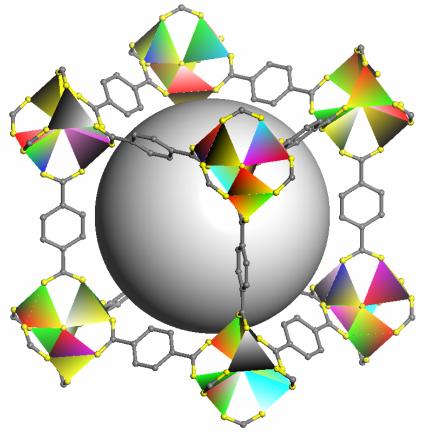
Gas Adsorption in Metal Organic Frameworks: an experiment using the NCNR Disk Chopper Spectrometer

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Abstract

Time-of-flight neutron spectroscopy will be used to examine the local potential of methane adsorbed in a metal-organic-framework (IRMOF-1). This experiment illustrates the important techniques of neutron tunneling and quasielastic neutron scattering (QENS). We shall discuss all aspects of the experiment, from sample preparation and the choice of instrumental setup through to data treatment and interpretation of results.



I. Introduction

Organic-inorganic hybrid materials have been known under numerous names for decades, in particular they have been studied extensively because of possible low-dimensional magnetism. Recent work on robust, high surface area, open pore materials has given rise to a new classification of materials called metal-organic frameworks or MOFs. They are three-dimensional porous structures composed of metallic centers coordinated in various ways, via multi-functional organic components. The practical interest in these materials stems from their innate capacities for gas storage.

Emphasis has been placed upon creating and expanding the hydrogen economy by the highest levels of government. It is apparent that one of the keys to being successful in meeting this "Grand Challenge" will be the creation of methods for the safe storage and retrieval of hydrogen. State-of-the-art materials currently do not satisfy the guidelines issued to make a commercially useful hydrogen storage system for competitive hydrogen-powered transportation. There are many research groups pursuing this research track, but there are already commercially viable storage materials for other energy carriers. A prime example is MOF5 (alternatively known as IRMOF-1, one of a series of isoreticular MOFs) composed of inorganic ZnO₄ tetrahedra linked together by *1,4-benzenedicarboxylate* ligands (see cover). MOF5 only has a modest hydrogen storage capacity at 77 K, but the methane adsorption capacity in a fixed volume can be almost twice what would be achievable in the absence of the MOF [Eddaoudi, Science, 295, 469, 2002].

Recent neutron diffraction studies have elucidated the hydrogen and methane adsorption sites in numerous compounds of interest for energy storage use. Multiple adsorption sites are apparent with increasing methane loading amounts in MOF5 [http://www.ncnr.nist.gov/AnnualReport/FY2006/AR2006.pdf#page=16] mimicking that observed for hydrogen adsorption in the same material [Yildirim, Phys. Rev. Letts. (2005) 95, 215504]. The first and strongest binding sites for methane lie in the so-called cupsite of the ZnO₄ clusters with the orientation of the methane molecule being well resolved, where one of the CH-bonds is along the 3-fold axis (figure 1).

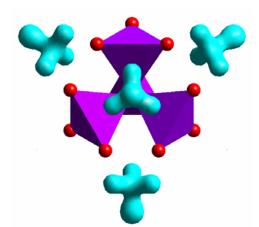


Figure 1.

Excess neutron scattering-length density of the methane loaded sample compared to the bare MOF5. The experimentally derived density is superimposed with the ZnO_4 clusters of the MOF5 host structure, indicating the location of cup-sites for the first methane adsorption.(lattice: Fm-3m and a = 25.91 Å).

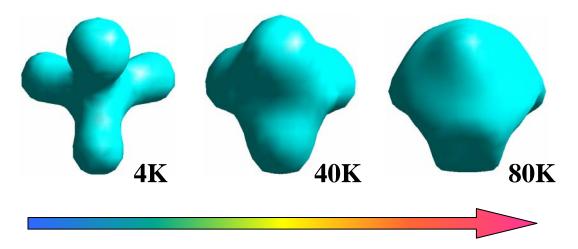


Figure 2. The excess neutron scattering-length density changes as a function of temperature but is still localized.

We will be using the Disk Chopper Spectrometer at the National Institute of Standards and Technology Center for Neutron Research to probe dynamics and the local potential at the methane adsorption site. After reading the background material you should be able to choose an appropriate spectrometer configuration to probe the rotational tunneling or temperature dependent diffusional dynamics present in the methane:MOF5 system.

The neutron has several properties that enable scattering experiments to measure properties of materials that other techniques can measure with much less precision or not at all. Neutrons with wavelengths on the order of interatomic spacings also possess energies on the same order as those characteristic of phonons and intermolecular interactions; for example, a 1.8 Å neutron has an energy of ~25 meV (~200 cm⁻¹) and speed ~2200 ms⁻¹. This means that structural and temporal information can be measured simultaneously.

