

Quantum Rotations in Methyl Iodide

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Summer School on Methods and Applications of
Neutron Spectroscopy

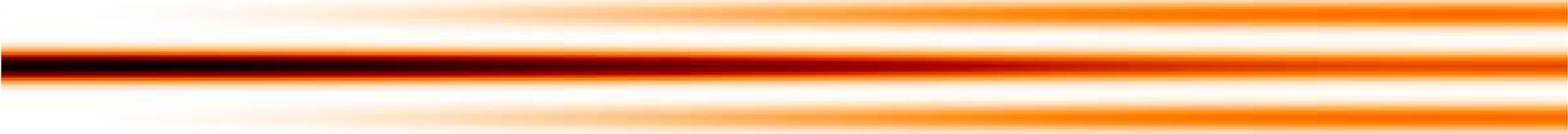
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NIST Center for Neutron Research
Gaithersburg, MD 20899



Quantum Phenomena

...weird things happen at small length scales...



- Wave/particle behavior of matter: $\Psi(x,t)$
- Quantized/discrete energy levels for confined particles
- Observable motion that is classically forbidden
- Fundamental limit to the precision with which we can simultaneously measure certain properties of matter

Classically *Forbidden* Phenomena



Prob $\sim 1/10^{(10^{39})}$

Stars in universe: 10^{21}

Size of universe (m): 10^{27}

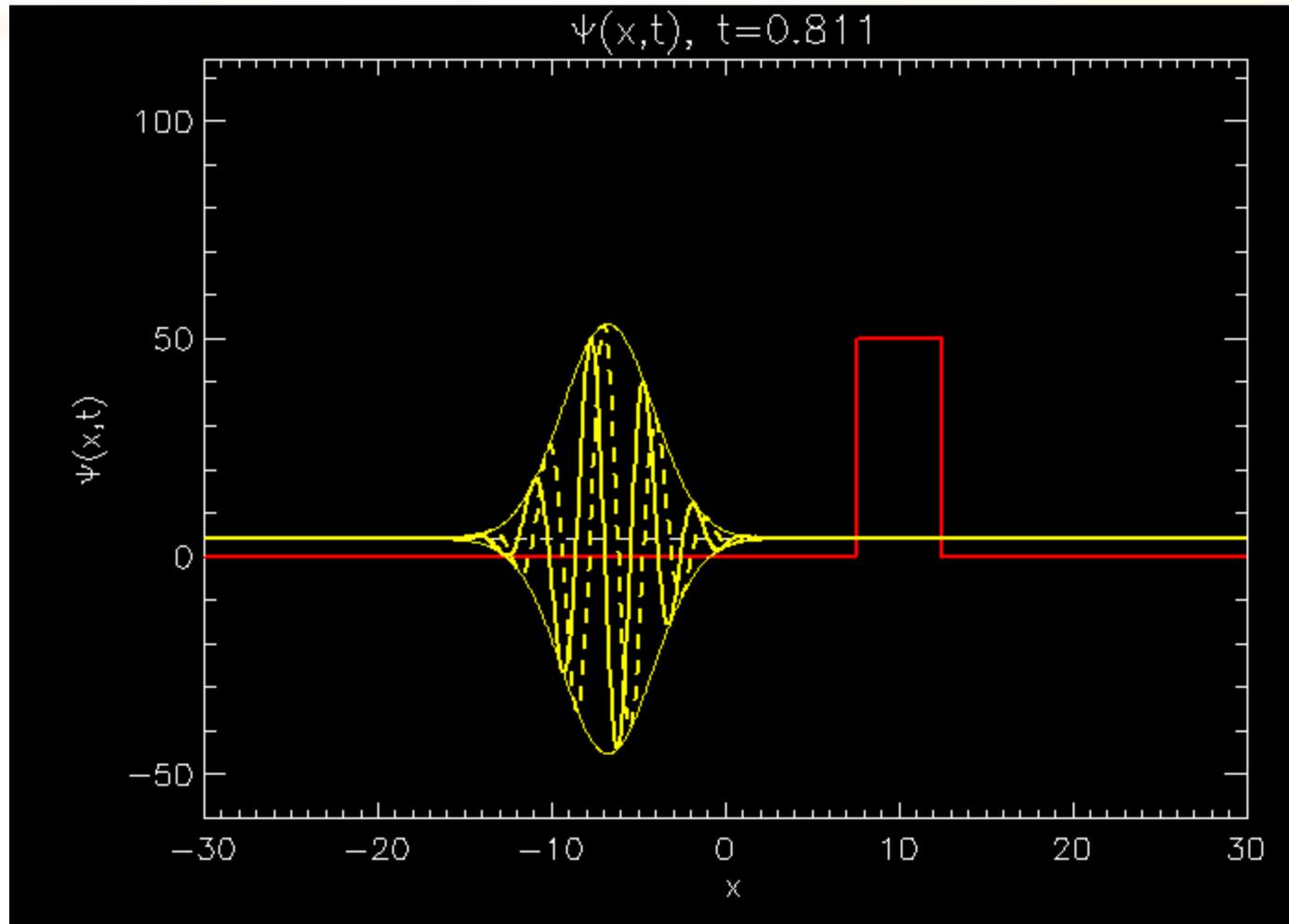
Water molecules in ocean:
 5×10^{46}

Hydrogen atoms in universe:
 10^{79}

Probability of a monkey
typing Hamlet with
random keystrokes:
 $1/10^{(10^5)}$

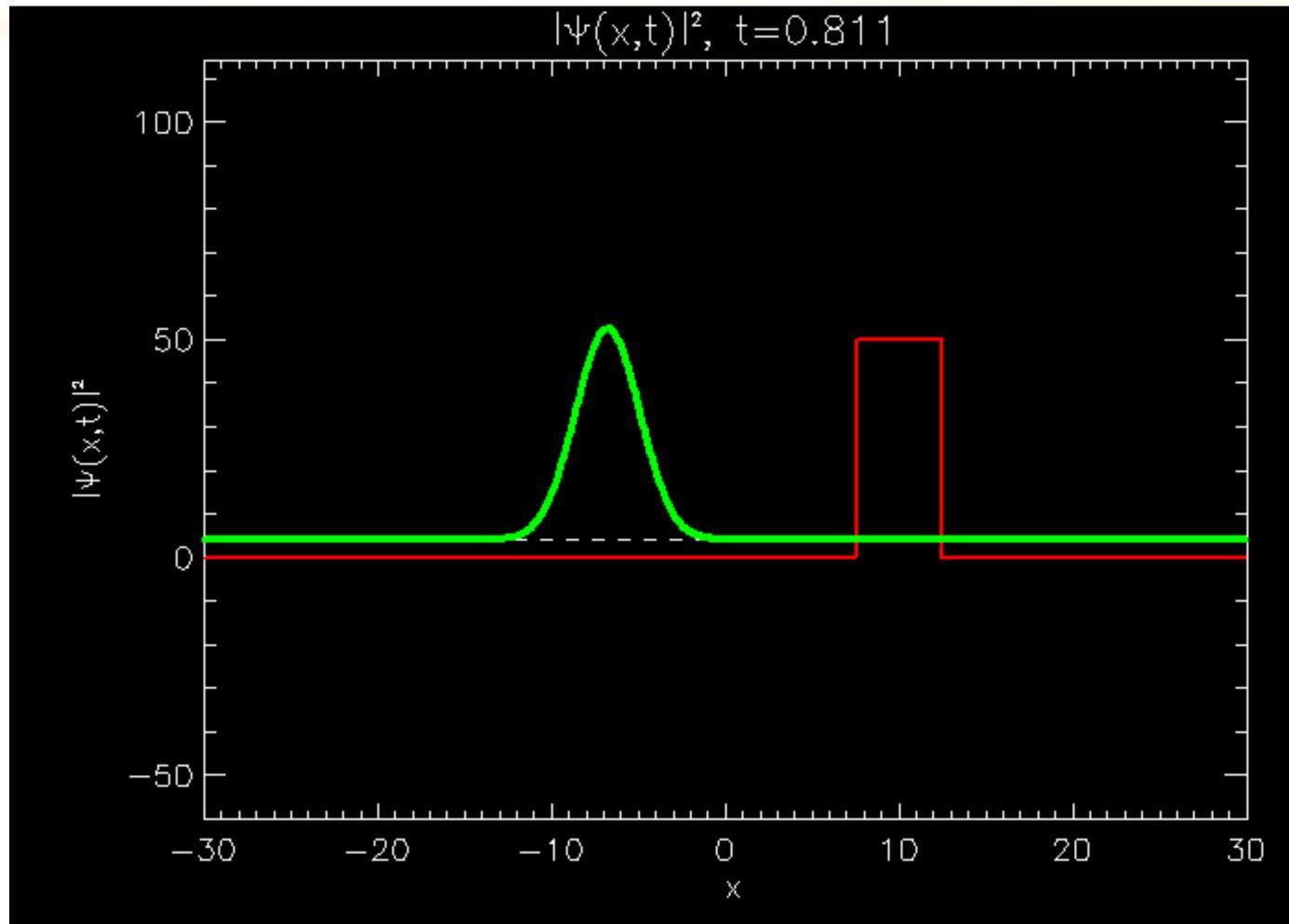
Quantum Phenomena

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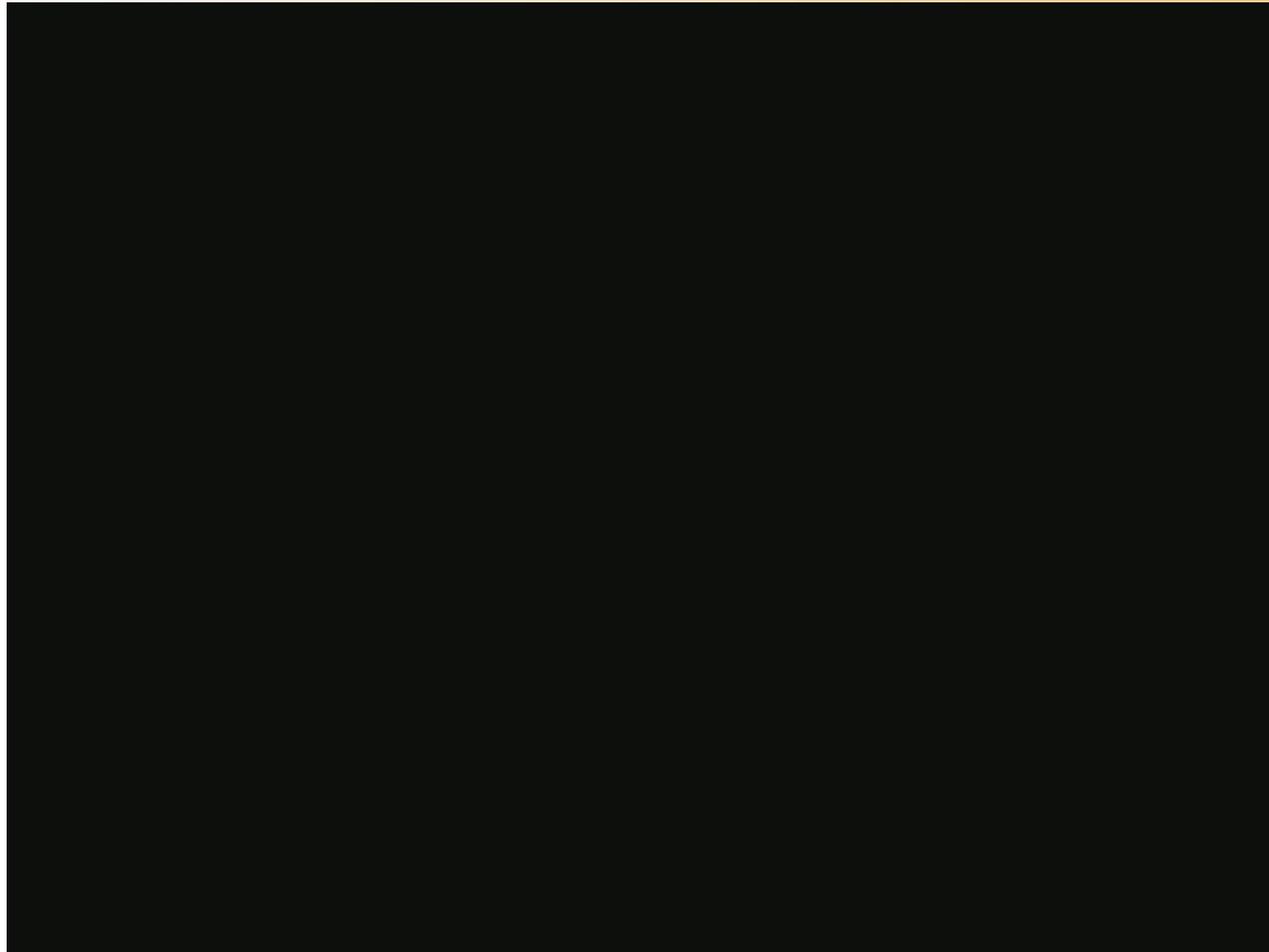
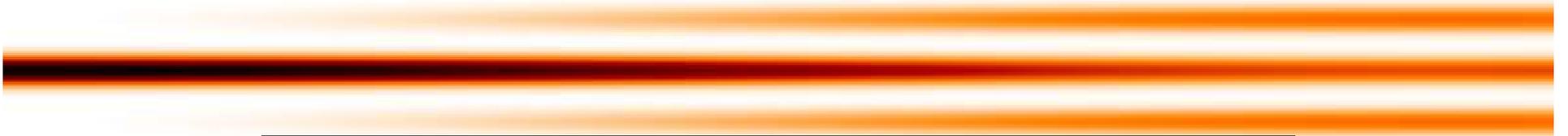
