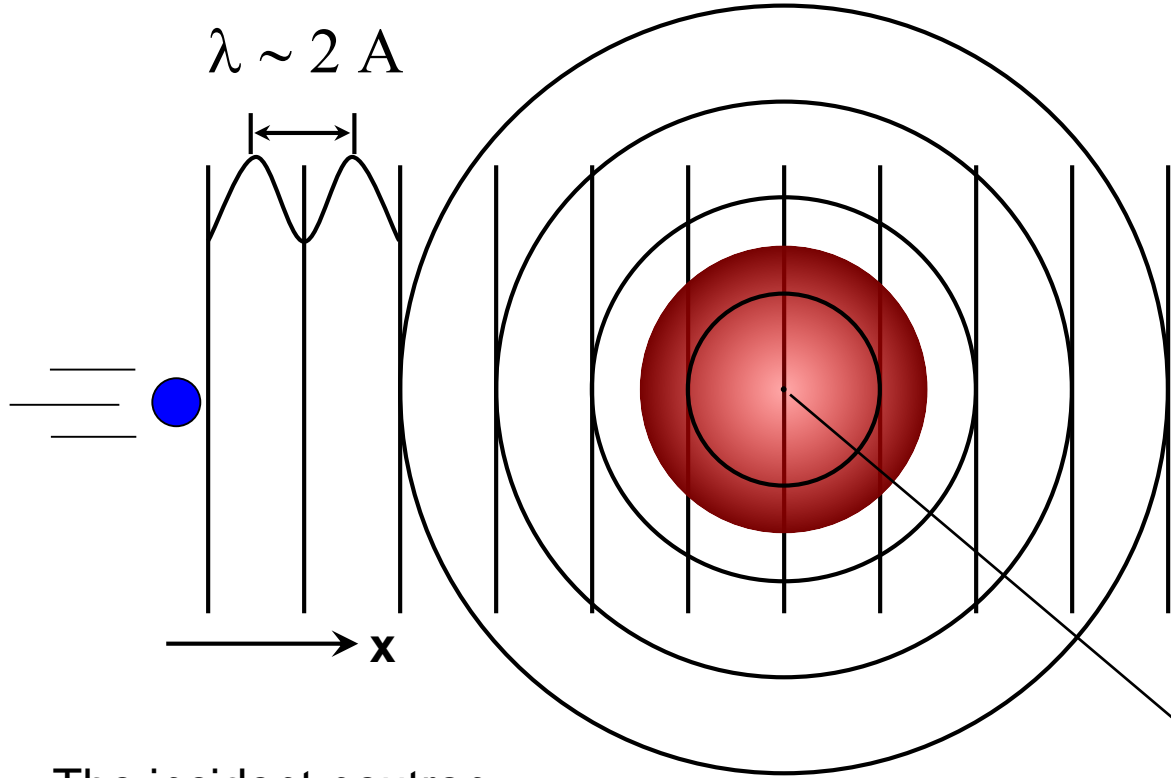
The scattered (final) neutron Ψ_f is a spherical wave:

$$\Psi_f(\mathbf{r}) \sim (-b/r)e^{ikr}$$

QUESTIONS:

1. The scattering is elastic ($k_i = k_f = k$). Why?

2. The scattering is isotropic. Why?



The incident neutron Ψ_i is a plane wave:

$$\Psi_i(\mathbf{r}) \sim e^{ikx}$$

$$(k = 2\pi/\lambda)$$

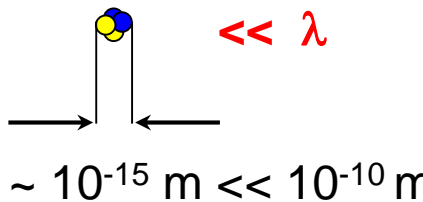
Nucleus
 $1 \text{ fm} = 10^{-15} \text{ m}$

Nuclear Scattering

ANSWERS:

1. The scattering is elastic because the nucleus is fixed, so no energy can be transferred to it from the neutron (ignoring any excitations of the nucleus itself).

2. A basic result of diffraction theory states: if waves of any kind scatter from an object of a size $\ll \lambda$, then the scattered waves are spherically symmetric. (This is also known as s-wave scattering.)



The diagram shows a nucleus represented by a vertical rod with a yellow and blue sphere on top. Two horizontal arrows point towards the nucleus from the left and right. Below the nucleus, the text reads $\sim 10^{-15} \text{ m} \ll 10^{-10} \text{ m}$. To the right of the nucleus, the text $\ll \lambda$ is written in red. A large green arrow points from the diagram towards the equation.

$$V(r) = \left(\frac{2\pi\hbar^2}{m_n} \right) \sum_{j=1}^N b_j \delta(r-r_j)$$

Details of $V(r)$ are unimportant!
 $V(r)$ can be parametrized by a scalar b that depends only the nucleus and isotope!

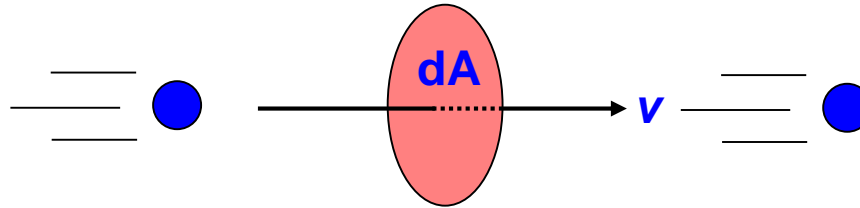
Fermi Pseudopotential

Nuclear Scattering

We can easily calculate σ for a single, fixed nucleus:

Def.

$\sigma \cdot \Phi_i =$ Total number of neutrons scattered per second by the nucleus.
 $v =$ Velocity of neutrons (elastic \rightarrow same before and after scattering).



$v dA |\Psi_f|^2 =$ Total number of neutrons scattered per second through dA .

$$\int v dA |\Psi_f|^2 = \int v (r^2 d\Omega) (b/r)^2 = \int v b^2 d\Omega = v 4\pi b^2 = \sigma \cdot \Phi_i$$

Since $\Phi_i = v |\Psi_i|^2 = v \rightarrow$

$$\sigma = 4\pi b^2$$

Try calculating $\frac{d\sigma}{d\Omega}$

Nuclear Scattering

The Neutron Scattering Length - b

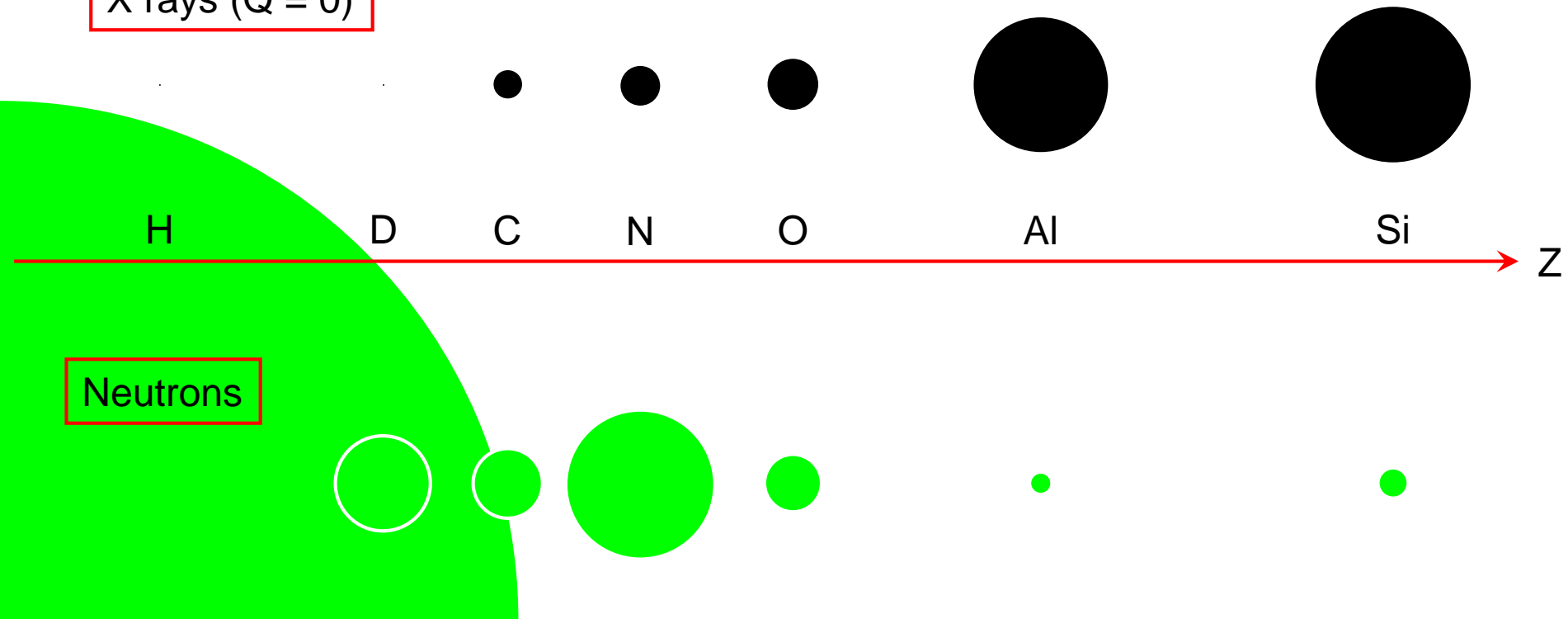
Units of length:
 $b \sim 10^{-12}$ cm.

Analogous to $f(Q)$, the x-ray scattering form factor.

Varies randomly with Z and isotope
→ Neutrons “see” atoms x rays can’t.