The reader is reminded that the scattering of neutrons is usually treated as the sum of two parts, known as <u>coherent</u> and <u>incoherent</u> scattering. To understand why such a separation is performed recall that the strength of the scattering from nuclei of the same element can vary (and generally does vary) with spin and/or isotopic species. Hence when a neutron is scattered by a collection of nuclei the interference between the different scattered waves is normally neither complete nor completely absent. For this reason the <u>double</u> <u>differential cross section</u> [$d^2\sigma/d\Omega d\omega$], which describes the probability that neutrons are scattered into solid angle $d\Omega$ and energy transfer window $d(\hbar\omega)$, is normally separated into two terms. The first term is the coherent part, which contains all of the interference effects such as Bragg scattering and small angle scattering. The second term is the incoherent scattering, which represents the scattering from individual nuclei and is approximately isotropic. For a single element [$d^2\sigma/d\Omega d\omega$] can be expressed as

$$\frac{d^2\sigma}{d\Omega d\omega} = \frac{1}{4\pi} \frac{k_f}{k_i} [\sigma_{coh} S(Q, \omega) + \sigma_{inc} S_{inc}(Q, \omega)]$$
 (1)

where k_i and k_f are the magnitudes of the initial and final neutron wavevectors, σ_{coh} and σ_{inc} are the coherent and incoherent scattering cross sections, and $S(Q,\omega)$ and $S_{inc}(Q,\omega)$ are the corresponding scattering functions which depend only on the momentum transfer $\hbar Q$ (or wave vector transfer Q) and the energy transfer $\hbar \omega$. (Note that in general Q is a vector but since we shall be working with a powder with cubic crystal symmetry, which has no preferred orientation, all that need concern us in this experiment is the magnitude of the vector.) The most important incoherent scatterer is hydrogen for which $\sigma_{inc}=80.3$ barns/atom whereas σ_{coh} is only 1.76 barns/atom (1 barn = $10^{-24} cm^2$). Since the incoherent scattering cross section of hydrogen is much larger than those of almost all other nuclei, it is often reasonable (as a first approximation) to neglect the coherent scattering in systems that contain a relatively large fraction of hydrogen atoms.

Can you explain the usefulness of deuteration, given that σ_{coh} (deuterium) and σ_{inc} (deuterium) are 5.6 barns/atom and 2 barns/atom, respectively?

Elastic neutron scattering is scattering with no change in neutron energy, i.e. with $\hbar\omega=0$, and inelastic neutron scattering is scattering with a change in neutron energy, i.e. with $\hbar\omega=0$. On the other hand, quasielastic neutron scattering (QENS) involves the Doppler-like broadening of otherwise elastically scattered neutrons due to reorientational or diffusive motions of atoms in the target material. Thus QENS is a special kind of inelastic neutron scattering. In this experiment you will use neutron scattering to perform low energy inelastic measurements on methane tunneling modes and measure the resulting QENS spectra at higher temperatures.

We shall first describe the sample to be used for the experiment, and the equipment that will be used to bring samples to the desired measurement conditions (methane concentration/temperature). The next section gives a brief discussion to the spectrometer as well as matters to be considered in choosing the incident wavelength for this experiment. We then describe the reduction of the data to obtain the scattering function, and we follow with some words about the scattering that is expected for these measurements. This then sets the scene for the analysis and discussion of the experimentally measured scattering function.

II. The sample

Prior to the experiment ~2.26 grams of the deuterated MOF5 sample will have been loaded into a vanadium cell with a valve and gas-line attached. Before the start of the experiment we will have mounted the sample in a helium cryostat, and taken select background measurements. We will then load methane into the sample using a known volume (37.5 cc). Through the night we shall collect data at temperatures and wavelengths decided by the experimental group.

Given that the molecular mass of MOF5, $Zn_4O(C_8O_4D_4)_3$, is 781.6 g/mole, what pressure of methane must be loaded to achieve a loading level of 1 CH₄ per ZnO₄? (The gas is at room temperature and in a volume of 37.5 cc.)

Why do we typically use aluminum for sample containers and cryostat windows?

Apart from indium, what materials might be used to seal sample containers?

To reduce the data we will need a detector normalization file obtained using a sample of vanadium metal, plus a run with the beam closed (a type of background). These runs will have been performed before the start of the summer school since there will not be time to complete them during the school.

Why do we use vanadium to normalize the data from different detectors? Hint: $\sigma_{coh} = 0.02 \text{ barns/atom}$, $\sigma_{inc} = 5.19 \text{ barns/atom}$.