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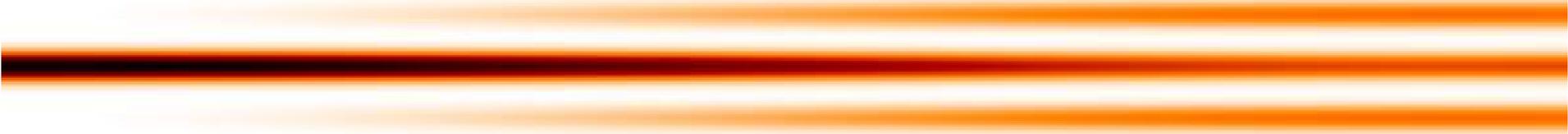


Quantum Phenomena

...weird things happen at small length scales...



What are quantum rotations?



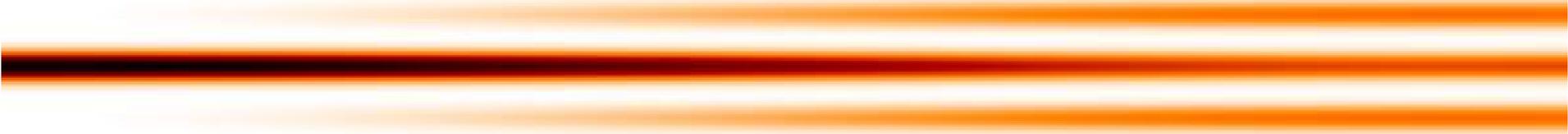
- Molecules in molecular solids can undergo reorientational motion
- H₂ is a dumbbell rotor and its quantum rotations are nearly "free" (i.e. no barrier hinders its motion)

$$E_l = Bl(l+1), \quad l = 0, 1, 2, \dots$$

$$B = \frac{\hbar^2}{2I}$$

- Hindered rotors can perform torsional oscillations and even rotational tunneling through the barrier!