Total Scattering Cross Section: $\sigma = 4\pi b^2$

X rays ($Q = 0$)

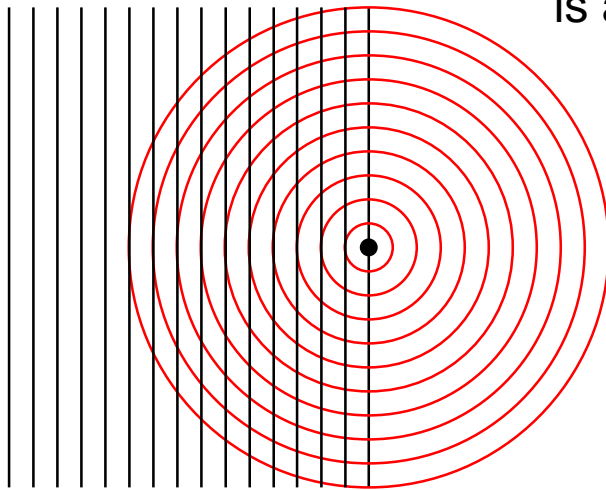


Neutrons

Nuclear Scattering

What if many atoms are present?

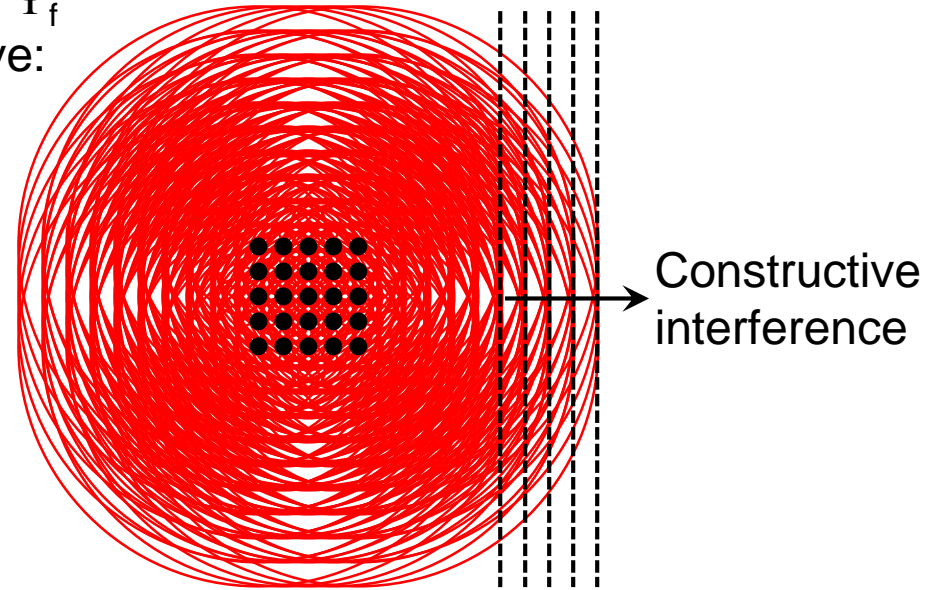
Scattering from one nucleus



Scattered neutron Ψ_f is a spherical wave:

The incident neutron Ψ_i is a plane wave:

Scattering from many nuclei



Get strong scattering in some directions, but not in others. Angular dependence yields information about how the nuclei are arranged or correlated.

Magnetic Scattering

What about magnetic scattering?

Nuclear Potential

$$V_N(\mathbf{r}) = \frac{2\pi\hbar^2}{m_n} b\delta(\mathbf{r})$$

Scalar interaction →
Isotropic scattering

Very short range

Depends on nucleus,
isotope, and nuclear spin

Magnetic Potential

$$V_M(\mathbf{r}) = -\boldsymbol{\mu}_n \cdot \mathbf{B}(\mathbf{r})$$

Vector interaction →
Anisotropic scattering

Longer range

Depends on neutron
spin orientation.



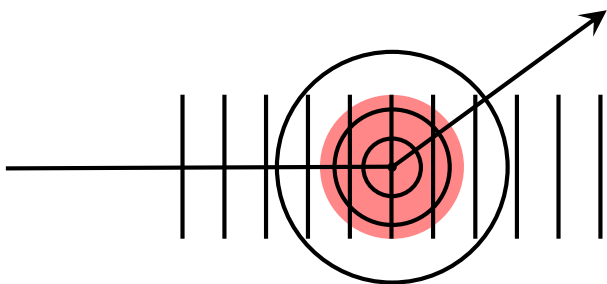
Polarized neutrons can measure
the different components of M .

Magnetic Scattering

What about magnetic scattering?

Nuclear Potential

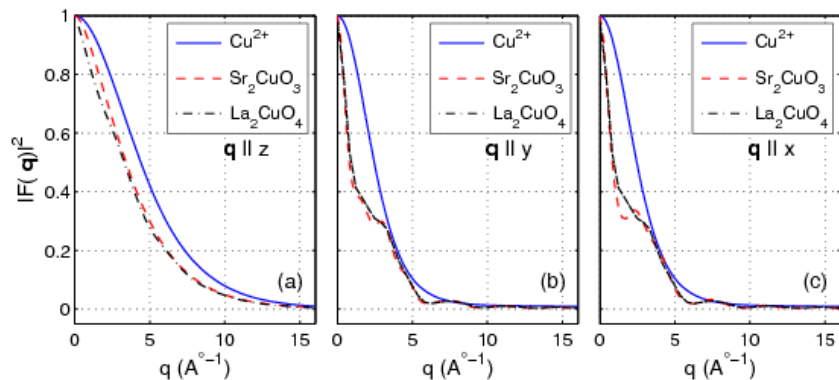
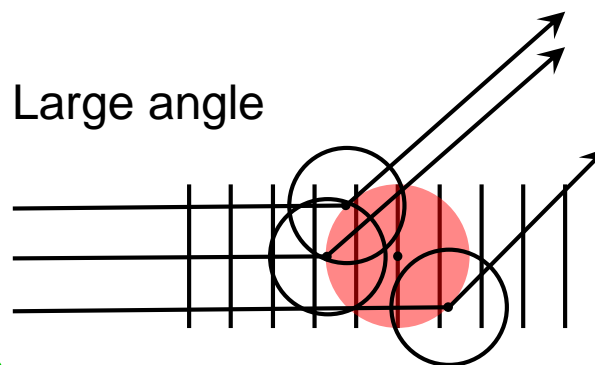
$$V_N(\mathbf{r}) = \frac{2\pi\hbar^2}{m_n} b\delta(\mathbf{r})$$