III. The spectrometer

We shall be performing this experiment using the Disk Chopper Spectrometer (DCS), which is a so-called "direct geometry" (fixed incident energy) time-of-flight spectrometer. In this type of instrument (figure 3) bursts of monochromatic neutrons strike the sample at equally spaced times. The energies of the scattered neutrons are determined from their arrival times at the detectors, since we know when the pulses were created as well as the distances D_{PS} from the pulsing device to the sample and D_{SD} from the sample to the detectors. There are two ways to produce a monochromatic pulsed beam at a steady state neutron source. One method is to use a single crystal to monochromate the white beam and a mechanical "chopper" to pulse it; the other method is to use multiple choppers, such as the seven (!) choppers of the DCS.

A monochromatic pulsed beam of neutrons can in principle be created using two choppers. How does that work? Can you think why more than two choppers might be needed and/or desirable?

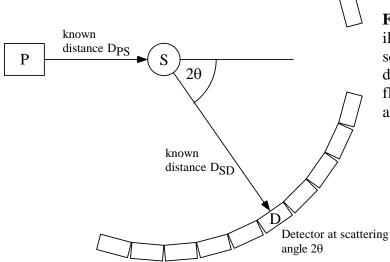


Figure 3. A schematic illustration of the scattering geometry for a direct geometry time-of-flight spectrometer such as the DCS.

Given the initial and final energies of the neutrons, E_i and E_f , the energy transfer $\hbar\omega = E_i - E_f$ is trivially obtained. Knowing the scattering angle 20 we can also calculate the magnitude of the momentum transfer to the sample, $\hbar Q$:

$$(\hbar Q)^2 = 2m_n \left[E_i + E_f - 2\sqrt{E_i E_f} \cos 2\theta \right]$$
 (2)

where m_n is the mass of the neutron. (This follows from the definition $\overrightarrow{Q} = \overrightarrow{k_i} - \overrightarrow{k_f}$ and the relationship between the magnitude of a neutron's wave vector, k, and its energy E: $E = \hbar^2 k^2 / 2m_n$.)

The data acquisition system separately accumulates neutron counts for each of the 913 DCS detectors. Furthermore the time between pulses, T, is normally divided into 1000 time channels of equal width $\Delta t = 0.001T$ and each neutron event in a given detector is stored in one of these time channels according to its time of arrival at the detector. Thus the data acquisition system generates a two-dimensional array of counts I(i,j) as a function of detector index i and time channel index j. This array is accumulated in a "histogramming memory" which is resident in the data acquisition computer and reflected to the instrument computer. At the end of each run cycle the array is saved, along with other pertinent information, to the hard disk of the instrument computer.

With the sample environment mounted on the spectrometer, we can control and monitor the temperature remotely, but care must be taken not to exceed 110 K so that methane does not desorb from the sample. A decision about which incident wavelength (together with other instrument parameters such as the "master speed" of the choppers) you will use must be made. The choice of wavelength is critical to the experiment and several factors must be considered. These include intensity at the sample (which peaks, remaining roughly constant, between ~2.5 and ~4.5 Å, see Appendix A), the width of the elastic energy resolution function (which roughly varies as $1/\lambda^3$), the available Q range

(which varies as $1/\lambda$), and concerns about "frame overlap" problems. A related consideration is the available range in sample energy gain (neutron energy loss).

What is the maximum theoretical sample energy gain that can be measured when the incident energy is E_i, and how long would it take to measure the intensity of neutrons scattered with this change in energy?

Over night you need to collect data. You will need to define a "sequence" consisting of several "runs" plus at least one change of sample temperature and probably wavelength. Each run is divided into a set of "cycles" that are of defined lengths of time. At the end of each cycle the temperature is recorded and the data are backed up to the disk. Having defined the runs we shall start the overnight sequence of measurements. Next day we shall stop the measurements and start into the data reduction.

In the experimental runs we shall collect intensity histograms I(i,j) for the sample at a temperatures and wavelengths of YOUR choice! Using previously acquired intensity histograms for a vanadium sample and for a "dark count" run with the beam shutter closed, we shall reduce the data to obtain the scattering function, $S(Q,\omega)$, with various cuts of the data in both Q and ω .

IV. Data reduction

In this section we shall simply indicate some of the more important steps in the data reduction process. We shall go into greater detail in our discussions at the time that the data reduction takes place.

The measured scattering from the sample has components from the intrinsic scattering from the sample and background intensity. Before doing any data analysis we need to subtract a time independent background from each of the runs.

Where does the time independent background come from?