Why study quantum rotations?



- Rotational dynamics as studied with neutrons reflect the molecular environment, i.e. the *energy landscape*
- Neutron tunneling spectroscopy provides extremely detailed information on the shape and magnitude of the potential energy of the molecular groups.
- Rotational tunneling measurements can be used to quantify interatomic interactions.
- Good test of first-principles/DFT calculations

Bulk CH_3I

A Canonical Rotational System

Properties

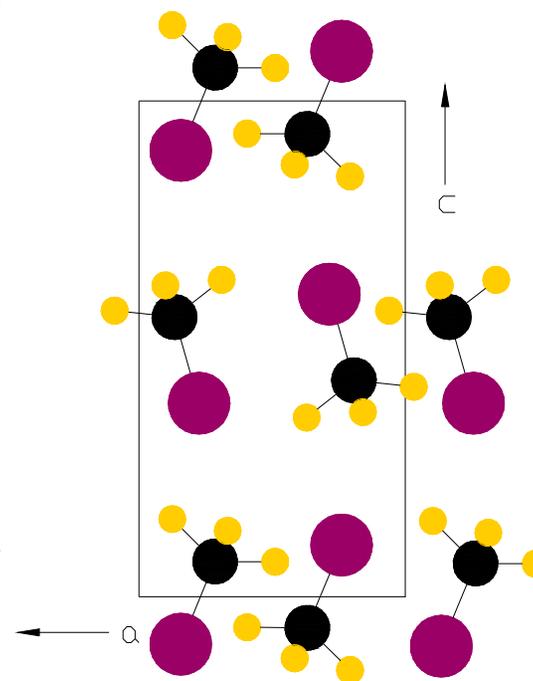
MP: -66.5°C

MW: 141.94 g/mol

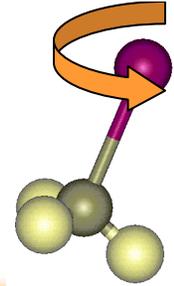
Dipole moment: $\mu = 1.62$ debye

Projection onto the a - c plane

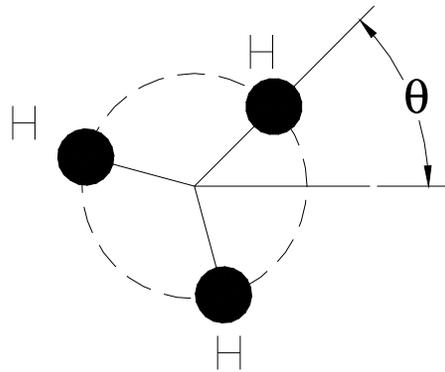
(Prager et.al., *J.Chem.Phys.* **86**, 2563 (1987))