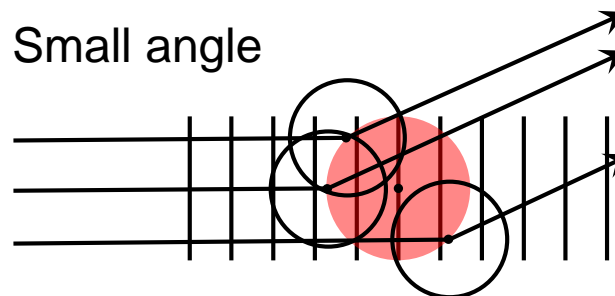
Magnetic Potential

$$V_M(\mathbf{r}) = -\boldsymbol{\mu}_n \cdot \mathbf{B}(\mathbf{r})$$

Large angle

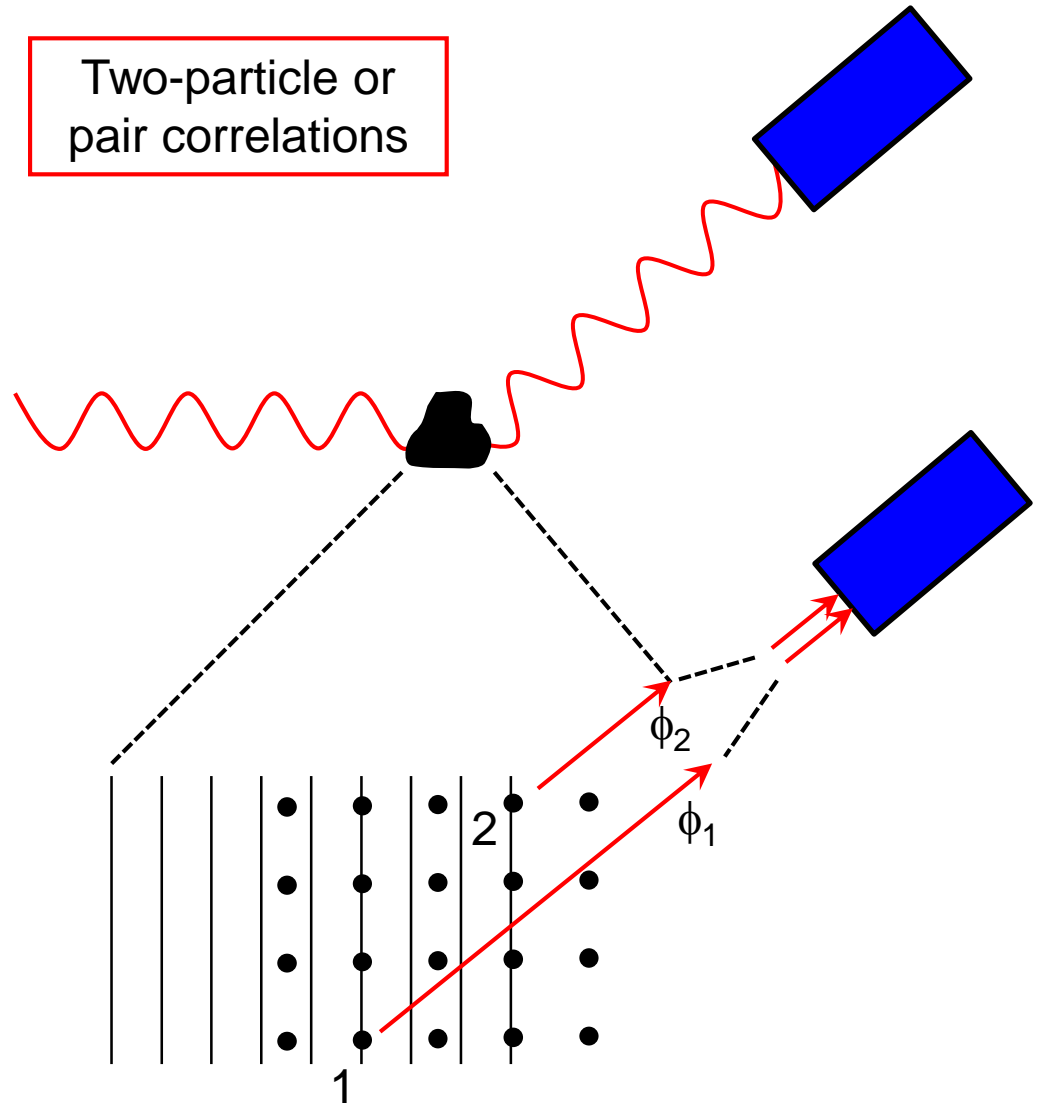


Small angle



Correlation Functions

Two-particle or pair correlations



(1) Born Approximation:

Assumes neutrons scatter only once (single scattering event).

(2) Superposition:

Amplitudes of scattered neutrons ϕ_n add linearly.

$$\Phi_s = \phi_1 + \phi_2 + \dots$$

$$\text{Intensity} = |\Phi|^2 = |\phi_1 + \phi_2 + \dots|^2 = |\phi_1|^2 + |\phi_2|^2 + \dots + \underbrace{\phi_1^* \phi_2^* + \phi_2^* \phi_1 + \dots}_{\text{pair correlations}}$$

Correlation Functions

From Van Hove (1954) ...

The measured quantity Φ_s depends only on time-dependent correlations between the positions of pairs of atoms.

This is true because neutrons interact only weakly with matter. Thus only the lowest order term in the perturbation expansion contributes.

Differential Cross-Section

$$\frac{d\sigma}{d\Omega} = \sum_{i,j} b_i b_j e^{-i(\mathbf{k}_i - \mathbf{k}_j) \cdot (\mathbf{r}_i - \mathbf{r}_j)}$$

Depends only on:
where the atoms are
and
what the atoms are.

Correlation Functions

From Squires (1996):
Introduction to the theory of thermal neutron scattering

Partial Differential Cross-Section

$$\left(\frac{d^2\sigma}{d\Omega dE} \right)_{k_0 \rightarrow k_1} = \frac{1}{N} \frac{k_f}{k_i} \left(\frac{m}{2\pi\hbar^2} \right)^2 \sum p_i p_f \sum | \langle \mathbf{k}_f | V | \mathbf{k}_i \rangle |^2 \delta(E + E_i - E_f)$$

$$\left(\frac{d^2\sigma}{d\Omega dE} \right)_{k_0 \rightarrow k_1} = N \frac{k_f}{k_i} b^2 S(\mathbf{Q}, \omega)$$

Neutron Structure Factor

$$S(\mathbf{Q}, \omega) = \frac{1}{2\pi\hbar} \int G(\mathbf{r}, t) e^{i(\mathbf{Q} \cdot \mathbf{r} - \omega t)} d\mathbf{r} dt$$

Pair Correlation Function

$$G(\mathbf{r}, t) = \left(\frac{1}{2\pi} \right)^3 \frac{1}{N} \int \sum_{jj'} e^{i\mathbf{Q} \cdot \mathbf{r}} \langle e^{-i\mathbf{Q} \cdot \mathbf{r}_{j'}(0)} e^{i\mathbf{Q} \cdot \mathbf{r}_j(t)} \rangle d\mathbf{Q}$$

Fourier Transform

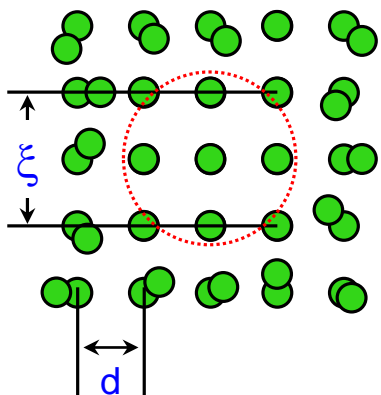
Correlation Functions

KEY IDEA – Neutron interactions are weak → Scattering only probes two-particle correlations in space and time, but does so simultaneously!

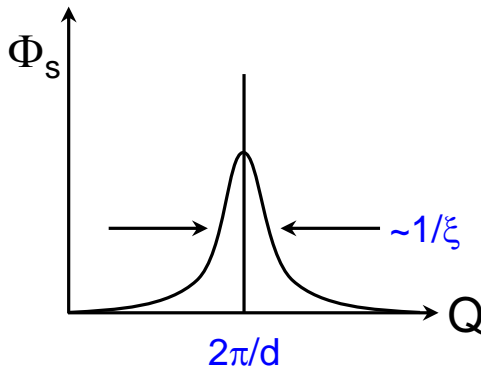
The scattered neutron flux $\Phi_s(\vec{Q}, \hbar\omega)$ is proportional to the space (\vec{r}) and time (t) Fourier transform of the probability $G(\vec{r}, t)$ of finding an atom at (\vec{r}, t) given that there is another atom at $r = 0$ at time $t = 0$.

$$\Phi_s \propto \frac{\partial^2 \sigma}{\partial \Omega \partial \omega} \propto \iint e^{i(\vec{Q} \cdot \vec{r} - \omega t)} G(\vec{r}, t) d^3 \vec{r} dt$$

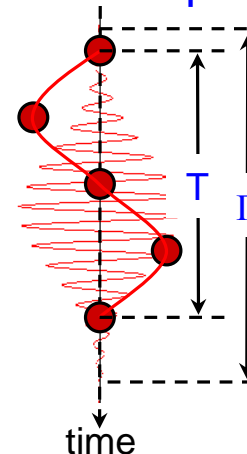
Real space



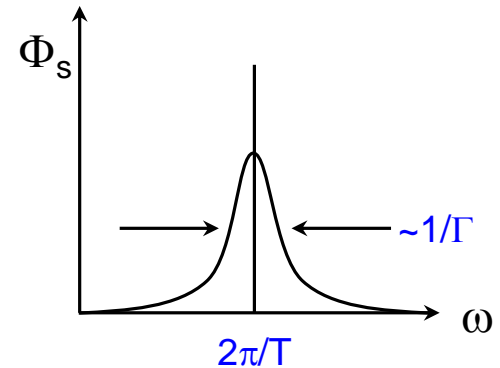
Q-space



Time space



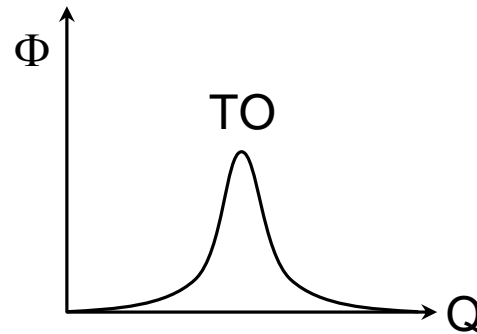
ω -space



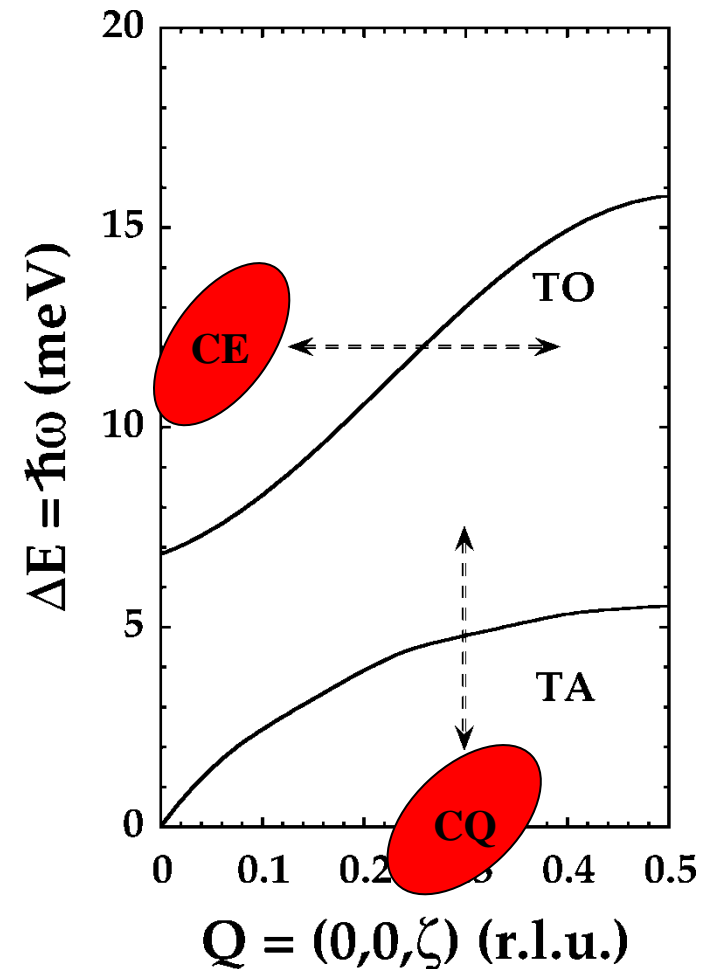
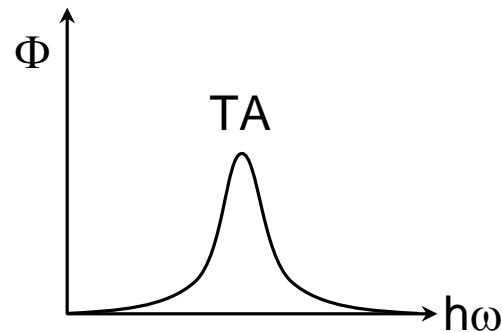
Correlation Functions

There are two main ways of measuring the neutron scattering cross section $S(Q, \omega)$.