Neglecting effects such as self-shielding and multiple scattering the scattering in detector i and time channel j may be related to the corresponding double differential cross section $[d^2\sigma/d\Omega dt]_{ij} \mbox{ (note that this is per unit time, not energy) in the following fashion:}$

$$I(i,j) = \frac{I_{\text{BM}}}{\eta_{\text{BM}}} \cdot \left[\frac{d^2 \sigma}{d\Omega dt} \right]_{ij} \Delta\Omega \Delta t \cdot N_{\text{m}} \cdot \eta_{ij}$$
 (3)

where $\Delta\Omega$, the solid angle subtended by detector i, and Δt , the width of time channel j, are (for these measurements) independent of i and j respectively, N_m is the number of

sample molecules in the beam, η_{ij} is the efficiency of detector i for neutrons detected in time channel j, and I_{BM} and η_{BM} are respectively the counts and the efficiency of the beam monitor (situated upstream of the sample).

Since we are not trying to extract an absolute cross section we can neglect the multiplicative constants in the above equation, but we should not ignore the detector efficiency function η_{ij} . Since all of the detectors are to first order identical it is not unreasonable to treat η_{ij} as the product of two terms, a function η_{i0} which represents the efficiency of detector i for elastically scattered neutrons and a detector-independent function f_j that describes the energy dependence of the efficiency of the detectors. The correction for differences in detector response, i.e. the determination of η_{i0} , is performed using the results of a measurement with a vanadium sample.

The correction of the data for the energy dependence of the efficiency is achieved by calculation, knowing the various factors that affect the probability that a neutron is absorbed within a detector.

What are these factors?

To improve statistics, in some instances, we may define several detector groups, each of which includes detectors within a specified range of angles. The differential cross section $[d^2\sigma/d\Omega dt]$ for all detectors in a group will be summed and divided by the number of detectors in the group. Having obtained a quantity proportional to $[d^2\sigma/d\Omega dt]$ we must now compute $[d^2\sigma/d\Omega d\omega]$ and finally $S(Q,\omega)$. Since a neutron's energy E is related to its time-of-flight t over a fixed distance as $E\propto t^{-2}$, it follows that $dE\propto t^{-3}dt$. Hence

$$\frac{d^2\sigma}{d\Omega d\omega} \propto \frac{d^2\sigma}{d\Omega dE_f} = \left[\frac{d^2\sigma}{d\Omega dt}\right] \left(\frac{dt}{dE_f}\right) \propto \left[\frac{d^2\sigma}{d\Omega dt}\right] t^3$$
 (4)

To obtain $S(Q, \omega)$ we simply divide by k_f (see eq. 1). Equivalently we multiply by another factor of t.

If a system in thermodynamic equilibrium can exist in a number of thermodynamic states and we consider two such states separated by an energy difference $\hbar\omega$, the probability that the system is in the lower energy state is greater by a factor $\exp(\hbar\omega/kT)$ than the probability that it is in the higher energy state. From this it can be shown that for systems in thermodynamic equilibrium the scattering function $S(Q,\omega)$ satisfies the so-called "detailed balance" relationship: $S(-Q,-\omega)=\exp(-\hbar\omega/kT)S(Q,\omega)$. Since we shall be fitting the data to a theoretical form that is symmetric in $\hbar\omega$ we shall first "symmetrize" the experimental $S(Q,\omega)$ by multiplying it by $\exp(-\hbar\omega/2kT)$.

Is symmetrization of $S(Q,\omega)$ likely to be a larger effect at low or high temperatures?

Having reduced the experimental data to a symmetrized scattering function it is time to relate the results to theory.



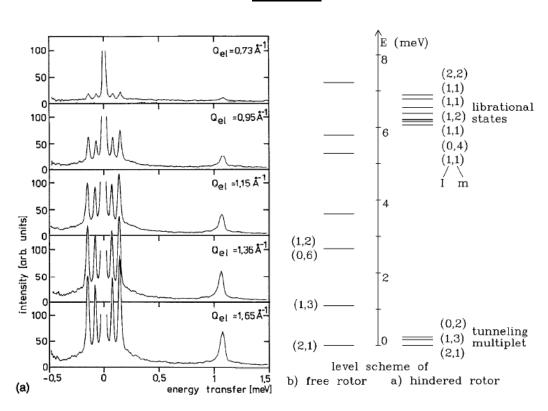


Figure 4.(a) Momentum transfer dependence of the neutron transitions of phase II methane. [Asmussen]. Data were taken on IRIS with the Graphite (002) reflection. (b) Rotational level scheme of methane in solid phase II (after Grieger). Parenthesis (I,mj) denotes the total nuclear spin I and the multiplicity m of the levels. The multiplicity of the rotational states is m(2I+1).

