

Dynamics of Methane Adsorbed on MOFs by Disk Chopper Spectrometer



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Motivation

➤ Adsorbed natural gas technology
Material-based storage

- Low pressure
- Light weight and portable
- Safe
- Cheaper than compressed natural gas technology

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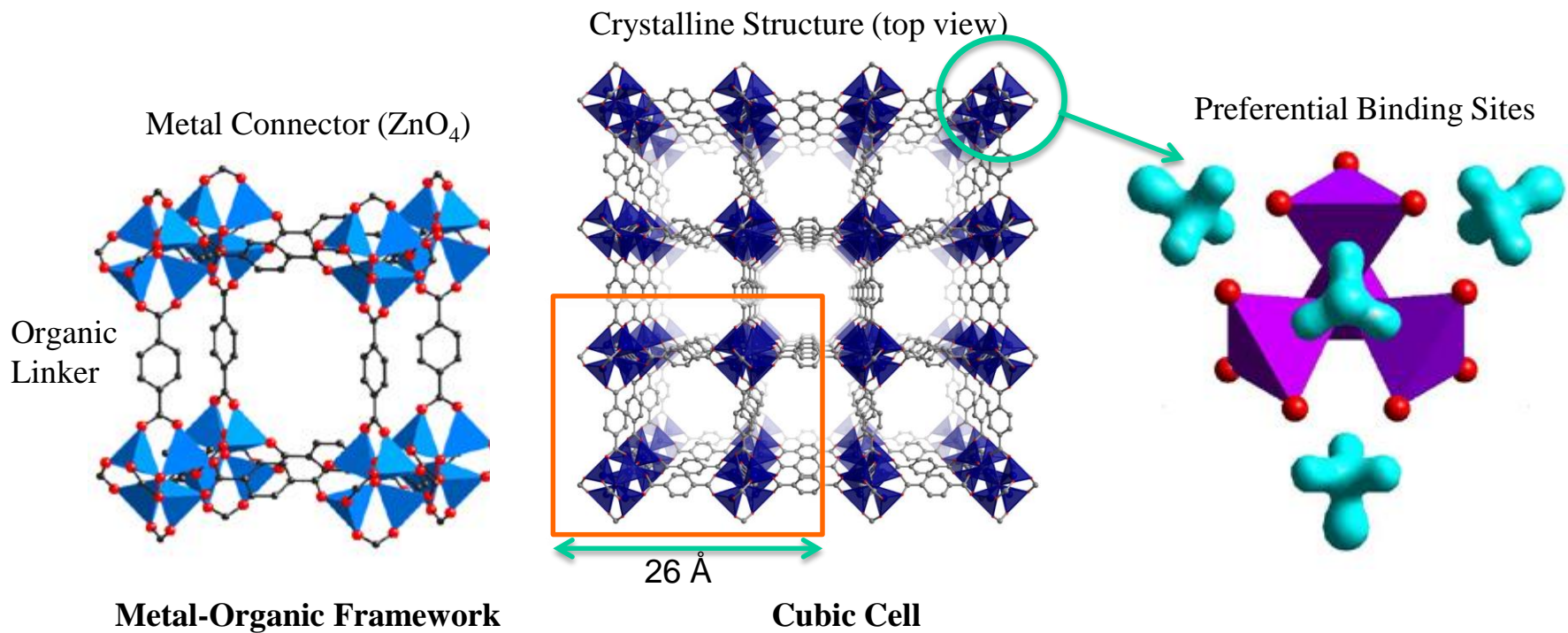
- DOE's Methane Opportunities Vehicular Energy (MOVE) projects are finding innovative ways to create natural gas storage tanks.