Constant-E scans:
vary Q at fixed $\hbar\omega$.



Constant-Q scans:
vary $\hbar\omega$ at fixed Q .

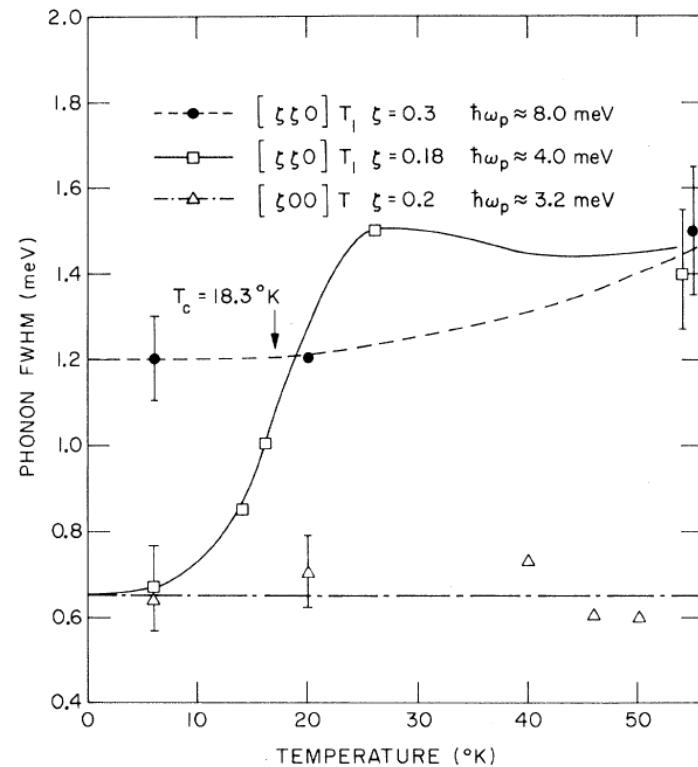


Influence of the Superconducting Energy Gap on Phonon Linewidths in Nb_3Sn

J. D. Axe and G. Shirane

Brookhaven National Laboratory, Upton, New York 11973

(Received 7 December 1972)



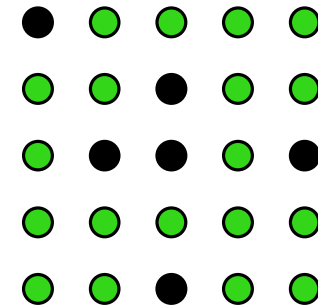
Coherent vs Incoherent

What happens when two different nuclei are randomly distributed throughout the crystal?

This situation could arise for two reasons.

1. Isotopic incoherence
2. Nuclear spin incoherence

Both reasons can occur because the scattering interaction is nuclear.



Recall that
$$\left(\frac{d^2\sigma}{d\Omega dE} \right)_{k_0 \rightarrow k_1} = N \frac{k_f}{k_i} b^2 S(\mathbf{Q}, \omega)$$

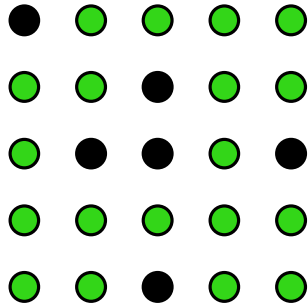
Then the above equation must be generalized:

$$\frac{d^2\sigma}{d\Omega dE_f} = \sum_{i,j} \overline{b_i b_j} S_{ij}(\mathbf{Q}, \omega) \quad \longrightarrow \quad \begin{aligned} \overline{b_i b_j} &= (\overline{b})^2, \text{ for } i=j \\ \overline{b_i b_j} &= \overline{b^2}, \text{ for } i \neq j \end{aligned}$$

— = average

Coherent vs Incoherent

What happens when two different nuclei are randomly distributed throughout the crystal?



Our partial differential cross section can then be recast into the form:



$$\frac{d^2\sigma}{d\Omega dE_f} = \sigma_c S_c(Q, \omega) + \sigma_i S_i(Q, \omega) \quad , \text{ where}$$

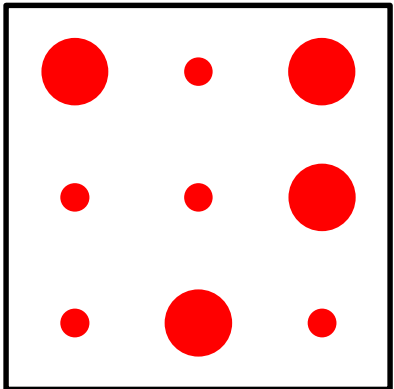
$$\sigma_c = 4\pi(\bar{b})^2 \quad c = \text{coherent}$$

$$\sigma_i = 4\pi\{\bar{b}^2 - (\bar{b})^2\} \quad i = \text{incoherent}$$

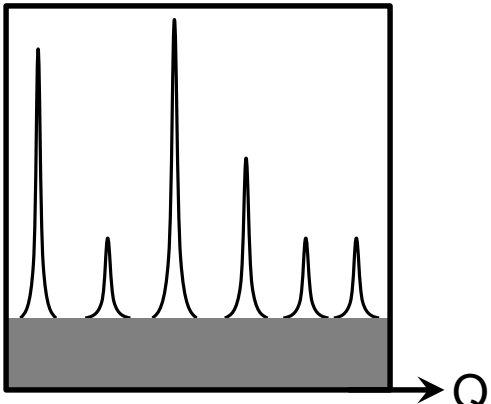
Coherent vs Incoherent

Consider a system composed of two different scattering lengths, b_1 and b_2 .

$b_1 =$ 
 $b_2 =$ 

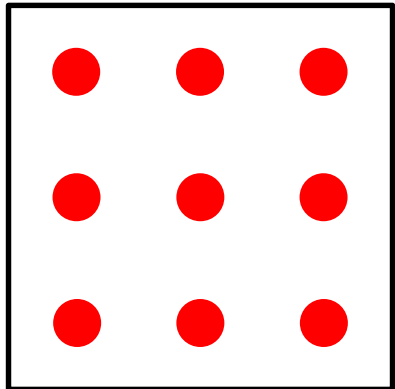


The two isotopes are randomly distributed.

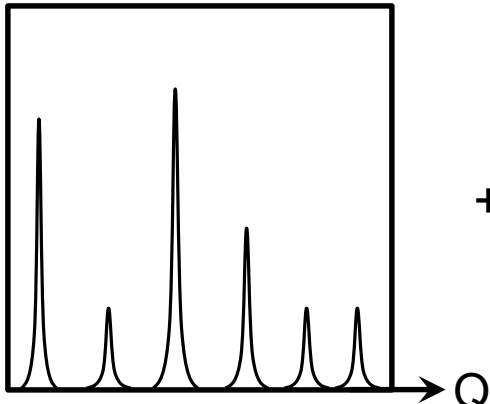


Total scattering

=

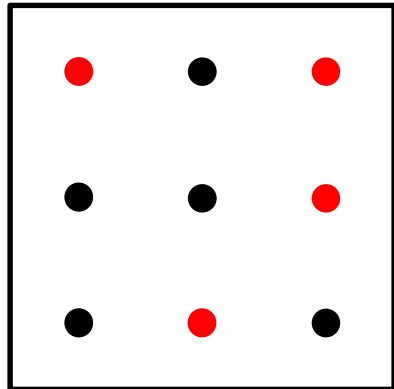


$$\frac{1}{2}(b_1 + b_2) = \bar{b}$$

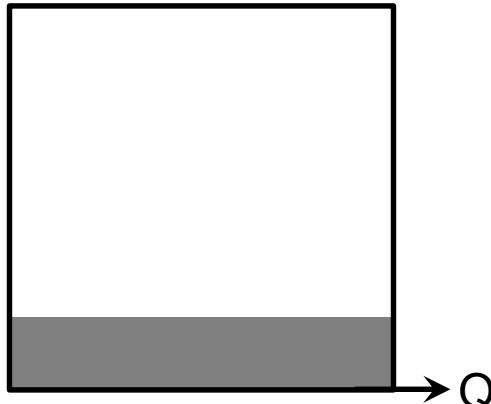


Coherent scattering

+



Deviations δb



Incoherent scattering

=

+

After Andrew Boothroyd
 PSI Summer School 2007

Coherent vs Incoherent

What do these expressions mean physically?

Coherent Scattering

Measures the Fourier transform of the **pair** correlation function $G(r,t) \rightarrow$ interference effects.

This cross section reflects collective phenomena such as:

Phonons

Spin Waves

Incoherent Scattering

Measures the Fourier transform of the **self** correlation function $G_s(r,t) \rightarrow$ no interference effects.