The Methyl Group: CH₃



- We want to study the dynamics about the main molecular axis



$$I[\text{CH}_3] = 5.3 \times 10^{-47} \text{ kg} \cdot \text{m}^2$$

$$B = \frac{\hbar^2}{2I} = 0.65 \text{ meV}$$

Free rotor energy levels: $E_j = B j^2$, $j = 0, 1, 2, \dots$

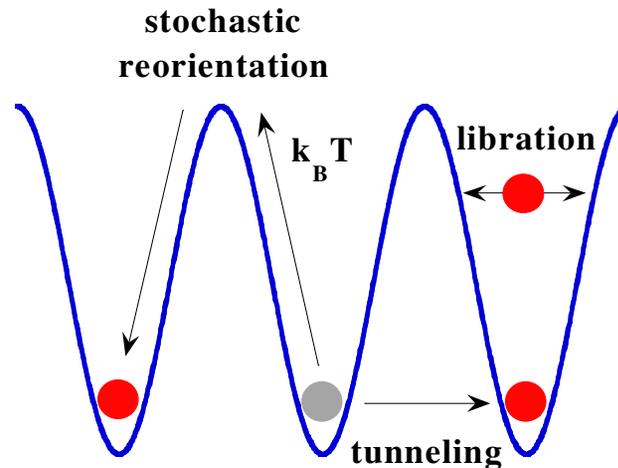
Useful conversions

$$1 \text{ meV} \leftrightarrow 4 \text{ ps}$$

$$1 \text{ } \mu\text{eV} \leftrightarrow 4 \text{ ns}$$

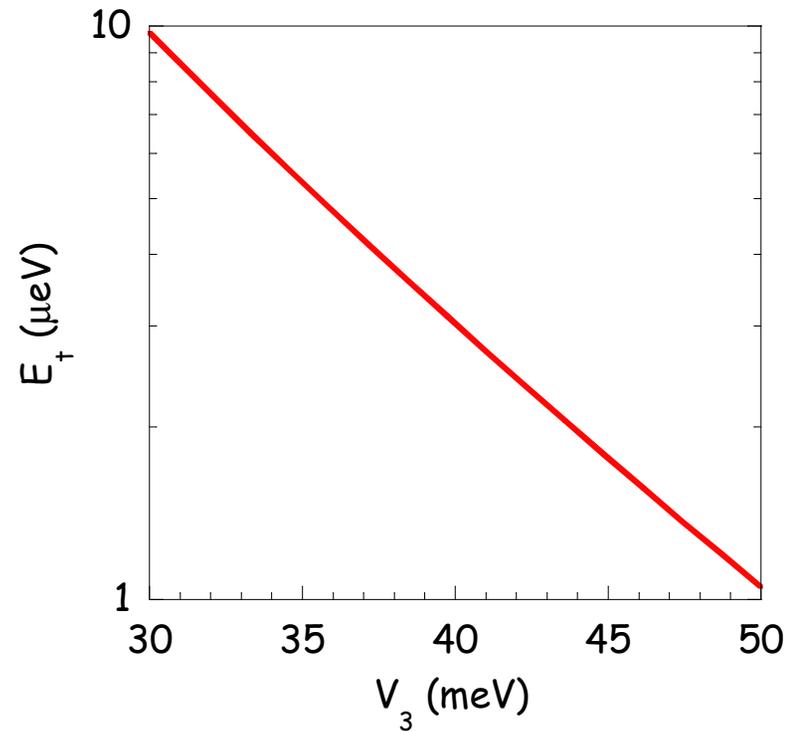
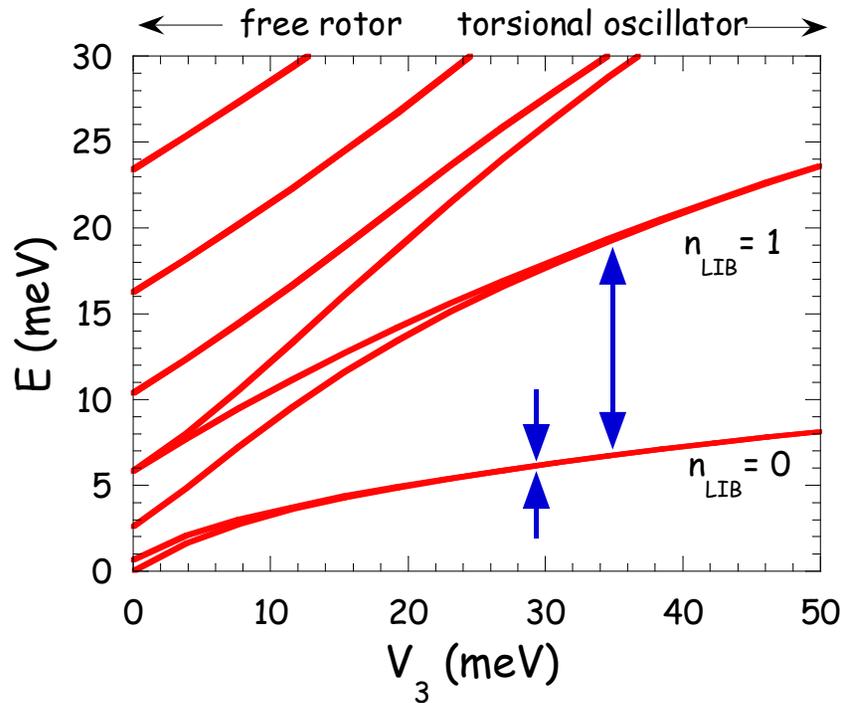
Bulk CH₃I Dynamics

- Interaction potential of methyl group (1) van der Waals term, (2) short-range steric repulsion, and (3) additional multipole terms
- Simplified model based on symmetry alone: $V(\theta) = \frac{V_3}{2}(1 - \cos 3\theta)$



Bulk CH₃I Dynamics

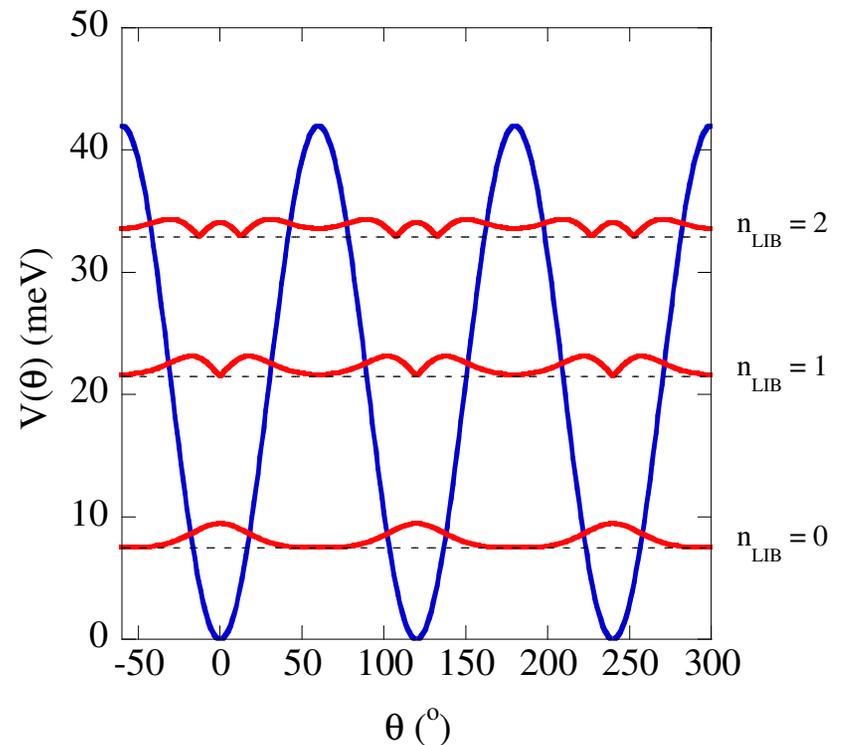
$$H = -B \frac{d^2}{d\theta^2} + \frac{V_3}{2} (1 - \cos 3\theta)$$



Tunneling energy very sensitive to the barrier height!