Porous Metal-Organic Frameworks (MOFs) for methane storage

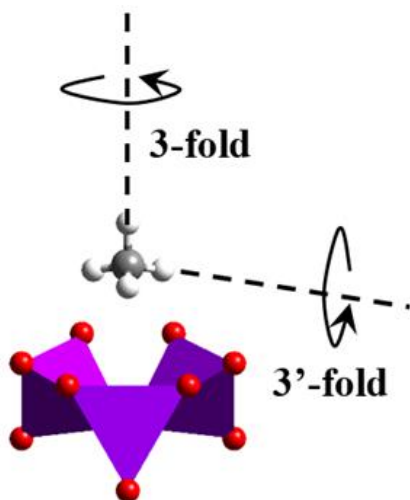


MOF-5 System



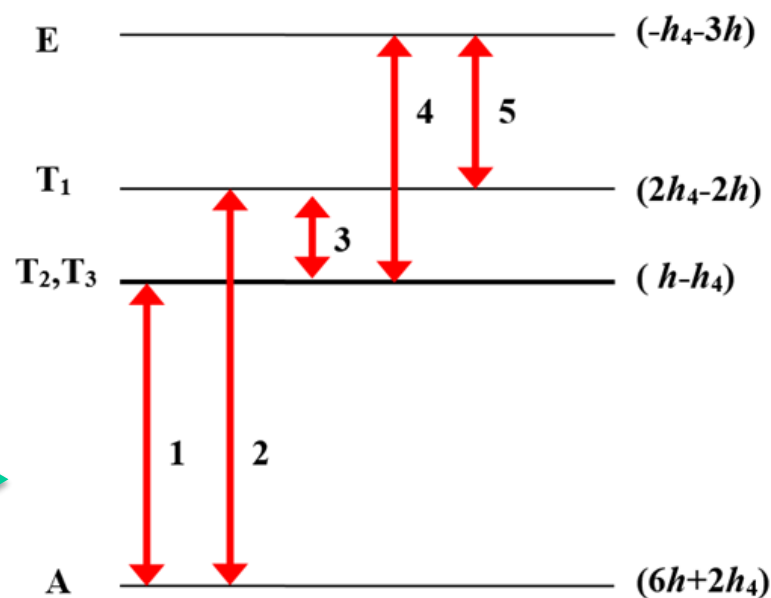
Each cubic unit cell contains **32 CH_4 molecules** on preferential sites. Given the C-H bond length ($\sim 1 \text{ \AA}$), if the surface area are fully occupied by CH_4 molecules, **1 cm^3 MOF-5** have surface area $\sim 2.3 \times 10^3 \text{ m}^2$ which can accommodate **1/6 mol CH_4 molecules**.