This cross section reflects single-particle scattering:

Atomic Diffusion

Vibrational Density of States

Brief Summary

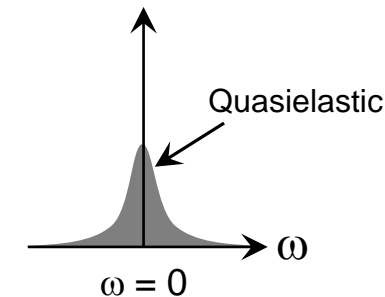
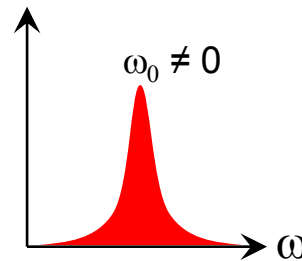
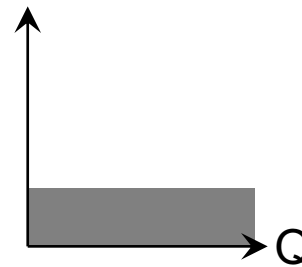
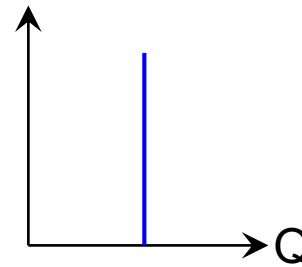
$$\left. \frac{d\sigma}{d\Omega} \right|_{\text{coh}} = \frac{\sigma_{\text{coh}}}{4\pi} S(Q)$$

$$\left. \frac{d\sigma}{d\Omega} \right|_{\text{inc}} = \frac{\sigma_{\text{inc}}}{4\pi}$$

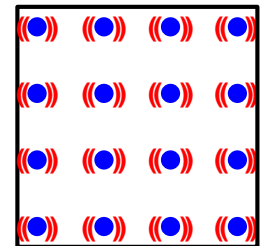
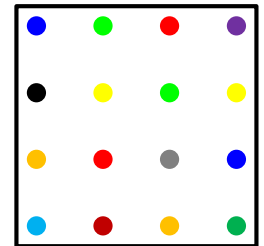
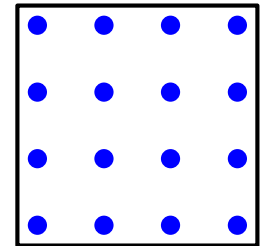
$$\left. \frac{d^2\sigma}{d\Omega dE_f} \right|_{\text{coh}} = \frac{k_f}{k_i} \frac{\sigma_{\text{coh}}}{4\pi} S_{\text{coh}}(Q, \omega)$$

$$\left. \frac{d^2\sigma}{d\Omega dE_f} \right|_{\text{inc}} = \frac{k_f}{k_i} \frac{\sigma_{\text{inc}}}{4\pi} S_{\text{inc}}(Q, \omega)$$