Rotational Tunneling

- Tunneling rate (...and energy) proportional to the overlap of the wavefunctions through the barrier
- Overlap increases with librational level (n_{LIB}) hence tunneling rate increases with librational level

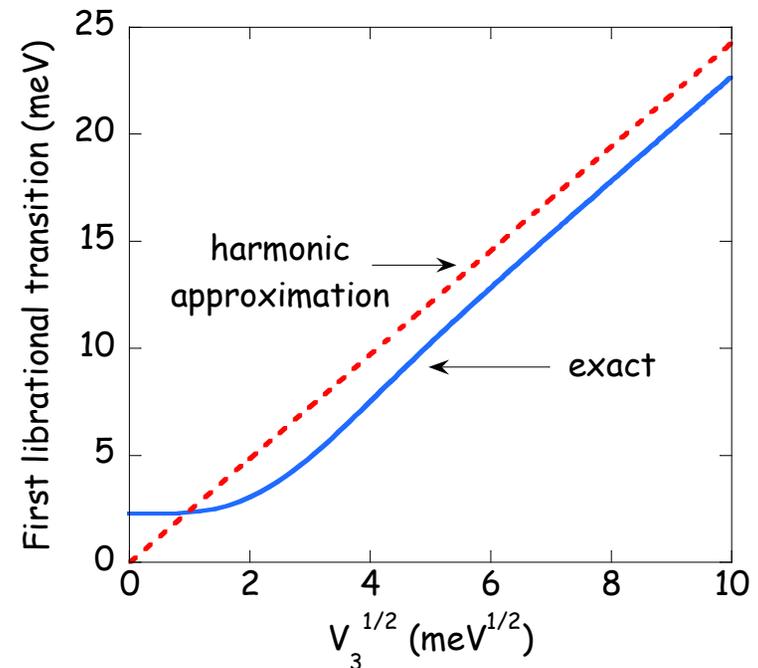
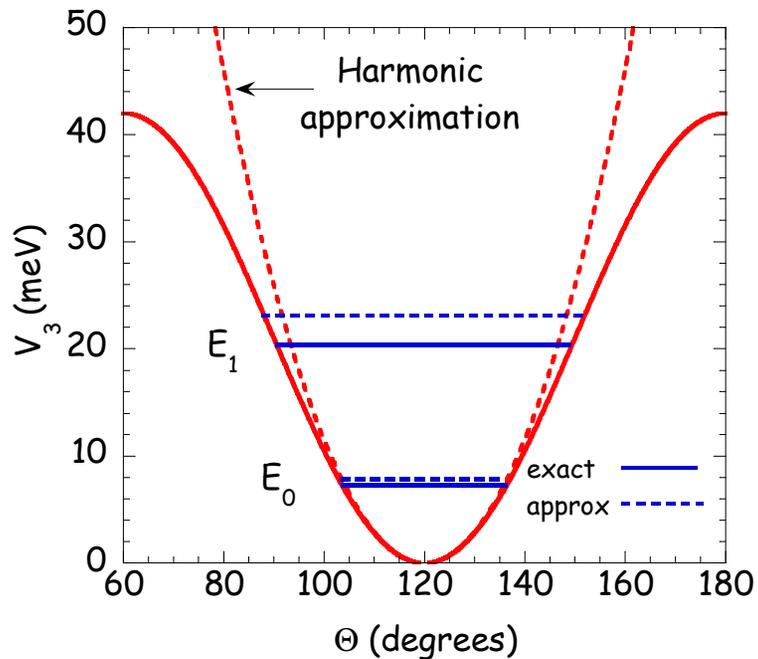