MOF-5 System



Adsorbed methane molecule
a hindered rotor

Symmetry
Breaking →

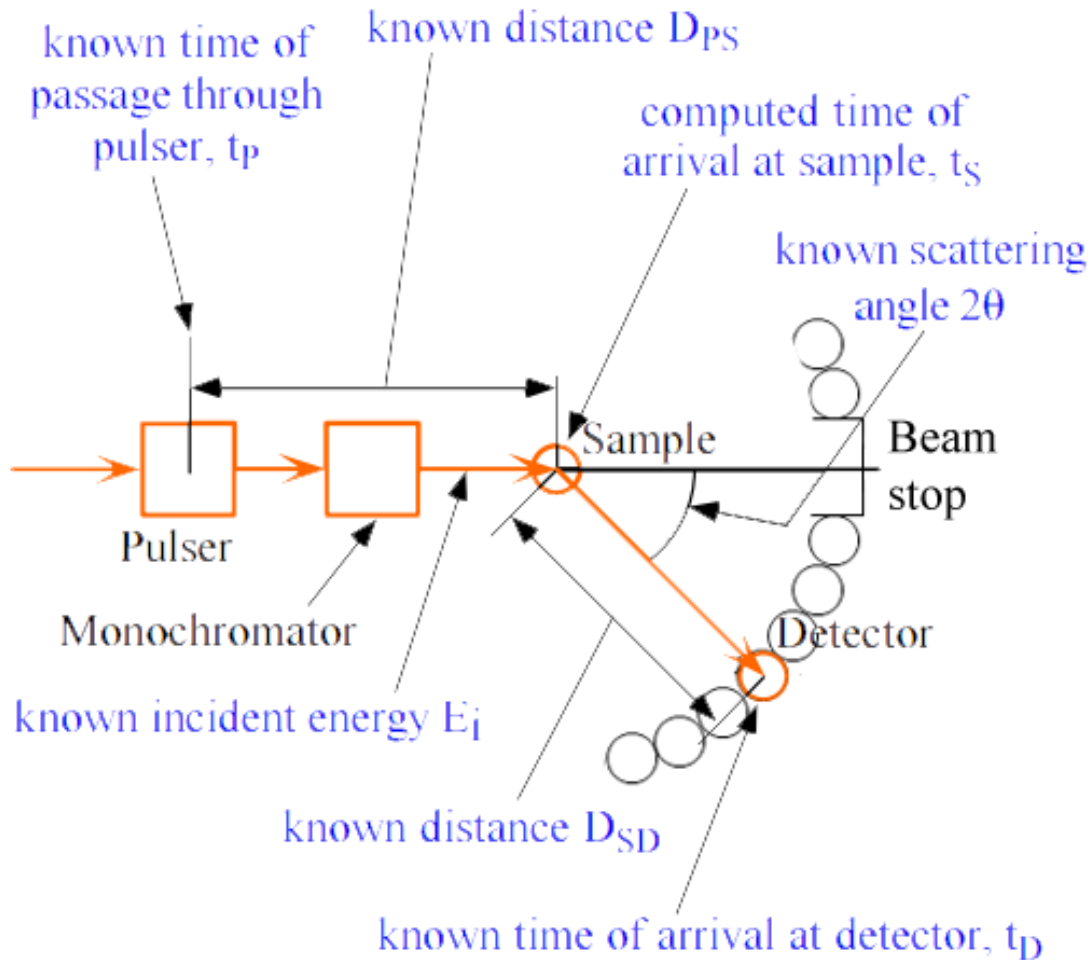


Ground Rotational Energy Splitting

Neutron spectroscopy: various vibrational and rotational motions of adsorbed CH₄ molecules, e.g.,

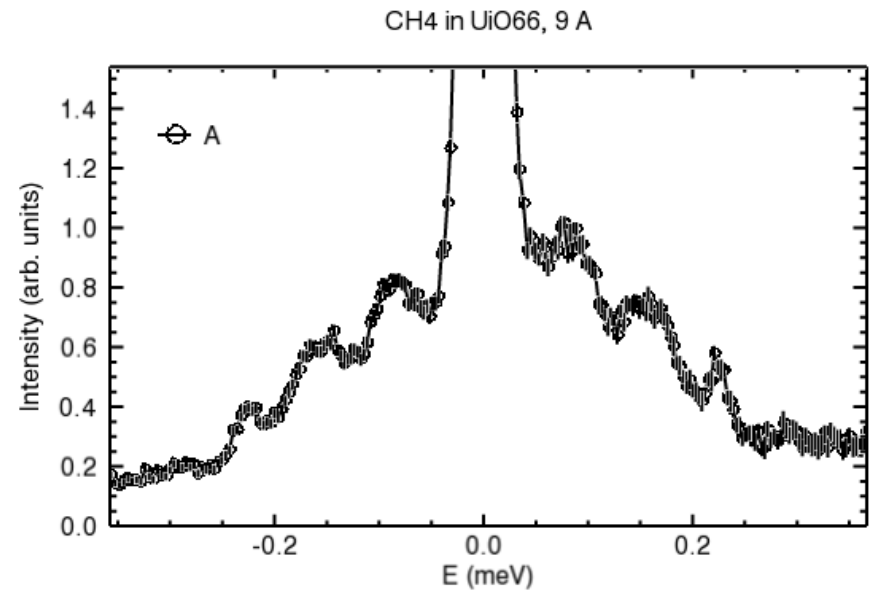
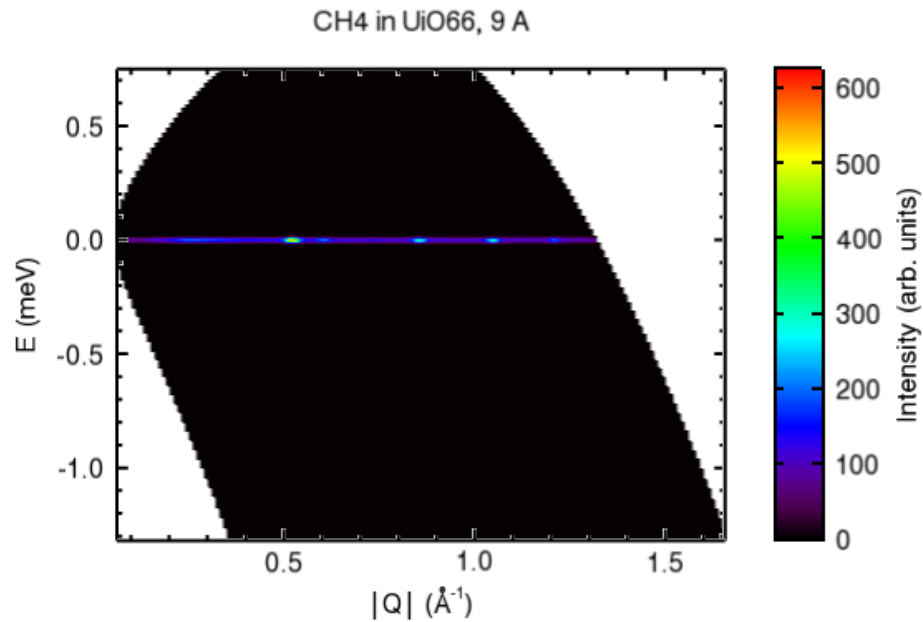
- CH₄ center-of-mass motion (phonons), ~1-20 meV, DCS
- CH₄ quantum rotational tunneling (at low T), ~1 - 600 ueV, DCS
- CH₄ jump diffusion (at high T), DCS