(Q, ω) Space

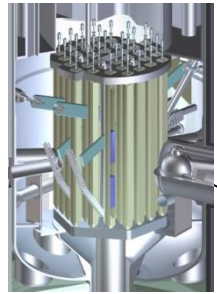


(r,t) Space

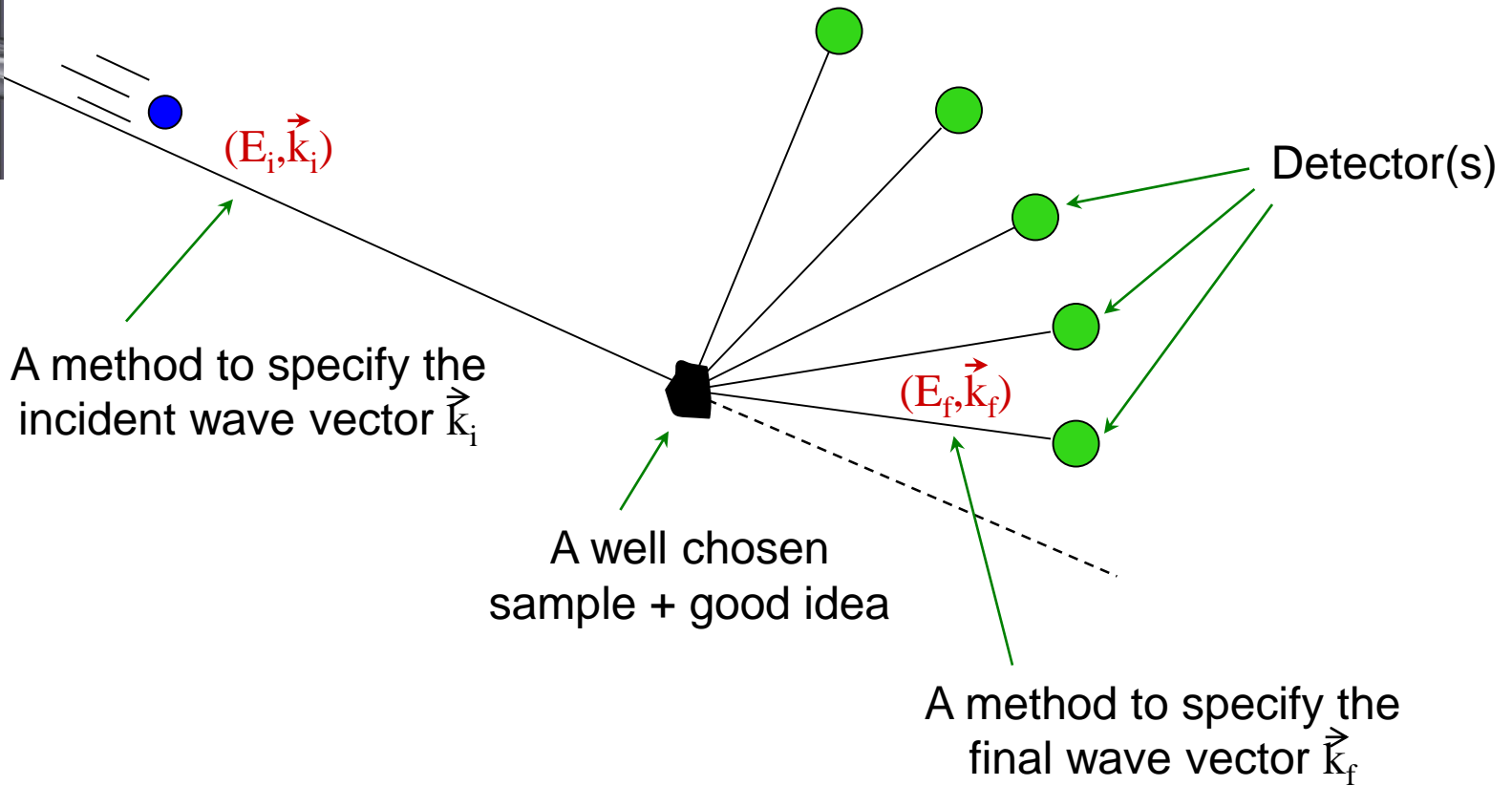


Basics of Scattering

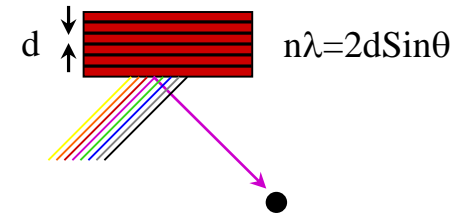
Elements of all scattering experiments



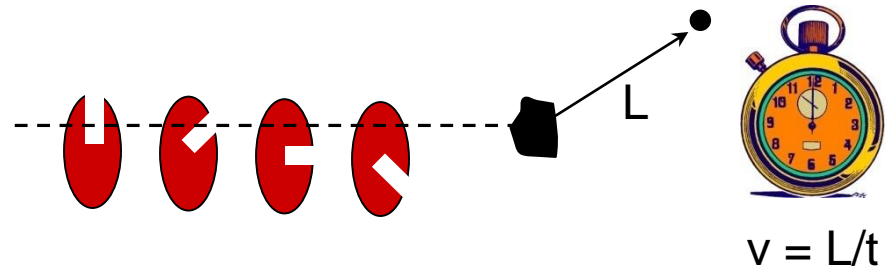
A source



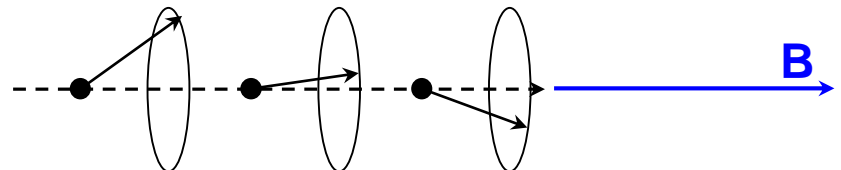
1. Bragg Diffraction



2. Time-of-Flight (TOF)



3. Larmor Precession

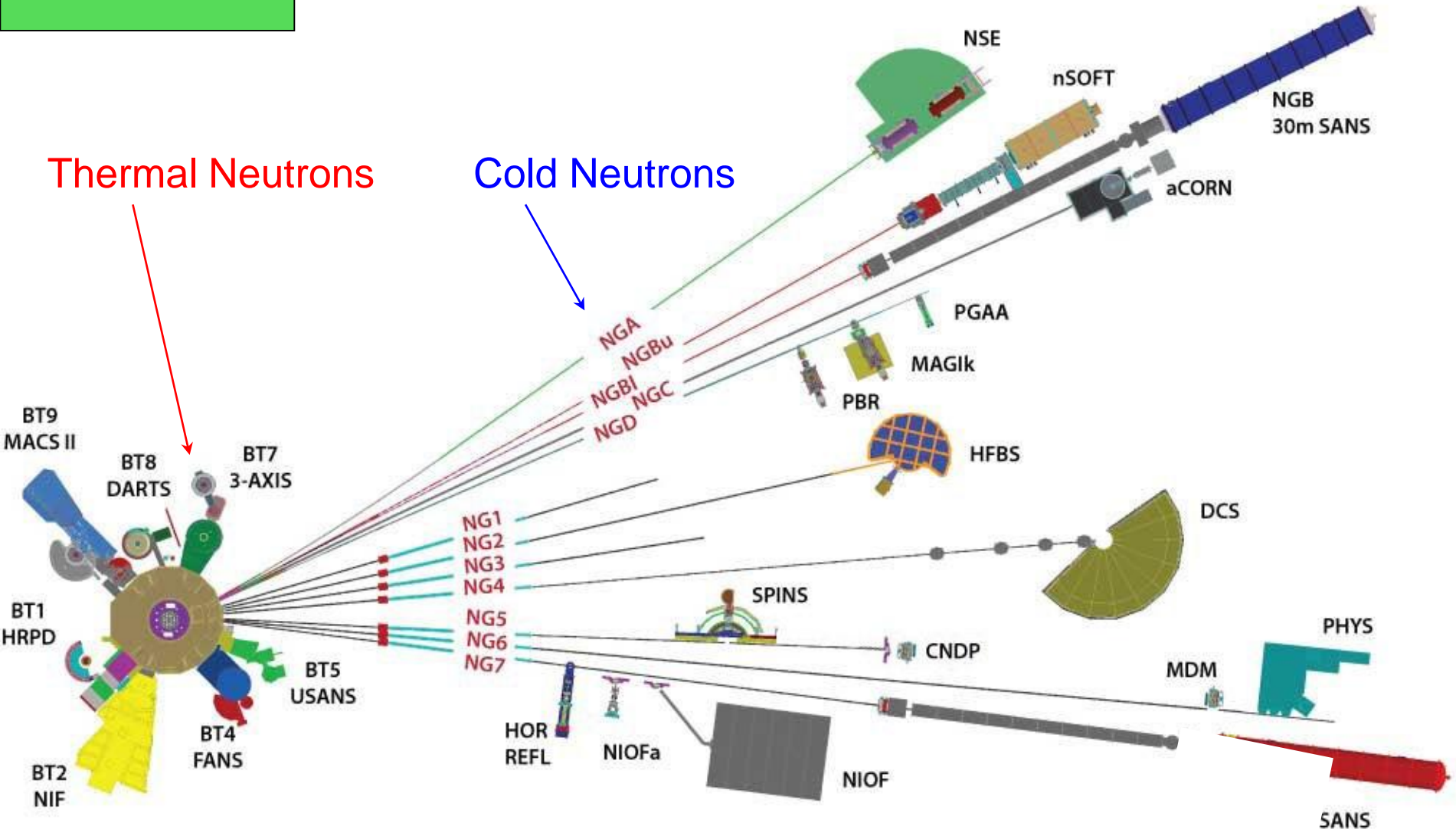


Neutron Instruments

You will see many of these on the tour ...

Thermal Neutrons

Cold Neutrons



- BT9 MACS II
- BT8 DARTS
- BT7 3-AXIS
- BT1 HRPD
- BT2 NIF
- BT4 FANS
- BT5 USANS

- NGA
- NGBu
- NGBI
- NGC
- NGD
- NG1
- NG2
- NG3
- NG4
- NG5
- NG6
- NG7

- NSE
- nSOFT
- NGB 30m SANS
- aCORN
- PGAA
- MAGik
- PBR
- HFBS
- DCS
- SPINS
- CNDP
- MDM
- PHYS
- SANS
- NIOFa
- NIOF
- HOR REFL

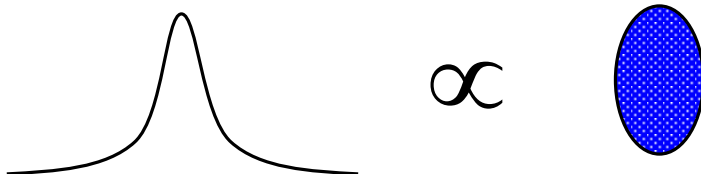
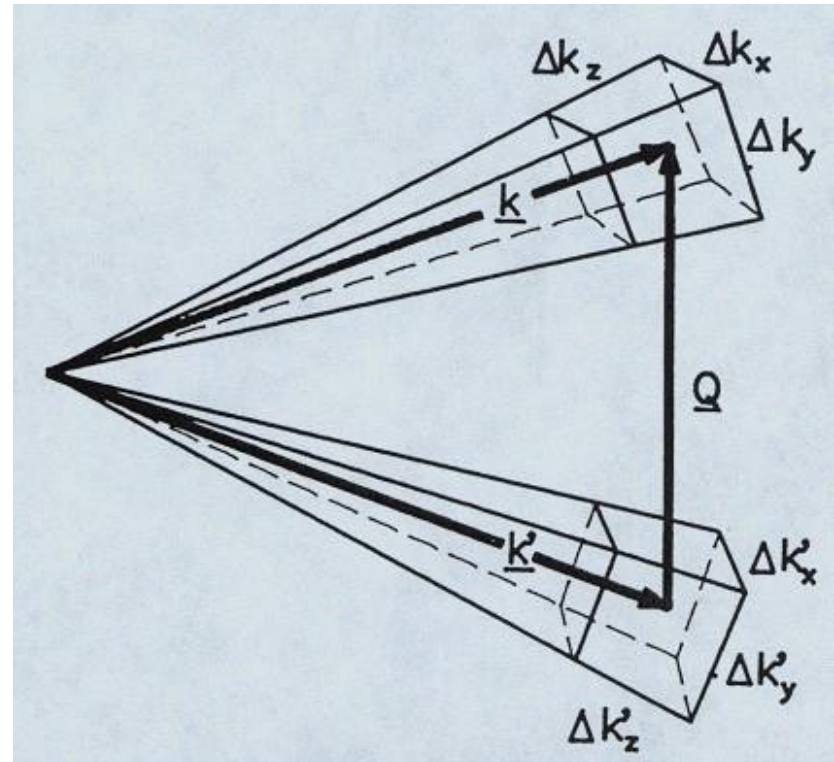
Neutron Instruments

Why so many different kinds?

Because neutron scattering is an intensity-limited technique. Thus detector coverage and resolution MUST be tailored to the science.

Uncertainties in the neutron wavelength and direction imply \mathbf{Q} and $\hbar\omega$ can only be defined with a finite precision.

The total signal in a scattering experiment is proportional to the resolution volume \rightarrow better resolution leads to lower count rates! *Choose carefully* ...



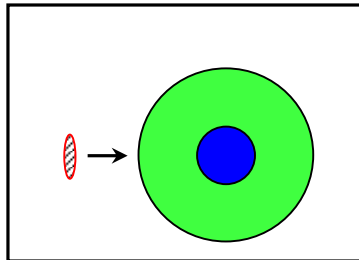
Courtesy of R. Pynn

Momentum Resolution

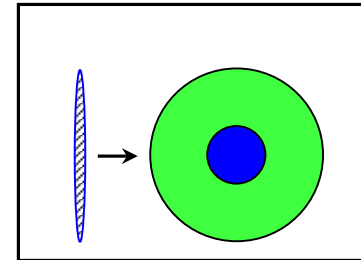
Q-Resolution Matters!

The “right” resolution depends on what you want to study.