Librational Motion

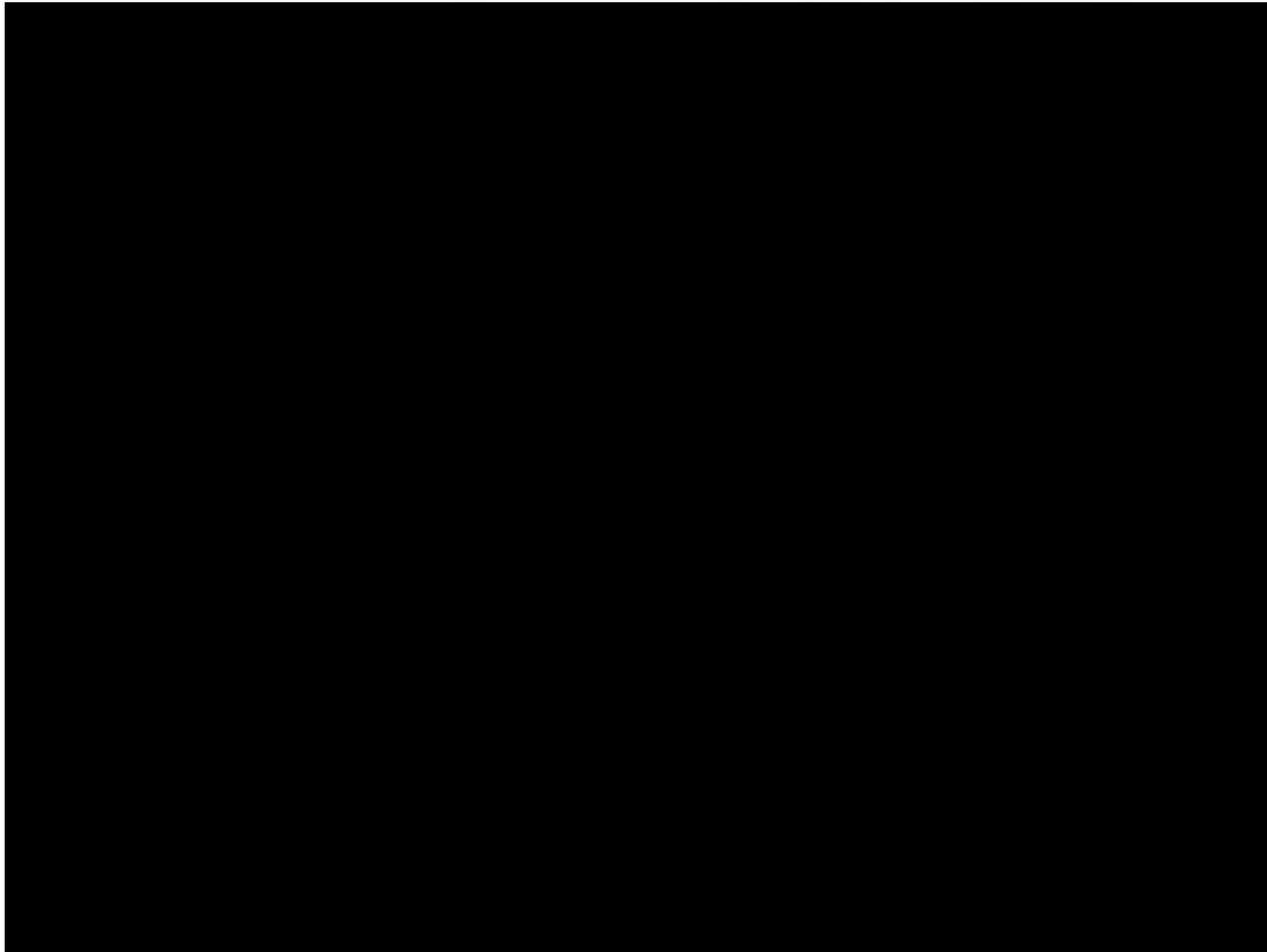
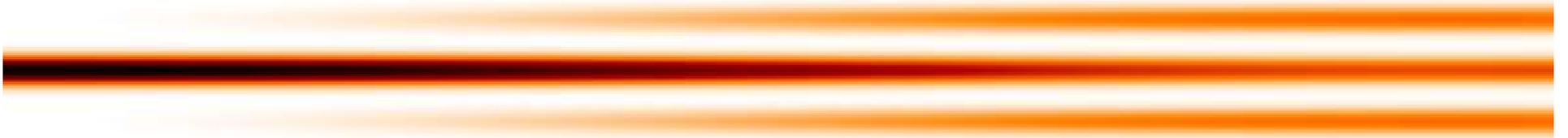
- Librations are torsional oscillations
- Harmonic approximation:

$$E_n = \left(n_{\text{LIB}} + \frac{1}{2} \right) \hbar \omega_0$$

$$\hbar \omega_0 = 3\sqrt{BV_3}$$



Quantum Rotational Dynamics



Using Inelastic Neutron Scattering to See Quantum Rotations

- Neutrons can induce a spin flip in hydrogenous species
- Incoherent scattering
- Simple case: H_2

$\Psi = \Psi_{\text{rot}} \Psi_{\text{ns}} \Psi_{\text{el}} \Psi_{\text{vib}}$ ($\Psi_{\text{el}} \Psi_{\text{vib}}$ are in the totally symmetric ground state)

Ψ must be AS upon nuclear exchange (composed of 2 fermions)

Ψ_{ns} must be AS(S) if Ψ_{rot} is S(AS)

$$\Psi_{\text{ns}}^{\text{S}} = \begin{cases} |\uparrow\uparrow\rangle \\ |\downarrow\downarrow\rangle \\ \frac{1}{\sqrt{2}} (|\uparrow\downarrow\rangle + |\downarrow\uparrow\rangle) \end{cases} \quad J = 1, \text{ ortho}$$

$$\Psi_{\text{ns}}^{\text{AS}} = \frac{1}{\sqrt{2}} (|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle) \quad J = 0, \text{ para}$$