Disk Chopper Spectrometer (DCS)



- Inelastic neutron scattering
- Choppers: Select initial energy of neutrons incident on sample
- Neutrons scattered by sample gain or lose energy
 - Time-of-flight

Measured Intensity

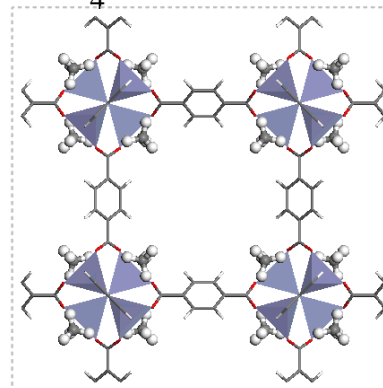
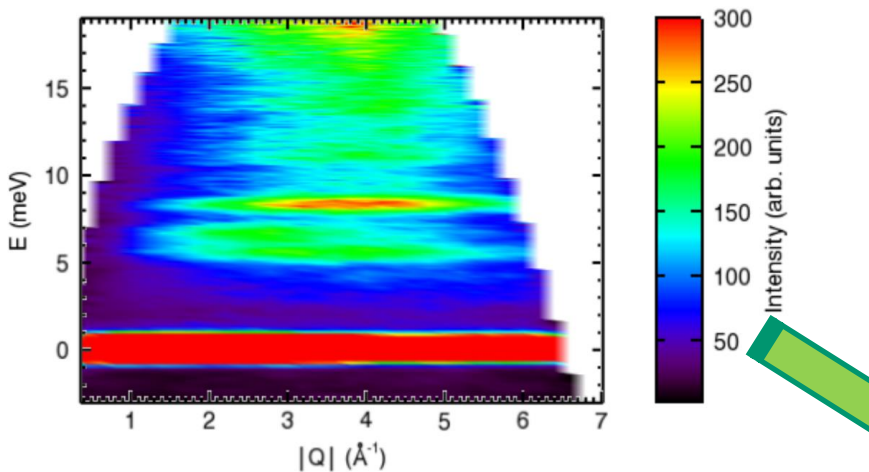