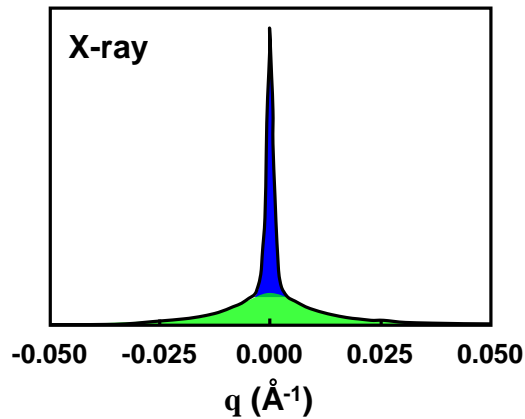
X-ray



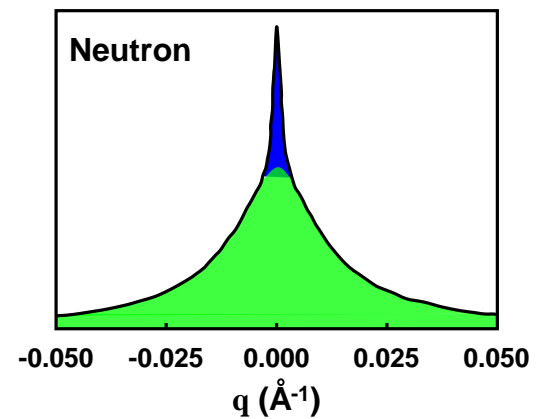
Neutron



X-ray



Neutron

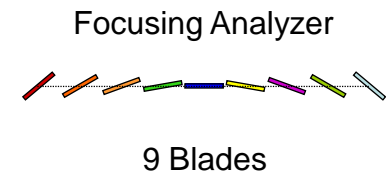
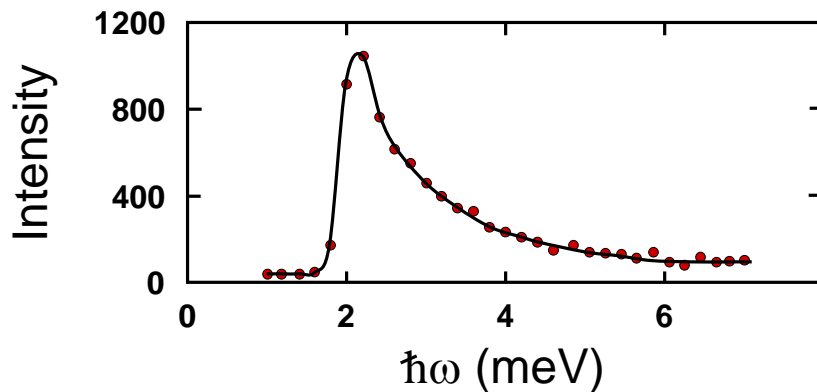
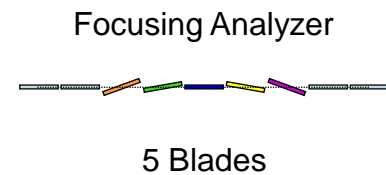
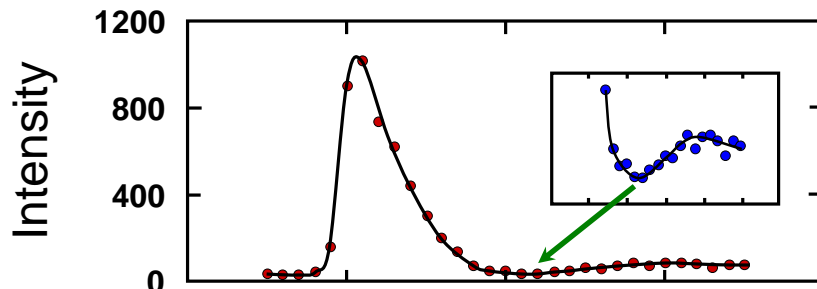
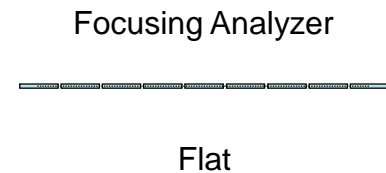
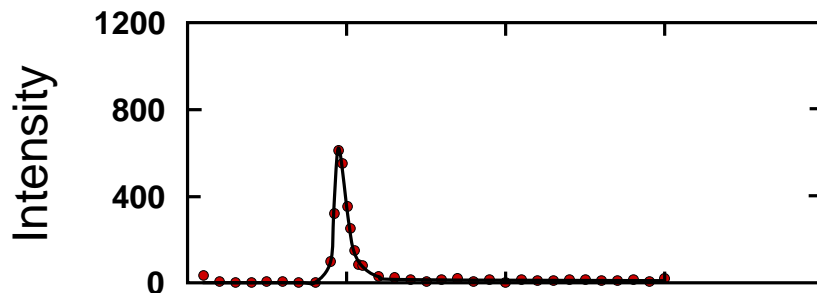
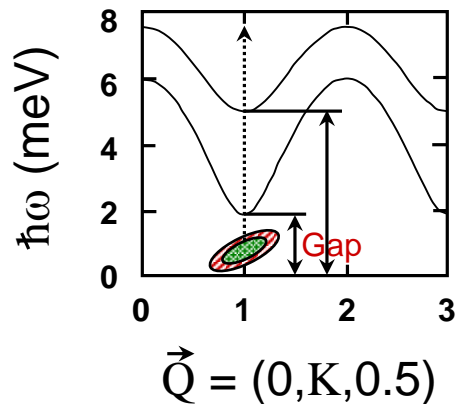


Energy Resolution

$\hbar\omega$ -Resolution Matters!

Example – CuGeO_3

SPINS



Quick Review

Please try to remember these things ...

1 Neutrons scattering probes two-particle correlations in both space and time (simultaneously!).

2 The neutron scattering length, b , varies randomly with $Z \rightarrow$ allows access to atoms that are usually unseen by x-rays.

3 Coherent Scattering

Measures the Fourier transform of the pair correlation function $G(r,t) \rightarrow$ interference effects.

This cross section reflects collective phenomena.

4 Incoherent Scattering

Measures the Fourier transform of the self correlation function $G_s(r,t) \rightarrow$ no interference effects.

This cross section reflects single-particle scattering.

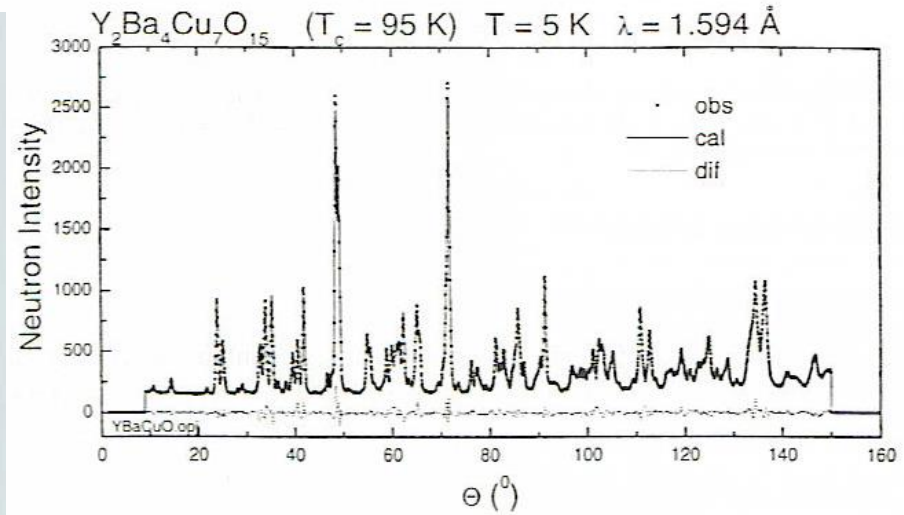
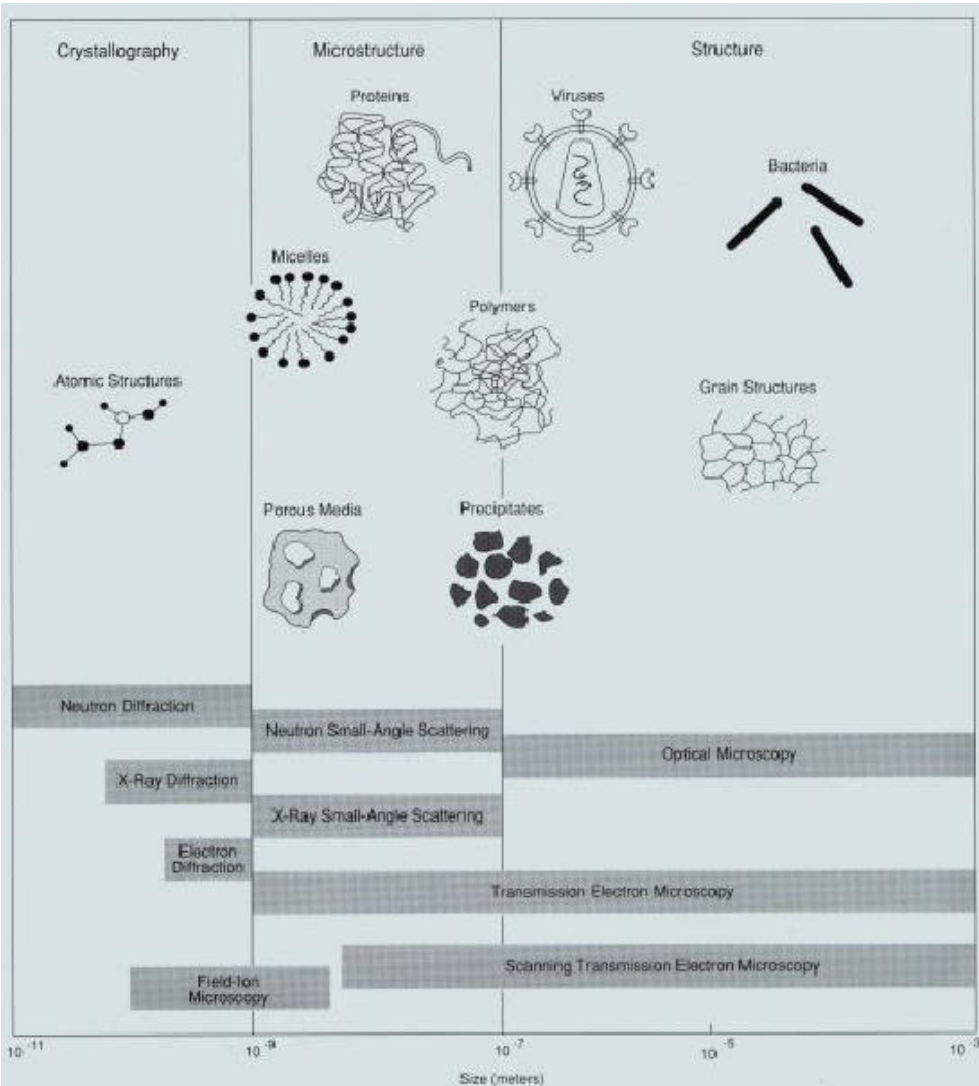
More Examples

OK, after all of this, just exactly what
can neutron scattering do for you?

Let's look at a few examples ...