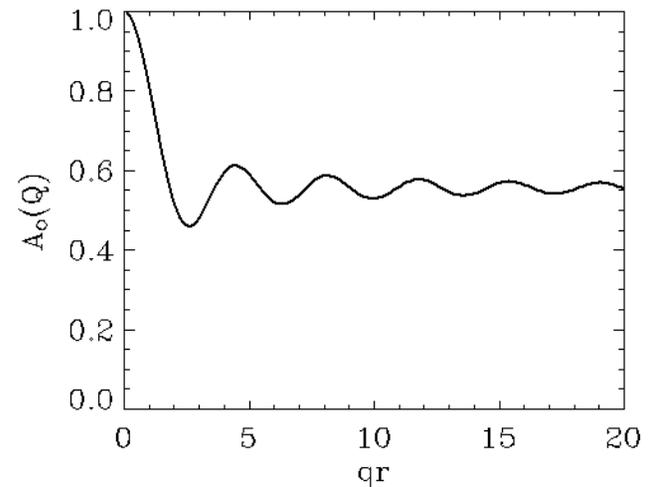
Using Inelastic Neutron Scattering to See Quantum Rotational Tunneling

Neutron scattering law for methyl tunneling

$$S(Q, \omega) = A_0(Q)\delta(\omega) + (1 - A_0(Q))\frac{1}{2}[\delta(\omega - \omega_t) + \delta(\omega + \omega_t)]$$

$$A_0(Q) = \frac{5 + 4j_0(QR\sqrt{3})}{9}$$

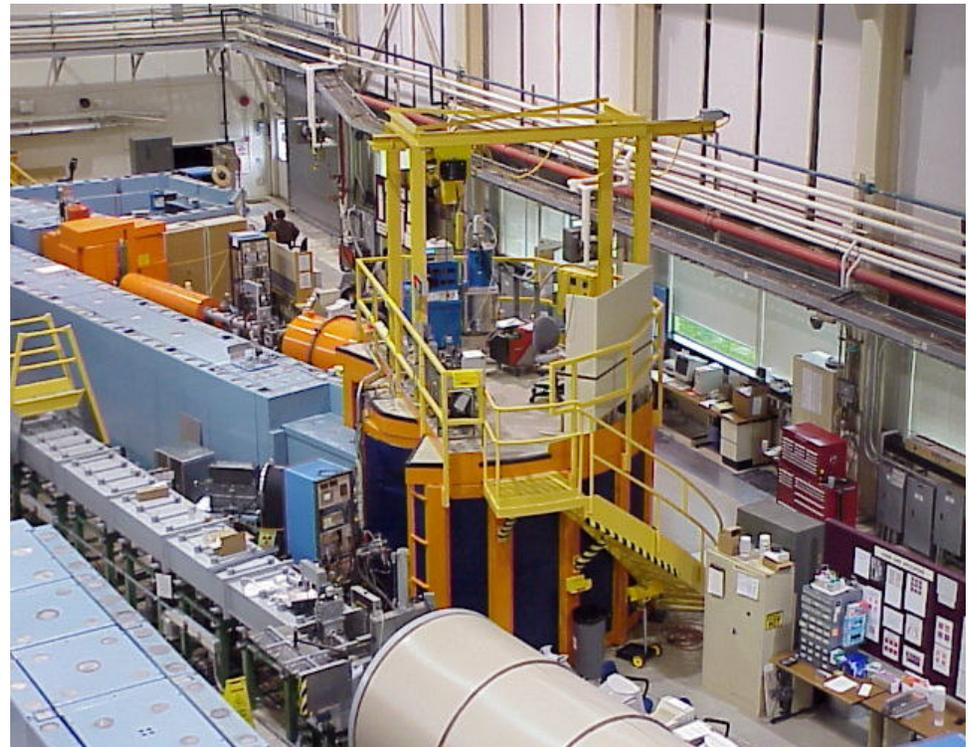
R: radius of methyl group
 ω_t : tunneling energy
 A_0 : elastic incoherent structure factor



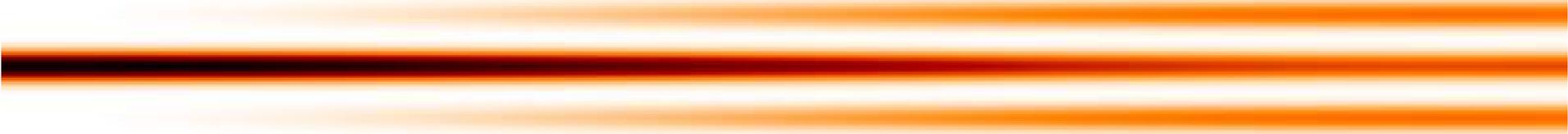
High Flux Backscattering Spectrometer

NIST Center for Neutron Research

- High energy resolution is often necessary to observe rotational tunneling directly.
- Typical neutron techniques to study tunneling include TOF, backscattering, and neutron spin-echo
- No other neutron spectrometer in North America is capable of measuring the tunnel splitting of CH_3I !



Are the HFBS measurements enough?



- Measuring the tunneling energy allows you to estimate the barrier height V_3

Can we stop here and declare victory?....NO!

- With knowledge of the barrier height you can estimate the librational transition energy E_0
- Confirmation that this model is correct requires that we perform an independent measurement like measuring the librational transition and comparing the measurement with our estimate

Filter Analyzer Neutron Spectrometer

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- $S(Q,E)$ reflects density of vibrational (librational) modes, $G(E)$
- Vary initial energy, fix final energy
- Measures energy transfers of order 10's-100's meV

Filter Analyzer Spectrometer