- Inelastic peaks \rightarrow Transition energies between tunneling levels

Phonons on CH₄ MOFs

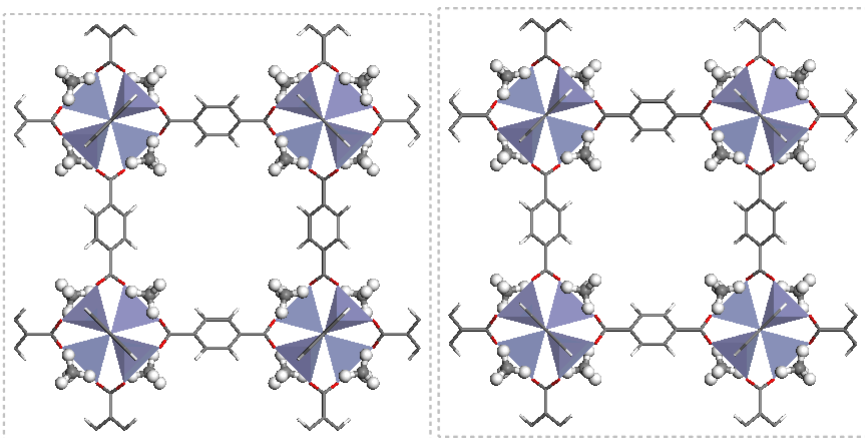
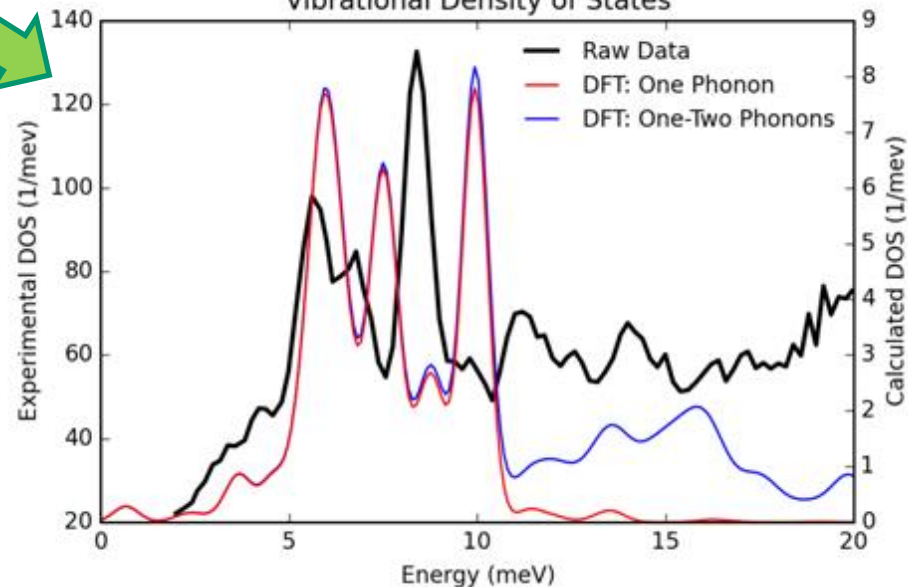
➤ Deuterated organic framework allows us to see the collective CH₄ framework motions.

CH₄ in D-MOF-5, 1.81Å

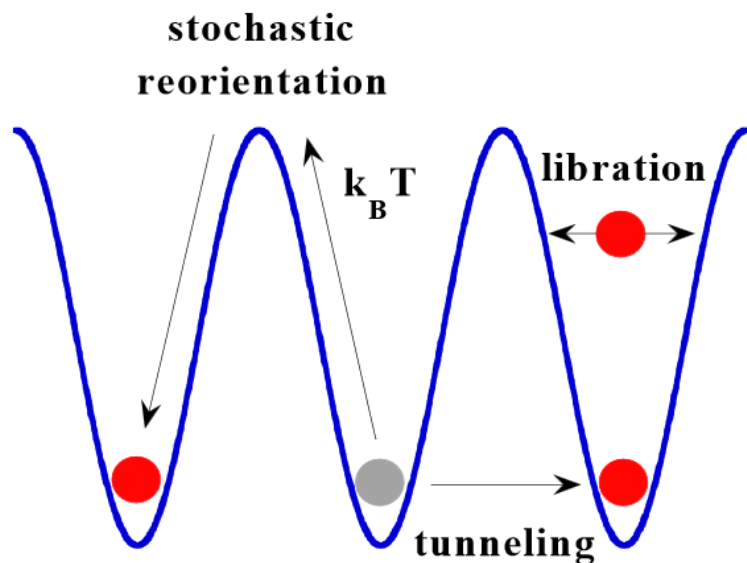


Intensity \approx Weighted phonon density of states (integrated over all Q)