Consider pure methane. In the phase II crystal structure there are actually two different types of methane molecules below $T=20~\rm K$. These molecules show two different kinds of rotational excitations: 75% of the molecules are orientationally ordered lying in a deep minima of a strong potential, while the other 25% are orientationally disordered. The high resolution neutron scattering spectra consists of tunneling lines at 75 μeV and 145 μeV with a librational feature at 1.07 meV (figure 4(a)) [Asmussen] . Much research has been performed to measure and understand the dynamics of tetrahedral shaped molecules in various environments. Some of the work is embodied by such authors as Press and one of several theoretical descriptions given by the 'pocket' formalism of Huller. The rotational wave functions can be constructed considering the tetrahedral symmetry of the CH₄ molecule with the corresponding total spin of I=2 (ortho), I=1 (meta) and I=0 (para) having representations A (five times one-fold degenerate), T (three times triply degenerate) and E (one doubly degenerate), respectively. The rotational levels of the

hindered rotor molecules are shifted strongly in comparison to the free rotor molecules (figure 4(b)) and we can immediately see that the free rotor is responsible for the peak above 1 meV, while the low energy features arise from the tunneling multiplet (note that the $A \leftarrow \rightarrow E$ transitions are forbidden since they require a nuclear spin change of 2) that occurs due to wavefunction overlap between adjacent deep potential minima. In addition to the classically forbidden rotation of the molecule, there are also classical librational motions of the molecule within the potential minimum that gives rise features above 5 meV.

In general, there are 9 possible transitions between energy levels for a regular tetrahedron in a potential field that has no symmetry. These are shown in Figure 5. However in a symmetric field this number is much reduced and the energy level diagram and tunneling spectra are simplified. For a tetrahedral molecule in a tetrahedral (cubic) field, then the T levels are degenerate and hence only two transitions are observed as shown above for orientationally ordered methane.

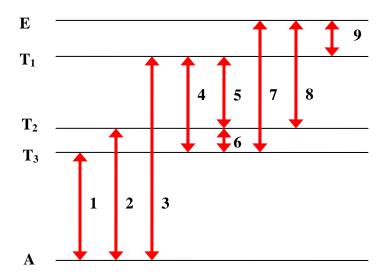


Figure 5.

A schematic illustration of the possible tunneling transitions for a general tetrahedral molecule.

In the current case we need to consider a tetrahedral molecule in a trigonal field, since that is the symmetry of the first methane adsorption site in MOF5. In this case Huller constructed an energy level scheme which is controlled by two matrix elements, h and h_4 (see Huller). These two variables can then be related to the barriers to rotation about both the three equivalent three-fold axes and the unique direction through the center of the cupsite.

The matrix elements are related to the barrier to rotation about the specified axes and the relative intensities may be arrived at through a lengthy set of integrals (see Huller). A separate sheet of instructions will be available during the course to calculating the barrier from the matrix elements. For the specific case of a tetrahedral molecule in a trigonal symmetric field, the transitions 1 through 5 have relative ratios 10: 5:2:8:4. Further, the spherical average of the Q-dependence has been shown to be related to a series of spherical Bessel functions [see Press].

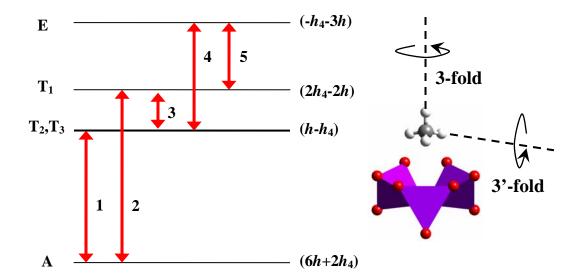


Figure 6. A schematic illustration of the tunneling transitions for a methane molecule in a trigonal field as expected for the first adsorption site in MOF5 (right). The energy levels for each manifold is given as a function of the matrix elements h and h_4 .

Upon heating the MOF5-CH₄ complex to temperatures up to 110 K, the populations of the methane rotational levels gradually redistribute according to Boltzmann statistics and ultimately there is enough energy in the system to overcome the classical barrier to rotation and a quasielastic feature will develop in the spectra.

Most quasielastic experiments are performed on systems that predominantly scatter neutrons incoherently. The incoherent scattering function is the space and time Fourier transform of the <u>self-correlation function</u> $G_s(\vec{r},t)$ which (classically) represents the probability that a particle that was at the origin at time t=0 is at position \vec{r} at time t.

Use a similar definition to relate the coherent scattering function to a (different) space-time correlation function. Why is it easier to formulate models of atomic motion using incoherent scattering?