Elastic Scattering

Neutrons can probe length scales ranging from $\sim 0.1 \text{ \AA}$ to $\sim 1000 \text{ \AA}$



Mitchell et. al, *Vibrational Spectroscopy with Neutrons* (2005)

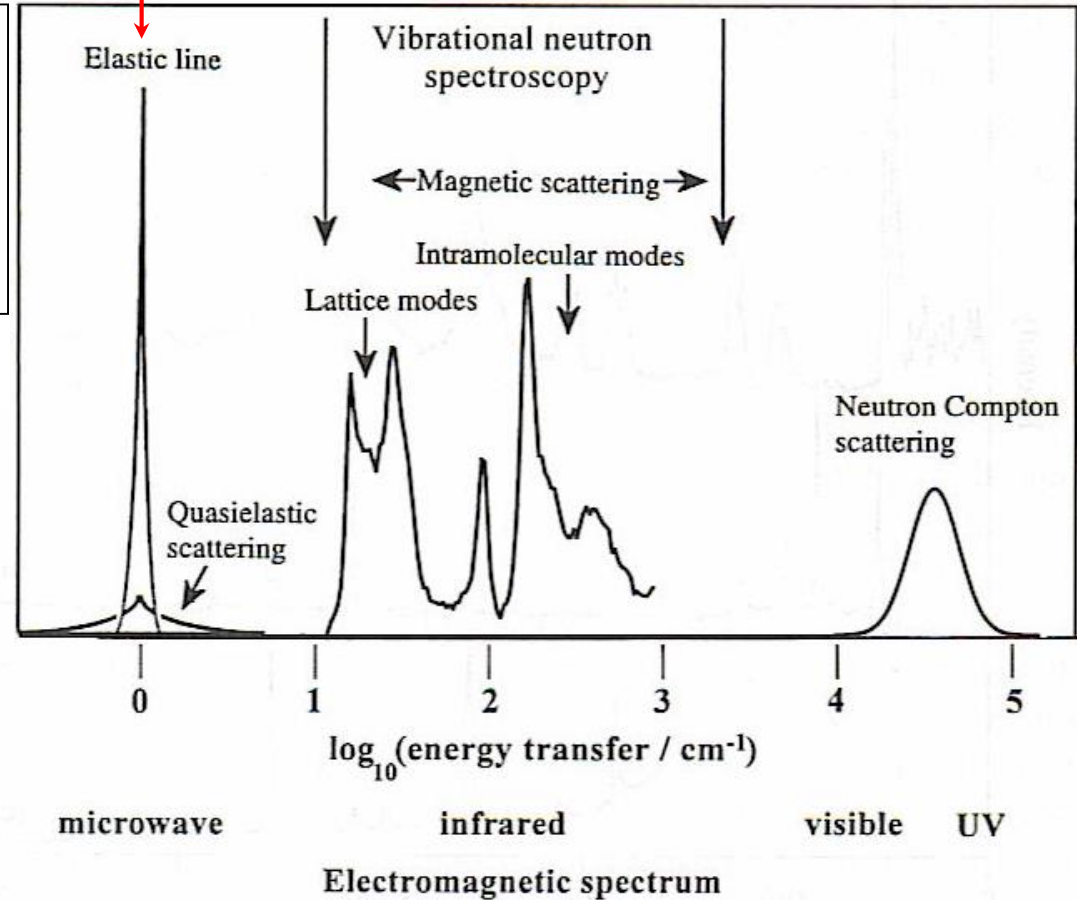
Neutrons needed to determine structure of 123 high- T_c cuprates because x rays weren't sufficiently sensitive to the oxygen atoms.

Inelastic Scattering

Neutrons can probe time scales ranging from $\sim 10^{-14}$ s to $\sim 10^{-8}$ s.

Probes the vibrational, magnetic, and lattice excitations (dynamics) of materials by measuring changes in the neutron momentum and energy simultaneously.

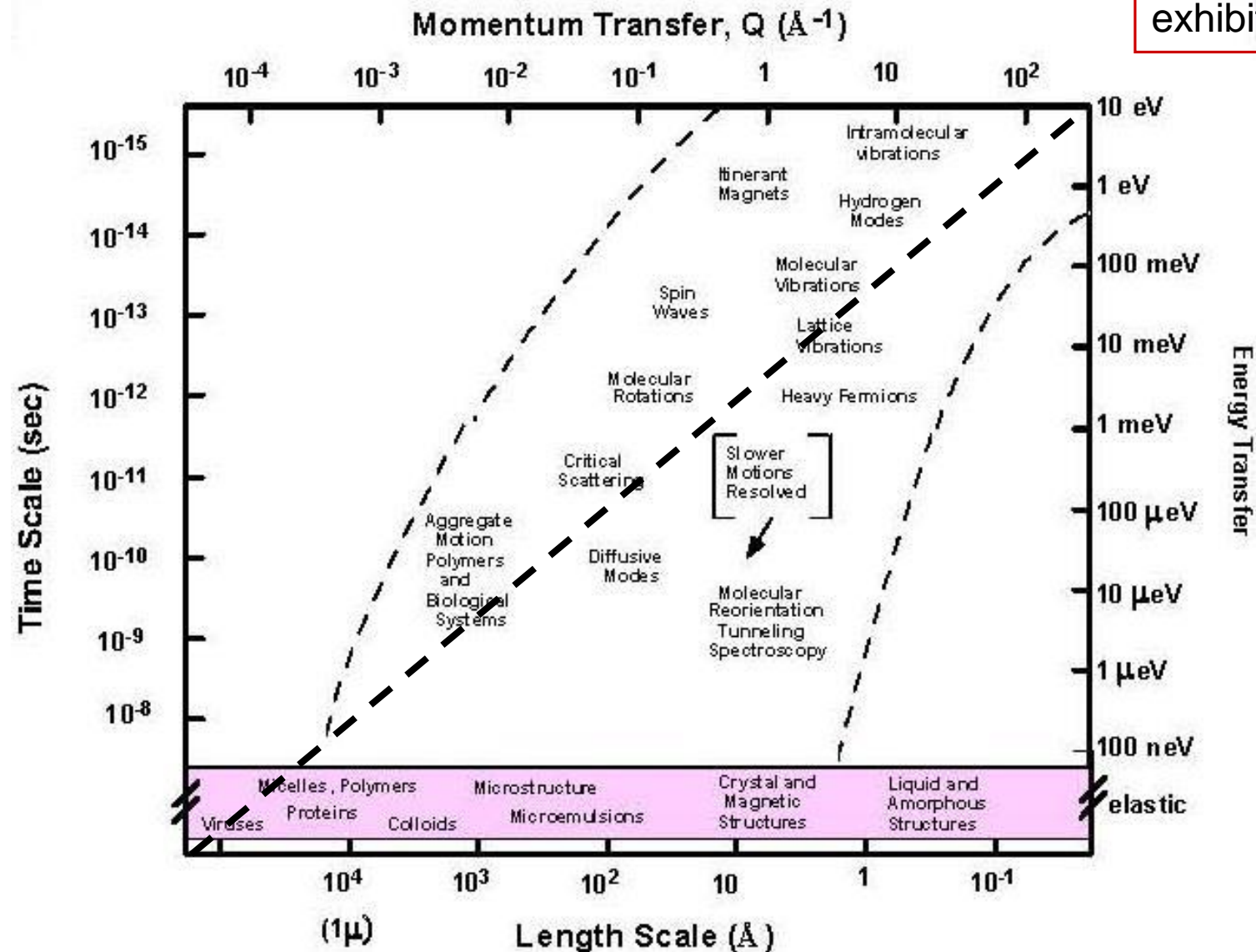
$$\hbar\omega = 0$$



Length and Time Scales

Do you see a pattern here?

Larger "objects" tend to exhibit slower motions.



Elastic Scattering

Pop Quiz

Can one measure elastic scattering from a liquid?



If Yes, explain why?
If No, explain why not?

Hint: What is the correlation of one atom in a liquid with another after a time t ?



Nobel Prize
in Physics
1994

The Fathers of Neutron Scattering

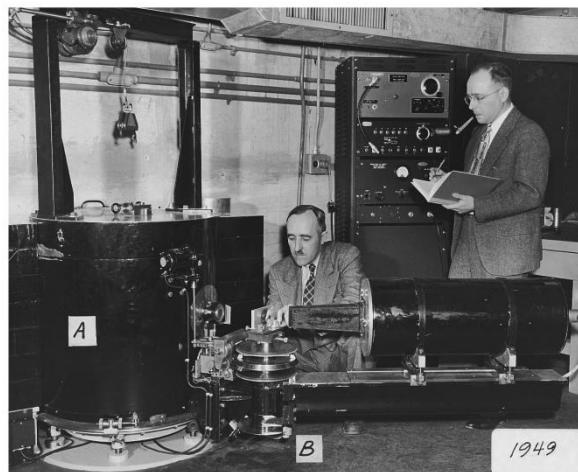
“For pioneering contributions to the development of neutron scattering techniques for studies of condensed matter”

“For the development of the neutron diffraction technique”



Clifford G Shull
MIT, USA
(1915 – 2001)

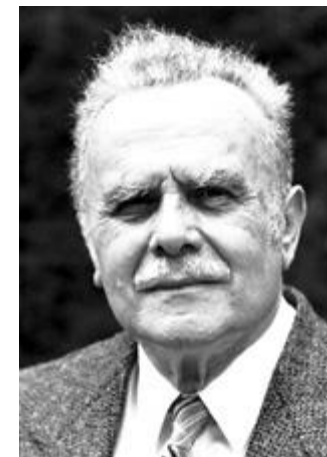
Showed us where
the atoms are ...



Ernest O Wollan
ORNL, USA
(1910 – 1984)

Did first neutron
diffraction expts ...

“For the development of neutron spectroscopy”



Bertram N Brockhouse
McMaster University, Canada
(1918 – 2003)

Showed us how
the atoms move ...

Useful References

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- G. L. Squires, Cambridge University Press
- “Theory of Neutron Scattering from Condensed Matter”
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- “Neutron Diffraction” (Out of print)
- G. E. Bacon, Clarendon Press, Oxford
- “Structure and Dynamics”
- M. T. Dove, Oxford University Press
- “Elementary Scattering Theory”
- D. S. Sivia, Oxford University Press