Vibrational Density of States

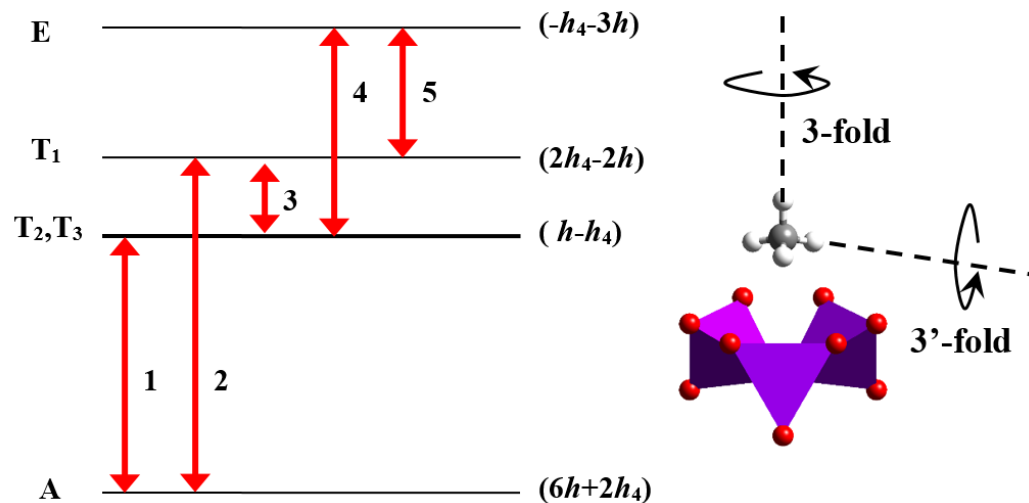


Tunneling of CH₄ in MOF



(Image credit: Rob Dimeo)

Due to Van der Waals interaction between CH₄ and MOF, the CH₄ experiences a potential which hinders rotation.



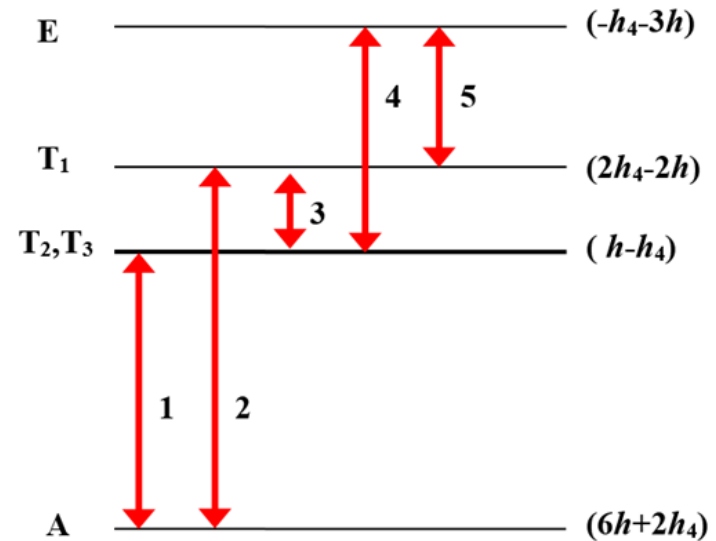
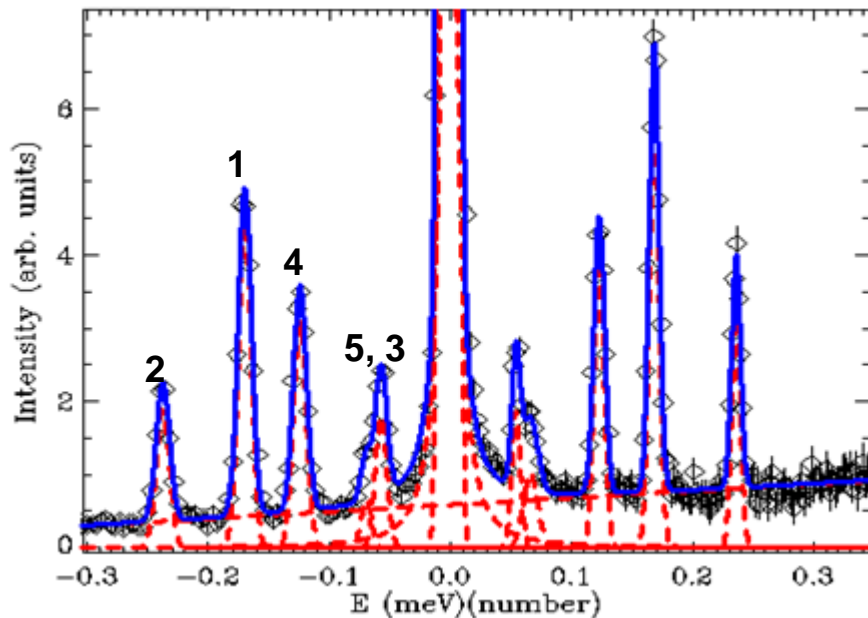
There are **12 positions** of the CH₄ which are degenerate in the gas phase. In potential caused by the MOF, the ground state splits into **4 energy levels**, with a total of **5 possible transitions**.