A common way of expressing $S_{inc}(\vec{Q},\omega)$ is in terms of the <u>intermediate self scattering</u> $\underline{\mathit{function}}\ I_s(\vec{Q},t)$, which is the space Fourier transform of $G_s(\vec{r},t)$:

$$S_{inc}(\vec{Q},\omega) = \frac{1}{2\pi} \int I_s(\vec{Q},t) e^{-i\omega t} dt$$
 (5)

An illustrative model for $G_s(\vec{r},t)$ (though inappropriate in the context of the present experiment) is that of simple Brownian diffusion, where times of observation are much longer than typical times between collisions. Fick's Law governs this type of diffusion:

$$\frac{\partial}{\partial t} G_{s}(\vec{r}, t) = D\nabla^{2} G_{s}(\vec{r}, t), \qquad (6)$$

where D is the diffusion constant. A solution to this equation is given by a self-correlation function of the form

$$G_s(\vec{r},t) = \frac{\exp(-r^2/4Dt)}{(4\pi Dt)^{3/2}},$$
 (7)

the space Fourier transform of which is

$$I_{s}(\vec{Q},t) = \exp(-Q^{2}Dt). \tag{8}$$

Since this represents an exponential decay in time, the time Fourier transform yields a Lorentzian lineshape:

$$S_{inc}(\vec{Q},\omega) = \frac{1}{\pi} \left[\frac{DQ^2}{(DQ^2)^2 + \omega^2} \right]$$
 (9)

that is centered at zero energy transfer and has a full width at half maximum height (FWHM), Γ , given by

$$\Gamma = 2DQ^2. \tag{10}$$

In the current situation, we could expect that the methane undergoes numerous types of reorientation: three fold jumps about the 'unique' axis; rotational diffusion about 'unique' axis; isotropic jump reorientation between all four sites; preferred jump reorientation between pseudo-equivalent sites; or isotropic rotation on a sphere. These are not the only possibilities and complications may arise if the rotational motion couples with translational motions.

To simplify the analysis, we will assume an isotropic rotational diffusion model and work through to the expected characteristics of the scattering law. To first order, the rotation of methane can be considered as the diffusion of four rigidly coupled protons at radius R=1.093~Å from the center of mass. In a classical treatment the angular motion of each molecule satisfies the Fick's Law and a powder average can then be expressed as a delta function and a sum of Lorentzians:

$$S_{incc}^{rot}(\vec{Q}, \omega) = A_0 \delta(\omega) + \sum_{\ell=1}^{\infty} (2\ell + 1) j_{\ell}^2(\vec{Q}R) \frac{1}{\pi} \left[\frac{\tau_l}{1 + \omega^2 \tau_l^2} \right]$$
(11)

where, $A_0 = J_0^2(QR)$ and is often called the Elastic Incoherent Structure Factor (EISF).

The correlation time (or equivalently, twice the reciprocal of the FWHM) is related to the diffusion constant D_R through

$$\tau_{\ell} = \frac{2}{\Gamma} = \frac{1}{\ell(\ell+1)D_R} \tag{12}$$

and j_{ℓ} is a spherical Bessel function. Note that the HWHM is Q independent.

references:

Asmussen B., J. Chem. Phys (1992) 97, 1332.

Press W., *Single-Particle Rotation in Molecular Crystals*, Springer Tracts in Modern Physics, Vol. 92, Springer, Berlin, 1981.

Huller, Phys. Rev. B. (1977), 16. 1844. Grieger, J. Chem. Phys. (1998) 109, 3161. Smalley M.V., Mol. Phys., (1981), 44, 533

VI. Data analysis

We will take a few detours on the route to obtaining a symmetrized form of the experimental scattering function, $S(Q,\omega)$. We will sum data to obtain a view of the powder diffraction and we will also sum to obtain a generalized vibrational densities of states that we will then briefly compare to the expected librational features.

The next step will be to fit the $S(Q,\omega)$ data in the tunneling regime as a function of Q. We suggest that you try fitting each Q group to a series of phenomenological Gaussians and possibly a background function. Extract the fitted parameters.

Following the method of Smalley *et al.*, we can use the following figure of tunnel splitting (figure 7) to derive the rotational barriers (given that this plot was obtained with a mean parameter \tilde{A} =-3.8B). Note that the rotational constant, B, for methane is 0.656 meV. Then, using the relationships of $V = -\frac{8}{9}A_3 - \frac{4\sqrt{15}}{9}\tilde{A}$ and $V' = -\frac{8}{9}A_3 + \frac{4\sqrt{15}}{27}\tilde{A}$, extract the rotational barriers for the unique and other three fold axes, V and V', respectively.

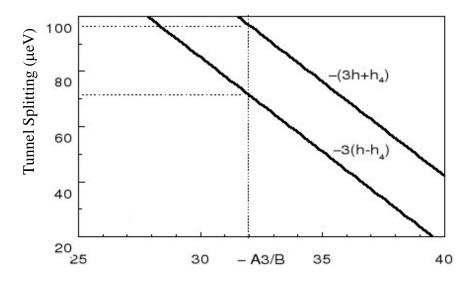


Figure 7. Tunnel splitting parameters as a function of the 'A₃' parameter.