Smalley & Huller 1981

Tunneling of CH₄ in MOF-5

- Inelastic peaks give the allowed transitions between energy levels.
- Data taken at 6K, scattering from other processes minimal.

CH₄ in D-MOF-5, 9 Å

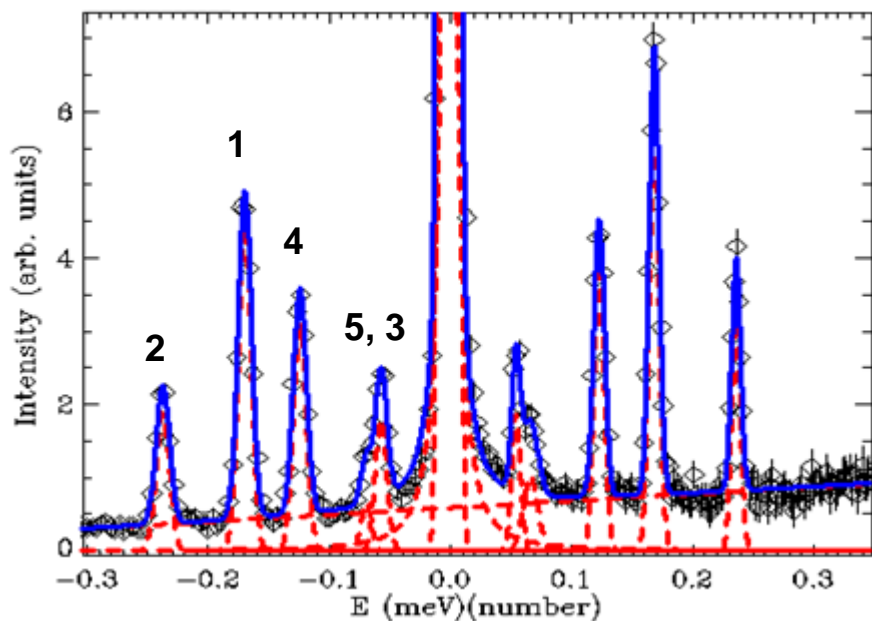


- The transition energies can be used to calculate the potential barrier in the over which the methane rotates.
- Barriers are **~23 meV** for 3 fold rotation, **~17 meV** for 3' fold rotation.
- Predicted barriers from DFT-D are 47 meV and 25 meV

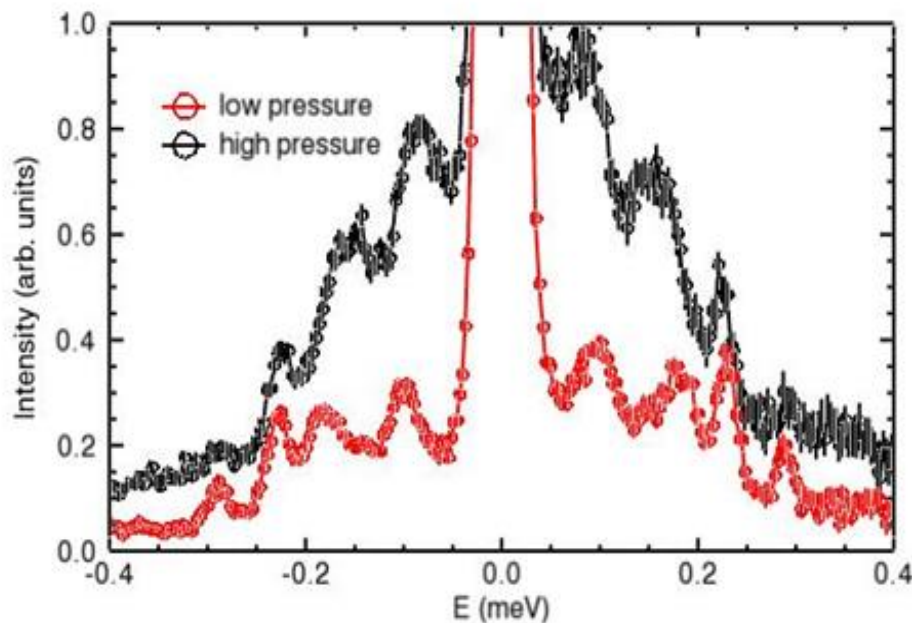
Comparison of Tunneling of CH₄ in MOF-5 and UiO-66