- (1) How well do the Gaussians fit?
- (2) Assign the features to the rotational transitions for a tetrahedron in a trigonal field
- (3) Do the ratios of intensities agree with expectations?
- (4) Sum the rotational transitions at each Q. How does this compare with the expected form factor?
- (5) Calculate values for h and h_4 .
- (6) Estimate the barriers to rotation about the hard and soft axes.
- (7) Methane on graphite has values of 25 meV and 21 meV, respectively. Comment.

If you have taken data at higher temperatures, use the isotropic rotational diffusion model described in the previous section (i.e. an elastic delta function and just *one* broader Lorentzian) to fit the 'empty sample' background subtracted data as a function of Q. In an actual experiment the scattering function is broadened with the instrumental resolution function so the model function must be numerically convoluted with the instrumental resolution function. Having fitted the experimental data to the model, the next step is to make plots of the Lorentzian line parameters as functions of Q.

- (8) How well does the single Lorentzian fit?
- (9) How do the Lorentzian parameters behave?
- (10) Can you extract a rotational diffusion constant or correlation time?
- (11) If you have access to other temperatures, can you extract an activation energy?
- (12) By taking a ratio of elastic to total scattering and hence extracting A_0 , can you fit the expected functional form of the Bessel function?
- (13) Does the methane perform isotropic rotational diffusion? Comment.

VII. Concluding remarks

In section V we discussed a scattering function that corresponds to very simple model of rotational diffusive motion. The situation is more complicated when a system displays more than one type of diffusive motion, or rotational axes. If the various motions are uncoupled, the intermediate scattering function is a product of the individual intermediate scattering functions so that the scattering function is a convolution of the scattering functions for the individual motions. The situation simplifies considerably if additional motions occur on very different time scales. Motions that are much slower than the time scale represented by the instrumental resolution show up as elastic scattering. On the other hand motions that are much faster give rise to an essentially flat background. Different instruments, with different dynamical windows and different resolution capabilities, are needed to observe such motions. For example motions that are too slow to see using the DCS may well show up if the sample is put on the backscattering

spectrometer. Conversely motions that are fast by DCS standards can usefully be studied using the FANS spectrometer.

This experiment was inspired by work of Wei Zhou and Taner Yildirim and the sample was synthesized by Michael Hartman.

VIII. General references

- G.E. Bacon, "Neutron Diffraction", Clarendon Press, Oxford (1975).
- M. Bée, "Quasielastic Neutron Scattering", Adam Hilger, Bristol (1988)
- R. Hempelmann, "Quasielastic Neutron Scattering and Solid State Diffusion", Clarendon Press, Oxford (2000).
- S.W. Lovesey, "*Theory of Thermal Neutron Scattering from Condensed Matter*", Clarendon Press, Oxford (1987).
- G.L. Squires, "*Introduction to the Theory of Thermal Neutron Scattering*", Cambridge University Press (1978), republished by Dover (1996).

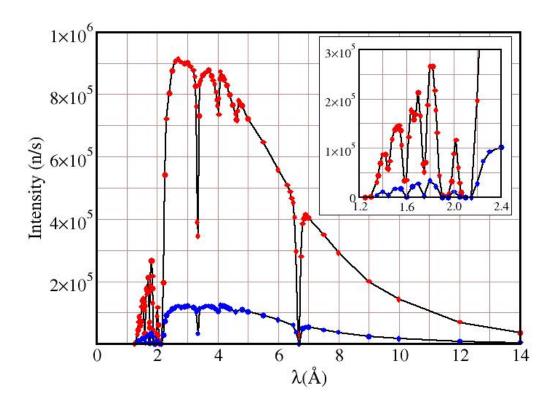
Appendix A. Instrument Characteristics for the Disk Chopper Spectrometer

(http://www.ncnr.nist.gov/instruments/dcs)

The white beam from the cold neutron source is cleaned of high energy neutron and gamma ray contamination using an "optical filter". This is basically a bent guide which ensures that there is no line of sight from the source to points beyond the local shutter. A cooled graphite filter removes short wavelength (~0.5 Å) neutrons that remain in the beam, permitting measurements at wavelengths down to roughly 1.5 Å.

A clean, pulsed, monochromatic neutron beam is produced using seven disk choppers. Chopper speeds may be varied from 1200 to 20000 rpm. The pulsing and monochromating choppers have three slots of different widths. In principle this permits three choices of intensity and resolution at a given wavelength and master chopper speed.

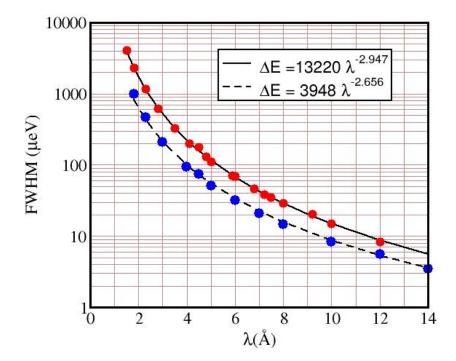
The measured intensity at the sample is reproduced below. Red and blue points (upper and lower plots) correspond to measurements using different chopper slot widths.



Why are there dips in the measured flux at wavelengths near 3.335 and 6.67 Å? What's going on around 2 Å?

The resolution of the instrument is approximately triangular and essentially independent of beam height (10 cm) but depends on the width of the beam. Hence samples should ideally be tall and thin rather than short and fat.

The measured elastic energy resolution, for the same choices of chopper slot width as in the intensity plot above, is shown in the figure below. Lines represent fits to the measurements.



An oscillating radial collimator, inside radius 200 mm, outside radius 300 mm, blade separation 2°, is used to reduce the scattering from sample environment structures.

Can you explain how the radial collimator works, and why it is oscillated?

There are 913 six atmosphere 3 He detectors covering an essentially continuous solid angle of \sim 0.65 steradians and arranged in three banks:

- Middle bank detector scattering angles range from -30° to -5° and from $+5^{\circ}$ to $+140^{\circ}$
- Upper and lower bank angles range from -30° to -10° and from $+10^{\circ}$ to $+140^{\circ}$

The flight distance from sample to detectors is 4010 mm. The flight chamber is purged with argon.

Why is the flight chamber purged with argon?

Appendix B. Possible Experiments on the Disk Chopper Spectrometer

Phenomena that can be investigated include:

- Translational and rotational diffusion processes, where scattering experiments provide information about time scales, length scales and geometrical constraints; the ability to access a wide range of wave vector transfers, with good energy resolution, is key to the success of such investigations
- Low energy vibrational and magnetic excitations and densities of states
- Tunneling phenomena
- Low Q powder diffraction

Research areas include:

- Chemistry --- e.g. clathrates, molecular crystals, fullerenes
- **Polymers** --- bound polymers, glass phenomenon, confinement effects
- Biological systems --- protein folding, protein preservation, water dynamics in membranes
- Physics --- adsorbate dynamics in mesoporous systems (zeolites and clays) and in confined geometries, metal-hydrogen systems, glasses, magnetic systems
- Materials --- negative thermal expansion materials, low conductivity materials, hydration of cement, carbon nanotubes, proton conductors, metal hydrides

Appendix C. Some useful properties and relationships

Neutron properties

Mass: $1.660 \times 10^{-24} \text{ g}$

Electric charge: 0Spin: $\frac{1}{2}$

Magnetic moment: -1.913 nuclear magnetons

Exact relationships

$$\lambda = \frac{h}{mv} \qquad \qquad E = \frac{1}{2} mv^2 \qquad \qquad k = \frac{2\pi}{\lambda}$$

Approximate relationships

$$E[\text{meV}] = \frac{81.8}{(\lambda |\mathring{A}|)^2}; \quad v[\text{mm/}\mu\text{s}] = \frac{3.956}{\lambda |\mathring{A}|}; \quad E[\text{meV}] = 2.07(k[\mathring{A}^1])^2; \quad 1 \text{ meV} = 8.1 \text{ cm}^{-1}$$

Appendix D. Spin Incoherence

The strength of the scattering of a neutron by a nucleus, i.e. the neutron scattering length, depends on the spin of the compound nucleus. For an isotope with nuclear spin I the combined "nucleus + neutron" spin, I', has two possible values, $I^+ = I + 1/2$ and $I^- = I - 1/2$, with which we associate two possible scattering lengths b^+ and b^- . Each of the possible values of the combined spin has 2I'+1 possible spin states, i.e. 2(I+1/2)+1=2I+2 and 2(I-1/2)+1=2I states respectively, for a total of 4I+2 spin states.

If the neutron and nuclear spins are randomly orientated, all states are equally probable, and the probabilities of the combined + and - spin states are $p^+ = (I+1)/(2I+1)$ and $p^-=I/(2I+1)$ respectively.

The mean scattering length, $\langle b \rangle$, and the mean of the scattering length squared, $\langle b^2 \rangle$,

$$< b> = p^+b^+ + p^-b^-$$
 and $< b^2> = p^+(b^+)^2 + p^-(b^-)^2$

are used to calculate the coherent and incoherent bound cross sections. These cross sections are defined as follows:

$$\sigma_{coh} = 4\pi \langle b \rangle^2$$
 and $\sigma_{inc} = 4\pi (\langle b^2 \rangle - \langle b \rangle^2)$.

Working through the numbers for hydrogen and deuterium is instructive. The relevant scattering lengths for hydrogen are $b^+ = 1.086 \times 10^{-12}$ cm and $b^- = -4.751 \times 10^{-12}$ cm, whereas the values for deuterium are $b^+ = 0.951 \times 10^{-12}$ cm and $b^- = 0.095 \times 10^{-12}$ cm.