- Data taken at 6K; all CH₄ absorbed
- Charged sample container
- Concentrations: 1 CH₄ per 1 Zn for MOF-5; 0.5 CH₄ per Zr UiO-66; 1 CH₄ per Zr UiO-66

CH₄ in D-MOF-5, 9 Å



CH₄ in UiO-66, 9 Å



Applications of DCS

➤ Diverse Phenomena

- Low energy vibrational and magnetic excitations
- Translational and rotational diffusion processes

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➤ Various Materials

- Magnetic and ferroelectric materials
- Organic molecules
- Molecular crystals

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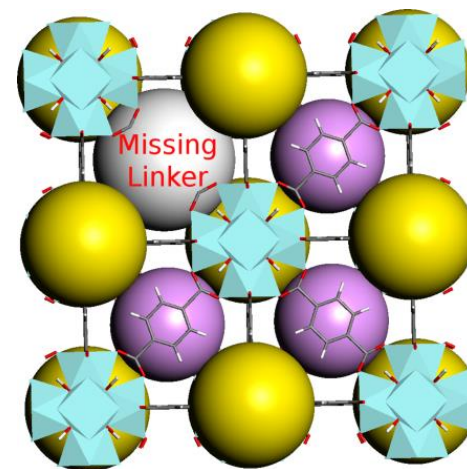


Conclusions

- DCS is used to investigate the rotational dynamics in CH₄-MOF systems, and to understand the CH₄-MOF interactions.
- By analyzing the inelastic neutron scattering spectra, the transitions and rotational barriers of MOF-5 were determined.
- UiO-66 exhibits broadened inelastic peaks with respect to MOF-5 due to its structural defects.
- The experimental results are a useful comparison to DFT calculations.

Future Directions

- Analyze higher quality samples of UiO-66
- Improve DFT methods
- Rational design towards improved hydrocarbon storage



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▪ Groups A and C



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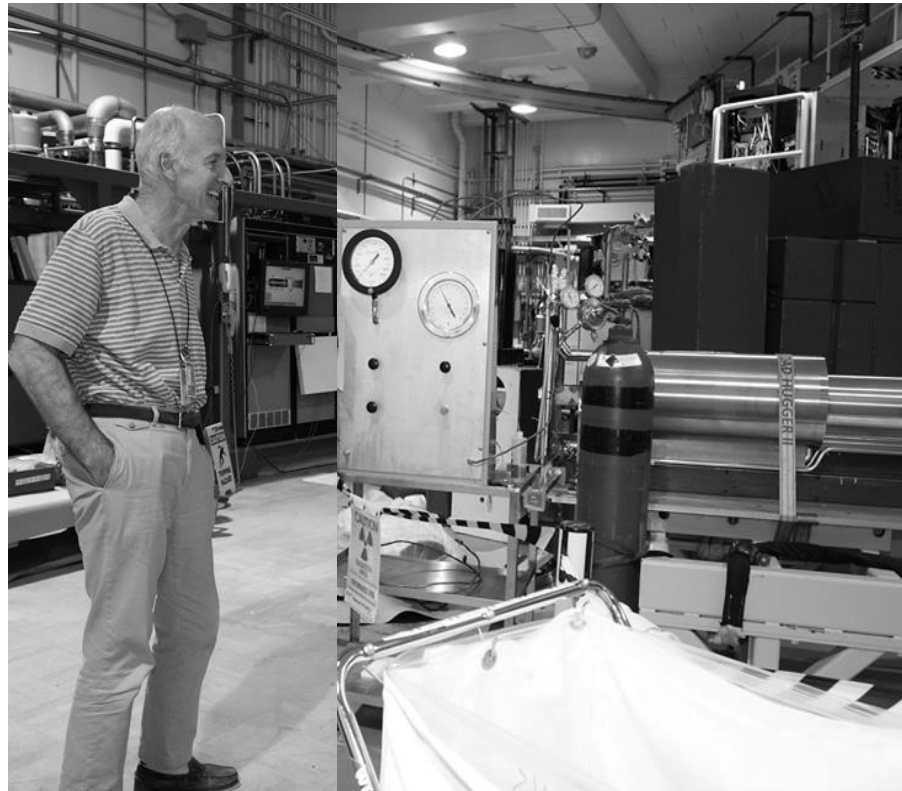


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Q&A

